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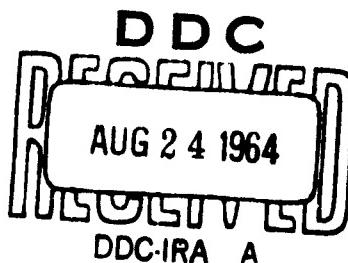
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Report of BAMIRAC

PROPERTIES OF COPPER PLASMA

Calculation of the Partition Function,
Species Concentration, and
Spectral-Line Intensities

STUART W. BOWEN



INFRARED LABORATORY
Institute of Science and Technology
THE UNIVERSITY OF MICHIGAN

July 1964

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Aircraft Propulsion Laboratory
Department of Aeronautical and
Astronautical Engineering

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NOTICES

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PREFACE

BAMIRAC, the Ballistic Missile Radiation Analysis Center, is a facility of the Institute of Science and Technology at The University of Michigan. It is supported by ARPA, the Advanced Research Projects Agency, under Contract SD-91, as a part of Project DEFENDER (research on and defense against ballistic missiles).

BAMIRAC functions as a center for analysis of and information on ballistic missiles and space vehicles, with the objective of formulating defense measures against them. Analyses of experimental and theoretical results, conducted in three general subdivisions corresponding to the launch, midcourse, and reentry phase of ballistic missile flight, are evaluated and examined for correlations. In its operation as a technical information center, BAMIRAC collects, processes, and disseminates information concerning electromagnetic radiation emanating from or caused by ICBM's or IRBM's, from reports of field measurements as well as from laboratory and theoretical results.

In addition to analyzing the results from external sources, BAMIRAC conducts its own theoretical and experimental investigations.

The results of the work within BAMIRAC are combined with the material obtained from outside sources to develop models for ballistic missile radiation. The information so obtained is disseminated in technical reports, which are published frequently. The technical content of the library holdings is made available to organizations whose representatives have proper authorization; in addition, the library prepares bibliographies on request. BAMIRAC also aids ARPA by handling various technical meetings and by publishing the Proceedings of the Anti-Missile Research Advisory Council.

BAMIRAC is under the technical direction of the Infrared Laboratory. It draws also, however, upon the capabilities of the Computation Department of the Institute, and upon those of the Aircraft Propulsion Laboratory of the Department of Aeronautical and Astronautical Engineering and of other departments of The University of Michigan, particularly within the College of Engineering.

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SYMBOLS, CONSTANTS, AND CONVERSION FACTORS

A_{n}^m	= Einstein transition probability from state m to n, sec ⁻¹
a	= Classical semi-major axis of electron energy state, cm
a_0, a_+	Defined by Equations 56 and 58
C	Defined by Equation 42
c	= Velocity of light, 2.997929×10^{10} cm/sec
d	= Debye-Hückel (subscript)
E_m	= Term value of state m, cm ⁻¹ = Energy level of state m, ergs
e	= Electron charge, 4.80286×10^{-10} esu
g_m	= Degeneracy of state m, $2J_m + 1$
h	= Planck's constant, 6.62517×10^{-27} erg sec
I	= Intensity of spectral line, erg sec ⁻¹ cm ⁻³ sterad ⁻¹
J_m	= Inner quantum number of state m
k	= Boltzmann's constant, 1.38044×10^{-16} erg deg ⁻¹ $= 1/1.4388$ cm ⁻¹ deg ⁻¹
K_p	= Equilibrium constant, atm
m	= Energy state
m_e	= Electron mass, 9.1083×10^{-28} g
N	= Original mole fraction of copper
n	= Total particle density
n_m	= Number density of atoms in state m, cm ⁻³
n_e	= Electron number density, cm ⁻³
n_q	= Number density of atoms in ionization state q. q = 0, 1, 2, . . .
p	= Pressure, atm
R	= Intensity ratio
T	= Temperature, °K
V_q	= Energy to ionize an atom from q-th to (q + 1)-st ionization state in the plasma
x_s	= Mole fraction of species s
Z_q	= Partition function for atoms in the q-th ionization state
σ	$= (2\pi m_e k)^{3/2}/h^3$ (Section 6)
β	$= Z_+/Z_0$
ν	= Frequency, sec ⁻¹
λ_d	= Debye radius, cm
λ	= Wavelength

$$\pi = 3.14159 \dots$$

ζ = Effective nuclear charge in units of e

Cu I = Neutral copper

Cu II = Singly ionized copper

Conversion Factors [1]

$$1 \text{ ev} = 1.60206 \times 10^{-12} \text{ erg}$$

$$= 8066.03 \text{ cm}^{-1}$$

$$= 2.41814 \times 10^{14} \text{ sec}^{-1}$$

$$1 \text{ cm}^{-1} = 1.98618 \times 10^{-16} \text{ erg}$$

$$= 12397.67 \times 10^{-8} \text{ ev}$$

$$1 \text{ sec}^{-1} = 6.62517 \times 10^{-27} \text{ erg}$$

$$= 4.13541 \times 10^{-15} \text{ ev}$$

$$1 \text{ erg} = 1.50940 \times 10^{26} \text{ sec}^{-1}$$

$$= 5.03479 \times 10^{15} \text{ cm}^{-1}$$

$$1 \text{ atm} = 1.013246 \times 10^6 \text{ dynes cm}^{-2}$$

Constants for Copper [7], [8]

For Cu I λ 5105.54

$$g_m A_n^m = .051 \times 10^8 \text{ sec}^{-1}$$

$$E_m = 30783.7 \text{ cm}^{-1}$$

For Cu I λ 5153.24

$$g_m A_n^m = 4.7 \times 10^8 \text{ sec}^{-1}$$

$$E_m = 49935.2 \text{ cm}^{-1}$$

Ionization potential for Cu I

$$= 7.72588 \text{ ev}$$

$$= 62317.2 \text{ cm}^{-1}$$

Ionization potential Cu II

$$= 20.2907 \text{ ev}$$

$$= 163665.6 \text{ cm}^{-1}$$

PROPERTIES OF COPPER PLASMA

Calculation of the Partition Function, Species Concentration, and Spectral-Line Intensities

ABSTRACT

The self-consistent partition function; ionization potential lowering; electron, ion, and neutral-particle species concentrations; and the absolute spectral-line intensities of Cu I λ 5106 and Cu I λ 5153 are reviewed and tabulated for pressures between 0.03 and 10 atm, for original mole fractions between 10^{-4} and 0.03 and for temperatures from 2000° to $10,000^{\circ}$ K, for copper vapor in a hot inert gas at thermal equilibrium. Only single ionization of copper is considered in this range of conditions. For the partition function cutoff, all electron states whose classical semi-major axis exceeds one-half the Debye distance are suppressed. The results should be of use in spectroscopic plasma diagnostics and other situations where the trace amounts of copper usually present can be used for temperature measurement.

1 INTRODUCTION

For the determination of temperature and other physical properties within a plasma, spectroscopic methods are of great interest and value [2, 3, 4].

In this report the formulas necessary to calculate species concentration and spectral line intensities are reviewed and then applied to copper vapor in an inert gas. These calculations, while straightforward for atomic spectral lines in optically thin plasmas, are rather tedious, particularly in view of the iteration necessary even for this rather simplified case; therefore a digital computer was used.

In order to make meaningful estimates of the required sensitivity of photomultipliers, amplifiers, and other light detectors, it is necessary to have quantitative information on the intensity of radiation that will be measured for the range of temperatures, pressures, and mole fractions that are expected.

The intensities of the spectral lines Cu I λ 5106 and Cu I λ 5153 are calculated for the idealized case of copper vapor in a hot inert gas. All electrons are presumed to originate from the single ionization of copper. Double ionization of copper has been neglected for the range of conditions covered here. This assumption gives an error of less than 1/2% at $10,000^{\circ}$ K.

The partition functions for Cu I and Cu II, ionization potential lowering, Debye radius, and species concentrations are also tabulated. This allows computation of other copper spectral line intensities if values for their transition probabilities are known.

It should be possible to use the calculations to help determine contamination levels of copper electrode material introduced in plasma jets, as well as temperatures within the plasma.

The internal partition function tabulation will permit calculation of the contribution of the internal degrees of freedom to the various thermodynamic functions.

2 FUNDAMENTAL EQUATIONS

2.1. INTENSITY OF AN ATOMIC SPECTRAL LINE

The radiant power from a spectral line emitted in all directions (4π sterad) from a unit volume of gas, arising from a transition between two energy states of an atom, is equal to the product of the number of transitions per unit time between the two states, and the energy change involved in the transition between the two states.

The number of transitions per second between an upper state m and a lower state n is given by the product of the number of atoms per cm^3 in the upper state m , and the transition probability per second from state m to state n .

If the number density of atoms in the upper state m is $n_m \text{ cm}^{-3}$, and the transition probability between upper state m and lower state n is $A_n^m \text{ sec}^{-1}$, then the number of transitions per second between upper state m and lower state n is

$$n_m A_n^m \text{ cm}^{-3} \text{ sec}^{-1}$$

The energy change of the transition between an upper state of energy E_m and a lower state of energy E_n is given by

$$E_m - E_n = h\nu \text{ ergs}$$

where ν is the frequency of emitted radiation and h is Planck's constant.

Therefore, the radiant intensity per cm^3 of gas emitted into 1 sterad from a spectral line is [4]

$$I = \frac{1}{4\pi} h\nu A_n^m n_m \text{ erg sec}^{-1} \text{ cm}^{-3} \text{ sterad}^{-1} \quad (1)$$

At thermal equilibrium, the number density of particles in energy state i , compared to the number density of particles in energy state m , is given by the Boltzmann energy distribution as [5, 6]

$$\frac{n_i}{n_m} = \frac{g_i}{g_m} e^{\frac{-E_i/kT}{-E_m/kT}}$$

where g is the statistical weight or degeneracy of energy level E_i , and k is the Boltzmann constant. Summing n_i over all energy levels in the same ionization state, we find the number density of atoms in the q -th state of ionization to be [4, 5]

$$\begin{aligned} n_q &= \sum_i n_i = \frac{n_m}{g_m e^{-E_m/kT}} \sum_i g_i e^{-E_i/kT} \\ &= \frac{n_m Z_q}{g_m e^{-E_m/kT}} \text{ cm}^{-3} \end{aligned} \quad (2)$$

where

$$Z_q = \sum_i g_i e^{-E_i/kT} \quad (3)$$

is the internal partition function for atoms in the q -th state of ionization. Therefore,

$$n_m = \frac{n_q g_m e^{-E_m/kT}}{Z_q} \text{ cm}^{-3} \quad (4)$$

Substituting Equation (4) into Equation (1), we find [4]

$$I = \frac{n_q h\nu}{Z_q} \frac{g_m A_n^m e^{-E_m/kT}}{4\pi} \text{ erg sec}^{-1} \text{ cm}^{-3} \text{ sterad}^{-1} \quad (5)$$

Values of E_m are tabulated in Reference 7, and values of $g_m A_n^m$ can be found in Reference 8. Other compilations and bibliographies of transition probabilities can be found in References 9, 10, and 11.

2.2. SAHA EQUATION

For a given element, the number density of atoms in successive ionization states at thermal equilibrium is given by the Saha Equation [4, 12, 13]:

$$\frac{n_{q+1} n_e}{n_q} = \frac{2Z_{q+1}}{Z_q} \frac{(2\pi m_e kT)^{3/2}}{h^3} e^{-V_q/kT} \quad (6)$$

where n_q = number density of atoms in the q -th ionization state, cm^{-3}

n_e = number density of electrons, cm^{-3}

Z_q = partition function for atoms in q -th ionization state

m_e = electron mass

V_q = ionization energy necessary to ionize an atom from the q -th state of ionization to the $(q + 1)$ -st ionization state within the plasma.

Note that the mass of the atom in the q -th state of ionization, m_q , has been taken equal to m_{q+1} and $Z_e = 2$.

Considering single ionization of neutral atoms ($q = 0$), Equation 6 becomes

$$\frac{n_+ n_e}{n_0} = \frac{2Z_+}{Z_0} \frac{(2\pi m_e kT)^{3/2}}{h^3} e^{-V_0/kT} \quad (7)$$

Within the plasma we have charge neutrality [14]

$$n_e = \sum_q q n_q \quad (8)$$

where the sum is taken over all ionization states, $q = 0, 1, 2, \dots$

Equations 5, 6, and 8 are the fundamental equations necessary for the calculation of species concentration and spectral-line intensities. They can be manipulated in several ways for convenience in a particular application.

3

COPPER SPECTRAL LINES, Cu I λ 5106 AND Cu I λ 5153

The Cu I λ 5106 and Cu I λ 5153 lines are of interest here for the following reasons:

1. Their wavelengths are nearly equal, which will minimize problems of spectral sensitivity of a detector.

2. Their upper energy levels have relatively large differences, so that the population ratio of these levels is a relatively strong function of temperature, through the Boltzmann factor.
3. Neither line is a ground-state transition; thus problems of self-reversal of the central radiation by the relatively cold outer atoms is avoided.
4. They have relatively well-determined transition probabilities.
5. They arise from neutral copper atoms so that partition functions do not enter into their relative intensity.
6. They have reasonable intensities and intensity ratios for temperatures in the range 3000° to 10,000°K.

The intensity ratio of these two lines is calculated below. Their absolute values are tabulated in the appendix, having been calculated from Equation 5.

The intensity ratio of two lines, 1 and 2, arising from an atom in the same ionization state ($n_{01} = n_{02}$, $Z_{01} = Z_{02}$) is, from Equation 5,

$$\frac{I_1}{I_2} = \frac{\nu_1}{\nu_2} \frac{g_{m1} A_{n1}^m}{g_{m2} A_{n2}^m} \exp\left(-\frac{E_{m1} - E_{m2}}{kT}\right) \quad (9)$$

so that

$$\log \frac{I_1}{I_2} = \log \frac{\nu_1}{\nu_2} + \log \frac{(g_m A_n^m)_1}{(g_m A_n^m)_2} - \frac{E_{m1} - E_{m2}}{kT} \log e$$

Assuming the index of refraction for the two wavelengths as measured in air is nearly constant, then

$$\nu_1/\nu_2 = \lambda_2/\lambda_1$$

If E_m is measured in cm^{-1} , then

$$\log \frac{I_1}{I_2} = \log \frac{(g_m A_n^m)_1}{(g_m A_n^m)_2} - \log \frac{\lambda_1}{\lambda_2} - \frac{0.624856(E_{m1} - E_{m2})}{T} \quad (10)$$

For Cu I [7, 8] (see also constants in Symbols List),

$$\lambda_1 = 5153.24$$

$$\lambda_2 = 5105.54$$

$$(gA)_1 = 4.7 \times 10^8 \text{ sec}^{-1}$$

$$(gA)_2 = .051 \times 10^8 \text{ sec}^{-1}$$

$$E_{m1} = 49935.2 \text{ cm}^{-1}$$

$$E_{m2} = 30783.7 \text{ cm}^{-1}$$

If we denote L/I_2 by $[\lambda 5153 \text{ Cu I}] / [\lambda 5106 \text{ Cu I}]$, Equation 10 becomes

$$\log \left[\frac{\lambda 5153 \text{ Cu I}}{\lambda 5106 \text{ Cu I}} \right] = 1.96049 - \frac{11966.9}{T} \quad (11)$$

We can also write this as

$$T = \frac{11966.9}{1.96049 - \log \left[\frac{\lambda 5153 \text{ Cu I}}{\lambda 5106 \text{ Cu I}} \right]} \quad (12)$$

Figure 1 is a plot of Equation 11.

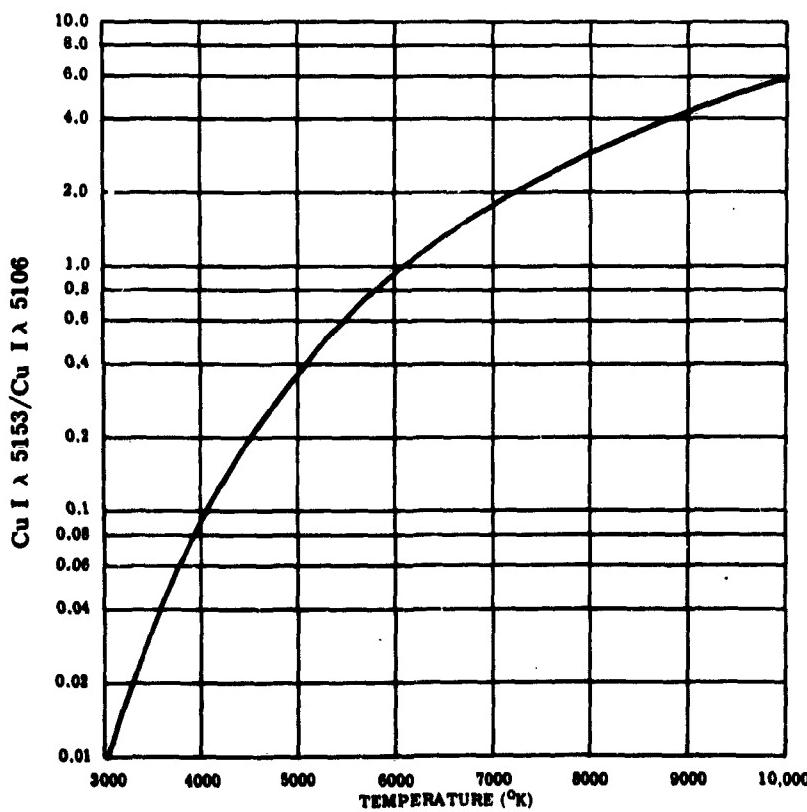


FIGURE 1. INTENSITY RATIO VS. TEMPERATURE

4
**IONIZATION-POTENTIAL LOWERING AND TRUNCATION
 OF THE PARTITION FUNCTION SUM**

Because of the interaction of an atom with the nearby particles in a plasma, the ionization potential will be lowered for a particle inside a plasma. Physically, an excited atom inside a plasma can spontaneously ionize because of the perturbation of its neighbors if its energy state is between this lowered ionization potential and the ordinary ionization limit of an unperturbed atom.

Coupled with the physical process described above is the observation that for large temperatures the exponential factor in the internal partition function sum (Equation 3)

$$Z = \sum_K g_K e^{-E_K/kT}$$

approaches unity, allowing the sum to diverge as more and more terms are taken. The degeneracy g of the state having principle quantum number n is $2n^2$ [15]. A method for avoiding this difficulty is to truncate the partition function sum. The amount that the energy level of the last term taken in the truncated sum lies below the ordinary ionization limit corresponds to the lowering of the ionization potential.

Several criteria have been put forward as to where to truncate the partition function and how much to lower the ionization potential [4, 16, 17]. The actual value of the partition function is rather insensitive to the particular cutoff criterion used. The partition function sum will be truncated when the classical semimajor axis of the electron orbit is greater than or equal to one-half the Debye screening distance. The results obtained here by physical reasoning agree with the Debye-Hückel theory for dilute plasmas [18]. Good experimental confirmation of this Debye distance-truncation criterion for argon plasmas has been shown [19].

The semimajor axis of a classical electron orbit is [15]

$$a = \frac{\zeta e^2}{2W} \text{ cm} \quad (13)$$

where ζ is the effective nuclear charge ($\zeta = 1$ for atoms, $\zeta = 2$ for singly ionized ions, etc.), e is the electron charge, and W is the energy of the electron state in ergs measured downward from the ionization limit. If we measure energies upward from the ground state, which lies

W_{ion} below the ionization limit, we replace W by $W_{ion} - W$. Replacing energies W measured in ergs by term values E measured in cm^{-1} , $E = W/hc$

$$a = \frac{\zeta e^2}{2hc(E_{ion} - E)} \quad (14)$$

Evaluating constants we find

$$a = \frac{5.80699 \times 10^{-4}}{E_{ion} - E} \zeta \text{ cm} \quad (15)$$

as the semimajor axis of a state having term value E of an atom with ionization term value E_{ion} .

The Debye radius is given by [14, 18]

$$\lambda_d = \sqrt{\frac{kT}{4\pi e^2 \left[n_e + \sum n_i (\zeta_i - 1)^2 \right]}} = 6.90087 \sqrt{\frac{T}{n_e + \sum n_i (\zeta_i - 1)^2}} \quad (16)$$

since we are considering the screening of a stationary charge. The sum is over all the positive-ion species. If only singly charged ions are present, $n_e = n_+$

$$\lambda_d = 4.87965 \sqrt{T/n_e} \quad (17)$$

If we denote by E_d the term value of a state having classical semimajor axis $a = \lambda_d/2$, then from Equation 15

$$E_{ion} - E_d = \frac{1.16140 \times 10^{-3}}{\lambda_d} \zeta \text{ cm}^{-1} \quad (18)$$

Note that E_d may not correspond to any actual term value of the particle. Since $1 \text{ cm}^{-1} = 1.23977 \times 10^{-4} \text{ ev}$, the ionization potential lowering is [18]

$$\Delta V = 1.23977 \times 10^{-4} (E_{ion} - E_d) = \frac{1.43987 \times 10^{-7}}{\lambda_d} \zeta \text{ ev} \quad (19)$$

For plasmas having only single-charged ions present,

$$\Delta V = 2.95075 \times 10^{-8} \sqrt{n_e/T} \zeta \text{ ev} \quad (20)$$

where $\zeta = 1$ for the atoms and $\zeta = 2$ for the positive ions. The ionization potential for a neutral atom inside the plasma is therefore

$$V_{\text{ion}} = V_0 - 2.95075 \times 10^{-8} \sqrt{n_e/T} \text{ ev} \quad (21)$$

The Debye-Hückel pressure is [18, 19]

$$p_d = -\frac{kT}{24\pi\lambda_d^3} \quad (22)$$

For single ionization, using Equation 17,

$$p_d = -1.55517 \times 10^{-26} \frac{n_e^{3/2}}{T} \text{ atm} \quad (23)$$

The total pressure is

$$p = nkT - \frac{kT}{24\pi\lambda_d^3} \quad (24)$$

where n is the total particle density.

5 PARTITION FUNCTION FOR COPPER I AND II

Knowledge of the partition function, (Equation 3),

$$Z = \sum_m g_m e^{-E_m/kT}$$

for a substance enables us to calculate all thermodynamic properties of interest, using statistical thermodynamics [5, 6, 20].

Because of the dependence of spectral-line intensities upon the populations of the various energy levels, the partition function enters directly (Equation 5) into the calculation of these line intensities.

The calculation of the internal partition function is not difficult but is rather tedious for elevated temperatures where large numbers of energy states must be taken into account and an iterative procedure followed for an accurate calculation.

As noted in Section 4, the degeneracy of a state with principal quantum number n is $2n^2$. Examination of the number of energy levels listed by Moore [7] shows that many may be missing.

No attempt was made in the work reported here to use isoelectronic sequences [7, 17, 21] to add them.

Levels above the first ionization limit for Cu I, at 62317.2 cm^{-1} , were neglected. Lines arising from states above this limit are quite broad [22] because of the short lifetimes. Spontaneous ionization has a high probability, and for the purposes of the partition-function calculation an atom in these states was considered as ionized. This point requires further study.

Figure 2 is a plot of the internal partition function of Cu I and Cu II; the data are taken from the appendix. The values of the partition functions are functions of temperature alone for the range of conditions covered here.

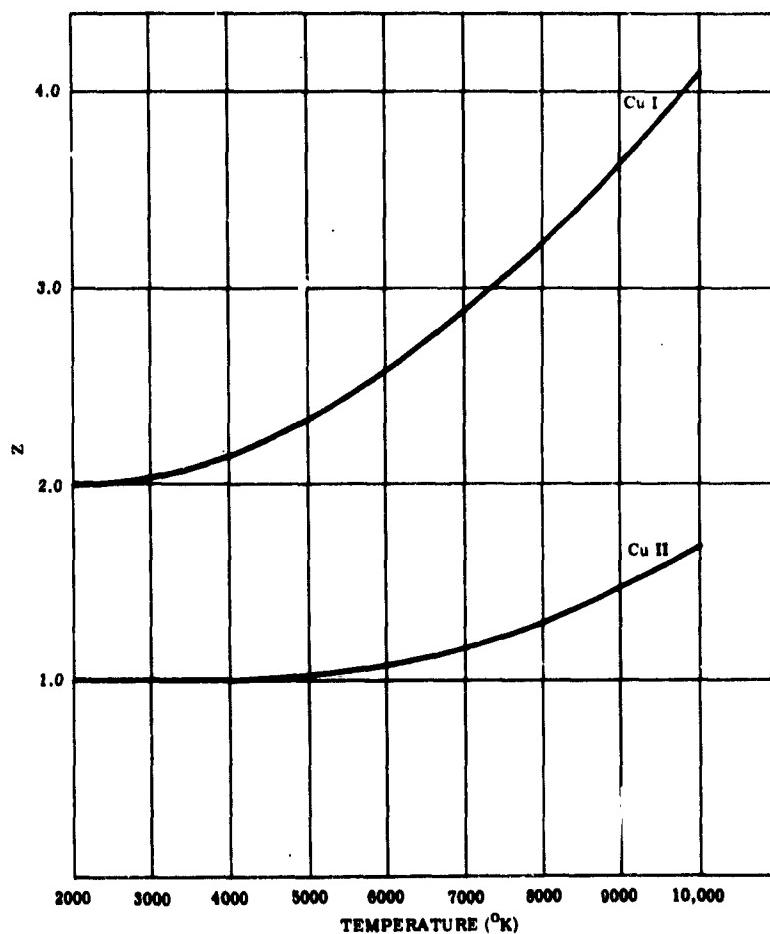


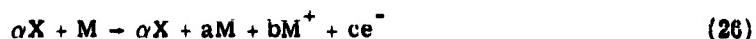
FIGURE 2. INTERNAL PARTITION FUNCTION

6 SINGLE IONIZATION OF AN ELEMENT IN AN INERT DILUENT

Consider 1 mole of element M undergoing the ionization reaction



together with α moles of an inert diluent X. The reaction is written



where a, b, c are the moles of M, M^+ , e^- , respectively, at equilibrium.

We assume each species acts as a perfect gas, with partial pressure of species s given by [5]

$$p_s = x_s p \quad (27)$$

where x_s is the mole fraction of species s and p is the total pressure.

The equilibrium constant K_p is written [5]

$$K_p = \frac{p_e p_+}{p_M} = \frac{x_e x_+}{x_M} p \quad (28)$$

where p is the total pressure in atmospheres, due to particles.

Let N = original mole fraction of species M

$$N = \frac{1}{1 + \alpha} \quad (29)$$

We require charge neutrality, so

$$b = c \quad (30)$$

The total number of moles at equilibrium is

$$\alpha + a + b + c = M_T \quad (31)$$

Conserving total mass of species M,

$$M = aM + bM^+ \quad (32)$$

Assuming $M = M^+$, i.e., the atomic mass of species M equals the ionic mass of species M^+ , Equation 32 becomes

$$1 = a + b \quad (33)$$

The total number of moles, Equation 31, becomes, from Equations 29, 30, 31, and 33,

$$M_T = \frac{1}{N} + b \quad (34)$$

at equilibrium. The neutral mole fraction is

$$x_M = \frac{a}{M_T} = \frac{1 - b}{(1/N) + b} \quad (35)$$

and the ionized mole fraction is

$$x_+ = \frac{b}{M_T} = \frac{b}{(1/N) + b} \quad (36)$$

$$x_+ = x_e \quad (37)$$

by Equation 30. Let $x_+ = x_e = x$, and solving Equation 36 for b,

$$b = \frac{x}{N(1 - x)} \quad (38)$$

so that Equation 35 becomes

$$x_M = \frac{1 - \frac{x}{N(1 - x)}}{\frac{1}{N} + \frac{x}{N(1 - x)}} = N - (N + 1)x \quad (39)$$

Substituting Equation 39 into Equation 28 yields

$$K_p = \frac{x^2 p}{N - (N + 1)x}$$

Solving for x and taking the positive square root since $x > 0$, we find

$$x = \frac{K_p(N + 1)}{2p} \left[\sqrt{1 + \frac{4pN}{K_p(N + 1)^2}} - 1 \right] \quad (40)$$

Whenever

$$C = \frac{4pN}{K_p(N + 1)^2} \ll 1 \quad (41)$$

Equation 40 can be written

$$x = \frac{2N}{N + 1} \left[\frac{1}{2} - \frac{1}{8}C + \frac{1}{16}C^2 + \dots \right] \quad (42)$$

The neutral mole fraction is then given by Equation 39.

The inert mole fraction is obtained using

$$x_{\alpha} = 1 - (x_+ + x_e + x_M) \quad (43)$$

To obtain K_p we proceed as follows. The Saha equation (Equation 7) becomes

$$\frac{n_+ n_e}{n_M} = \frac{(2\pi m_e kT)^{3/2}}{h^3} \frac{2Z_+}{Z_M} e^{-V_0/kT} \quad (44)$$

where V_0 is the lowered ionization potential of the neutral atom, n_+ , n_e , n_M are the number densities of ions, electrons, and neutral atoms, and Z_+ , Z_M are the internal partition functions of ions and neutral atoms.

Using the perfect gas law for the partial pressure of species s [5],

$$p_s = n_s kT \quad (45)$$

We can rewrite Equation 44 as

$$\frac{p_+ p_e}{p_M} = (2\pi m_e)^{3/2} \left[\frac{(kT)^{5/2}}{h^3} \right] \frac{2Z_+}{Z_M} e^{-V_0/kT} \text{ dynes cm}^{-2} \quad (46)$$

to have the same form as Equation 28. Changing pressure units in Equation 46 from dynes cm^{-2} to atmospheres, and evaluating constants gives

$$K_p = (3.28976 \times 10^{-7}) T^{5/2} \frac{2Z_+}{Z_M} e^{-11605.4 V_0/T} \quad (47)$$

where K_p has units of atmosphere, T is measured in $^{\circ}\text{K}$ and V_0 is measured in electron volts.

We can also write Equation 47 in the form

$$\log K_p = -6.46284 + \frac{5}{2} \log T + \log \left(\frac{2Z_+}{Z_M} \right) - 5040.16 \frac{V_0}{T} \quad (48)$$

The total pressure is given by Equation 24:

$$p = nkT - \frac{kT}{24\pi\lambda_d^3}$$

Noting that $n_e = x_e n$, and evaluating constants, we can rewrite Equation 24:

$$0.734002 \times 10^{22} \frac{p}{T} + 1.14150 \times 10^{-4} \left(\frac{nx_e}{T} \right)^{3/2} - n = 0 \quad (49)$$

Since the Debye-Hückel correction must be small to be valid, an approximate root of Equation 49 is given by

$$n = p/kT = (0.734002 \times 10^{22})p/T \quad (50)$$

This estimate may be improved using Newton's method. Thus

$$n_{j+1} = n_j \frac{[(0.734002 \times 10^{22})p/T] + [1.1415 \times 10^{-4}(x_e n_j/T)^{3/2}] - n_j}{(1.71225 \times 10^{-4})(x_e/T)^{3/2} \sqrt{n_j} - 1} \quad (51)$$

where n_{j+1} is the improved value for the root n of Equation 49, using the results of the j -th approximation, n_j .

After satisfactory convergence of Equation 51 has given a value of n , the electron density is given by

$$n_e = x_e n \quad (52)$$

from which the Debye radius and ionization-potential lowering Equation 19 can be computed, since the temperature is known.

7 APPLICATION OF SPECTRAL-LINE INTENSITY MEASUREMENT TO TEMPERATURE AND COMPOSITION DETERMINATION

As a simple illustration we consider the single ionization of an easily ionized impurity in an inert carrier gas. We shall measure an impurity atom to impurity atom line ratio R_0 and an impurity atom to impurity ion line ratio R_+ . The subscript zero denotes the neutral state, and the subscript plus the ionized state. Let the two neutral atomic lines arise from transitions with upper upper levels E_{m1} and E_{m2} ; the ionic line has upper level E_{m+} . We shall assume the values of ν , $g_m A_n^m$, and E_m are known and that $Z_0(T)$, $Z_+(T)$ are known functions of the temperature T .

From Equation 5: the line-intensity ratios are

$$R_0 = \frac{I_1}{I_2} = \frac{\nu_1 (g_m A_n^m)_1}{\nu_2 (g_m A_n^m)_2} \exp \left[-\frac{(E_{m1} - E_{m2})}{kT} \right] \quad (53)$$

and

$$R_+ = \frac{I_1}{I_+} = \frac{n_0 \nu_1 (g_m A_n^m)_1}{n_+ \nu_+ (g_m A_n^m)_+} \frac{Z_+}{Z_0} \exp \left\{ \frac{[E_{m1} - (E_{m+} + V_0)]}{kT} \right\} \quad (54)$$

where V_0 is the impurity ionization energy. Note that E_{m+} is measured with respect to the ground state of the ion which lies V_0 above the ground state of the atom. If

$$\alpha = \frac{(2\pi m_e k)^{3/2}}{h^3} = 2.41469 \times 10^{15} \quad (55)$$

then from the Saha equation (Equation 7) and Equation 8,

$$\frac{n_+^2}{n_0} = \frac{2Z_+}{Z_0} \alpha T^{3/2} e^{-V_0/kT} \quad (56)$$

Equations 53, 54, and 56 constitute three equations in the three unknowns, n_0 , n_+ and T . If

$$a_0 = \frac{\nu_1 (g_m A_n^m)_1}{\nu_2 (g_m A_n^m)_2} \quad (57)$$

$$a_+ = \frac{\nu_1 (g_m A_n^m)_1}{\nu_+ (g_m A_n^m)_+} \quad (58)$$

$$\beta = Z_+/Z_0 \quad (59)$$

and Equation 53 is solved for T ,

$$T = \frac{E_{m1} - E_{m2}}{k \ln(a_0/R_0)} \quad (60)$$

Equation 54 then becomes

$$R_+ = \frac{n_0 a_+}{n_+} \beta e^{-(E_{m1} - E_{m+} - V_0)/kT} \quad (61)$$

and Equation 56 becomes

$$\frac{n_+^2}{n_0} = 2\alpha \beta T^{3/2} e^{-V_0/kT} \quad (62)$$

From Equations 61 and 62, we find the impurity ion density to be

$$n_+ = \frac{2\alpha R_+ T^{3/2}}{a_+} e^{-(E_{m+} - E_{m1} + 2V_0)/kT} \quad (63)$$

In view of the charge neutrality

$$n_+ = n_e \quad (64)$$

The neutral impurity density is

$$n_0 = \frac{n_e^2 V_0 / kT}{2\alpha \beta T^{3/2}} \quad (65)$$

A further measurement of a carrier gas-impurity atom line ratio R_c would give

$$R_c = \frac{I_c}{I_0} = \frac{n_c Z_0}{n_0 Z_c} a_c e^{-(E_{mc} - E_{m1})/kT} \quad (66)$$

where the subscript c = carrier and

$$a_c = \frac{\nu_c (g_m A_n^m)_c}{\nu_1 (g_m A_n^m)_1} \quad (67)$$

Since T is known from Equation 60,

$$n_c = \frac{R_c n_0 Z_c}{a_c Z_0} e^{(E_{mc} - E_{m1})/kT} \quad (68)$$

The pressure is

$$p = (n_c + n_0 + n_+ + n_e) kT \quad (69)$$

The mole fraction of neutral impurity is

$$x_0 = \frac{n_0}{n_0 + n_+ + n_c + n_e} \quad (70)$$

and the mole fraction of impurity ions is

$$x_+ = \frac{n_+}{n_0 + n_+ + n_c + n_e} = x_e \quad (71)$$

The total mole fraction of impurity is $x_+ + x_0$.

Note that the assumption of thermal equilibrium permitted the same temperature to be used in all equations. Note also that E_{mc} and E_{m1} must be referred to a common zero.

8 METHOD OF CALCULATION

Equations 3, 16, 20, 40, 48, and 51 were programmed for an IBM 7090 in MAD (23), according to the following scheme. The pressure, temperature, and original mole fraction were read. The partition functions for Cu I and Cu II were calculated by truncating the partition-function sum when the terms became less than 10^{-6} . From these trial partition functions, a trial K_p was calculated, assuming no ionization potential lowering. This trial K_p permitted a preliminary calculation of n_e , E_d , and λ_d . These values were used to re-evaluate the partition functions, and the sum was truncated when the terms became less than 10^{-6} or the term values were higher than the lowered ionization potential. Second values of K_p , n_e , λ_d , and lowered ionization potential were then obtained. When successive iteration of this process resulted in a fractional change in λ_d of less than 10^{-5} , satisfactory convergence was assumed and the rest of the particle densities and spectral-line intensities were calculated. New values of pressure, temperature, and mole fractions were read, and the process was repeated.

**Appendix
COMPUTER PROGRAM AND TABULATION**

Values of the internal partition function (Equation 3) for neutral copper (Cu I) and singly ionized copper (Cu II), the ionization-potential lowering of Cu I, Debye radius (Equation 17), Debye pressure correlation (Equation 22), equilibrium constant (Equation 47), particle-number densities, and absolute spectral-line intensities for Cu I λ 5106 and Cu I λ 5153 are presented as functions of temperature, original mole fraction, and total pressure.

2000 < T < 10,000°K
0.03 < p < 10 atm
0.0001 < N < 0.03

The number $.2002 \times 10^1$ is printed out as .2002E01, and $.1192 \times 10^{-1}$ is printed out as .1192E-01.

TEMP CEG K	PRESSURE = .13 ATM. INTERNAL PARTITION FUNCTION CuI	ORIGINAL MOLE FRACTION OF COPPER = .0001 CuI	LUMINATION POTENTIAL CuI, CN	DEBYE RADIUS, CM	DEBYE PRESSURE, ATM
2000	.2002E 01	.1000E 01	.00000	.7285CL-C1	-.93470E-17
2500	.2011E 01	.1000E 01	.00002	.84291E-C2	-.75428E-14
3000	.2035E 01	.1000E 01	.00007	.20322E-C2	-.64606E-12
3500	.2077E 01	.1001E 01	.00019	.74798E-C3	-.15112E-10
4000	.2141E 01	.1055E 01	.00039	.36582E-C3	-.14764E-09
4500	.2225E C1	.1012E 01	.00063	.22671E-C3	-.67969E-09
5000	.2329E 01	.1024E 01	.00077	.18606E-03	-.14027E-08
5500	.2449E 01	.1045E 01	.00078	.18571E-03	-.15517E-08
6000	.2563E 01	.1074E 01	.00073	.19843E-C3	-.13863E-08
6500	.2731E 01	.1113E 01	.00067	.21419E-C3	-.11968E-08
7000	.2890E 01	.1163E 01	.00063	.23031E-03	-.10353E-08
7500	.3061E C1	.1224E 01	.00058	.24668E-C3	-.90277E-09
8000	.3244E C1	.1297E 01	.00055	.26310E-03	-.79371E-09
8500	.3439E 01	.1379E 01	.00052	.27953E-03	-.70317E-09
9000	.3643E 01	.1473E 01	.00049	.29597E-03	-.62724E-09
9500	.3873E 01	.1575E 01	.00046	.31241E-C3	-.56296E-09
10000	.4115E 01	.1687E 01	.00044	.32885E-C3	-.50808E-09

TEMP DEG K	PRESSURE = .03 ATM.		ORIGINAL MOLE FRACTION OF COPPER = .0001		ATM
	TOTAL PARTICLES PER CC.	PARTICLES ATOMS PER CC.	CU ATOMS PER CC.	CU ICNS. ELECTRONS PER CC.	
2000	.1101E 16	.1101E 16	.1101E 14	.8973E 07	.1993E-17
2500	.6808E 17	.8807E 17	.8807E 13	.6378E 09	.2715E-13
3000	.7340E 17	.7339E 17	.7323E 13	.1730E 11	.1670E-10
3500	.6291E 17	.6291E 17	.6142E 13	.1490E 12	.1722E-08
4000	.5505E 17	.5504E 17	.4793E 13	.7117E 12	.5759E-07
4500	.4893E 17	.4893E 17	.2845E 13	.2048E 13	.9044E-06
5000	.4404E 17	.4403E 17	.9645E 12	.3439E 13	.8353E-05
5500	.4004E 17	.4003E 17	.2059E 12	.3797E 13	.5247E-04
6000	.3670E 17	.3669E 17	.4348E 11	.3626E 13	.2472E-03
6500	.3388E 17	.3387E 17	.1080E 11	.3377E 13	.9345E-03
7000	.3146E 17	.3145E 17	.3166E 10	.3142E 13	.2974E-02
7500	.2936E 17	.2935E 17	.1067E 10	.2935E 13	.8248E-02
8000	.2753E 17	.2752E 17	.4038E 09	.2752E 13	.2044E-01
8500	.2591E 17	.2590E 17	.1684E 09	.2590E 13	.4615E-01
9000	.2447E 17	.2446E 17	.7628E 08	.2446E 13	.9625E-01
9500	.2314E 17	.2317E 17	.3710E 08	.2310E 13	.1875E 00
10000	.2202E 17	.2202E 17	.1921E 08	.2202E 13	.3446E 00

TEMP CEG K	PRESSURE = .03 ATM. SPECTRAL LINE INTENSITIES, WATTS/ICC. STERI	ORIGINAL MOLE FRACTION OF COPPER = .0001 LAMBDA 5105.5
2000	.2993E-09	.1985E-13
2500	.1397E-07	.7084E-10
3000	.2200E-C6	.2060E-08
3500	.1490E-05	.5182E-07
4000	.5487E-05	.5106E-06
4500	.1072E-04	.2145E-05
5000	.9296E-05	.3431E-05
5500	.4723E-C5	.2572E-05
6000	.1654E-C5	.1529E-05
6500	.6858E-06	.9028E-06
7000	.3090E-C6	.5506E-06
7500	.1499E-06	.3472E-06
8000	.7743E-C7	.2257E-06
8500	.4218E-07	.1506E-06
9000	.2406E-07	.1028E-06
9500	.1428E-C7	.7171E-07
10000	.8786E-08	.5100E-07

TEMP DEG K	PRESSURE= .03 ATM.	INTERNAL PARTITION FUNCTION CUI	IONIZATION POTENTIAL LOWERING FOR CUI, EV	DEBYE RADIUS, CM	DENSYE PRESSURE, ATM
2000	.2002E 01	.1000E 01	.00000	.55354E-C1	-.21307E-16
2500	.2001E 01	.1000E 01	.00002	.64047E-C2	-.17195E-13
3000	.2005E 01	.1000E 01	.00009	.15436E-C2	-.1439E-11
3500	.2007E 01	.1001E 01	.00025	.56688E-03	-.34717E-10
4000	.2141E 01	.1005E 01	.00053	.27390E-03	-.35173E-09
4500	.2225E 01	.1012E 01	.00086	.16418E-C3	-.18372E-08
5000	.2329E 01	.1024E 01	.00118	.12201E-03	-.49747E-08
5500	.2449E 01	.1045E 01	.00129	.11195E-C3	-.70839E-08
6000	.2583E 01	.1074E 01	.00124	.11591E-C3	-.69615E-08
6500	.2731E 01	.1113E 01	.00116	.124C1E-C3	-.61587E-08
7000	.2890E 01	.1163E 01	.00108	.13312E-03	-.53621E-08
7500	.3061E C1	.1224E 01	.00101	.14249E-C3	-.46844E-08
8000	.3244E 01	.1297E 01	.00095	.15194E-03	-.41212E-08
8500	.3439E 01	.1379E 01	.00089	.16142E-03	-.36520E-08
9000	.3648E 01	.1473E 01	.00084	.17090E-03	-.32579E-08
9500	.3873E 01	.1575E 01	.00080	.18039E-03	-.29242E-08
10000	.4115E 01	.1687E 01	.00076	.18988E-03	-.26392E-08

TEMP C°G K	PRESSURE= .03 ATM.		ORIGINAL MOLL FRACTION OF COPPER= .0003		
	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	ATM CONSTANT,
2000	.1101E 18	.1101E 18	.3303E 14	.1554E 08	.1993E-17
2500	.8808E 17	.8605E 17	.2642E 14	.1451E 10	.2715E-13
3000	.7340E 17	.7338E 17	.2199E 14	.2998E 11	.1671E-10
3500	.6291E 17	.6290E 17	.1851E 14	.2593E 12	.1723E-08
4000	.5505E 17	.5503E 17	.1525E 14	.1270E 13	.5761E-07
4500	.4893E 17	.4891E 17	.1070E 14	.3975E 13	.9044E-06
5000	.4404E 17	.4402E 17	.5212E 13	.7998E 13	.8361E-05
5500	.4014E 17	.4001E 17	.1558E 13	.1045E 14	.5253E-04
6000	.3670E 17	.3669E 17	.3735E 12	.1063E 14	.2474E-03
6500	.3388E 17	.3186E 17	.9585E 11	.1006E 14	.9353E-03
7000	.3146E 17	.3144E 17	.2835E 11	.9406E 13	.2976E-02
7500	.2936E 17	.2934E 17	.9578E 10	.8796E 13	.8254E-02
8000	.2753E 17	.2751E 17	.3628E 10	.8251E 13	.2046E-01
8500	.2591E 17	.2589E 17	.1514E 10	.7768E 13	.4617E-01
9000	.2447E 17	.2445E 17	.6856E 09	.7337E 13	.9629E-01
9500	.2319E 17	.2317E 17	.3334E 09	.6951E 13	.1876E C0
10000	.2222E 17	.2214E 17	.1725E 09	.6604E 13	.3447E 00

TEMP DEG K	PRESSURE = .03 ATM.	ORIGINAL MOLE FRACTION OF COPPER = .0003	SPECTRAL LINE INTENSITIES, WATTS/(CC STER)
	LAMBDA 5105.5	LAMBDA 5153.2	
2000	.6278E-09	.5954E-13	
2500	.4192E-07	.6253E-10	
3000	.6609E-06	.6188E-08	
3500	.4515E-05	.1570E-06	
4000	.1745E-04	.1624E-05	
4500	.4035E-04	.8071E-05	
5000	.5023E-04	.1854E-04	
5500	.3195E-04	.1946E-04	
6000	.1420E-04	.1313E-04	
6500	.6084E-05	.8010E-05	
7000	.2766E-05	.4929E-05	
7500	.1345E-05	.3117E-05	
8000	.6957E-06	.2020E-05	
8500	.3791E-06	.1353E-05	
9000	.2162E-06	.9741E-06	
9500	.1283E-06	.6444E-06	
10000	.7877E-07	.4580E-06	

TEMP DEG K	PRESSURE = .03 ATM.	INTERNAL PARTITION FUNCTION CUI	ORIGINAL MOLE FRACTION OF COPPER = .9010	IONIZATION POTENTIAL	DEBYE LOWERING FOR CUI, EV	RADIUS, CM	DEBYE PRESSURE, ATM
2000	.2002E 31	.1000E 01	.00000	.40967L-01	.52563E-16		
2500	.2011E 01	.1000E 01	.00003	.47399E-02	.42420E-13		
3000	.2035E 01	.1000E 01	.00013	.11422E-02	.36380E-11		
3500	.2077E 01	.1000E 01	.00034	.41885E-03	.86066E-10		
4000	.2141E 01	.1005E 01	.00072	.20C86E-C3	.89191E-09		
4500	.2225E 01	.1012E 01	.00123	.11724E-03	.50456E-08		
5000	.2329E 01	.1024E 01	.00177	.81523E-04	.16675E-07		
5500	.2449E 01	.1045E 01	.00212	.67875E-C4	.31851E-07		
6000	.2583E 01	.1074E 01	.00219	.65735E-C4	.38168E-07		
6500	.2731E 01	.1113E 01	.00210	.68665E-04	.36278E-07		
7000	.2890E 01	.1153E 01	.00197	.73189E-04	.32263E-07		
7500	.3061E 01	.1224E 01	.00184	.76179E-C4	.28372E-07		
8000	.3244E 01	.1297E 01	.00173	.83292E-C4	.25016E-07		
8500	.3439E 01	.1379E 01	.00163	.88462E-04	.22187E-07		
9000	.3648E 01	.1473E 01	.00154	.93650E-C4	.1980CE-07		
9500	.3873E 01	.1575E 01	.00146	.98845E-04	.17775E-07		
10000	.4115E 01	.1687E 01	.00138	.10434E-C3	.16043E-07		

TEMP DEG K	PRESSURE = .03 ATM. PARTICLES PER CC.	.03 ATM. TOTAL PARTICLES PER CC.	ORIGINAL MOLE FRACTION OF COPPER = .0310 INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU ICNS. ELECTRONS PER CC.	EQUILIBRIUM CONSTANT. ATM
2000	.1101E 18	.1103E 18	.1101E 15	.2838E 08	.1993E-17	
2500	.8808E 17	.8799E 17	.8808E 14	.2650E 10	.2715E-13	
3000	.7340E 17	.7333E 17	.7335E 14	.5476E 11	.1671E-10	
3500	.6291E 17	.6285E 17	.6244E 14	.4750E 12	.1723E-08	
4000	.5505E 17	.5499E 17	.5269E 14	.2361E 13	.5764E-07	
4500	.4893E 17	.4988E 17	.4113E 14	.7795E 13	.9058E-06	
5000	.4404E 17	.4398E 17	.2611E 14	.1791E 14	.8372E-05	
5500	.4004E 17	.3997E 17	.1154E 14	.2847E 14	.5262E-04	
6000	.3670E 17	.3663E 17	.3605E 13	.3306E 14	.2479E-03	
6500	.3348E 17	.3381E 17	.1018E 13	.3283E 14	.9369E-03	
7000	.3146E 17	.3139E 17	.3098E 12	.3112E 14	.2981E-02	
7500	.2936E 17	.2930E 17	.1056E 12	.2923E 14	.8264E-02	
8000	.2753E 17	.2747E 17	.4013E 11	.2746E 14	.2048E-01	
8500	.2591E 17	.2585E 17	.1676E 11	.2586E 14	.4622E-01	
9000	.2447E 17	.2442E 17	.7597E 10	.2443E 14	.9638E-01	
9500	.2318E 17	.2313E 17	.3695E 10	.2315E 14	.1878E 00	
10000	.2202E 17	.2195E 17	.1911E 10	.2200E 14	.3450E 00	

TEMP	PR(SUPP)= .003 ATM.	INITIAL MOLT. FRACTION OF COPPER= .0010
CLG X	LAMADA 5105.5	LAMADA 5153.2
2000	.2693E-08	.1985E-12
2500	.1397E-06	.2084E-09
3000	.2794E-05	.2064E-07
3500	.1515E-04	.5268E-06
4000	.6031E-04	.5612E-05
4500	.1550E-03	.3101E-04
5000	.2516E-03	.9766E-04
5500	.2367E-03	.1441E-03
6000	.1371E-03	.1267E-03
6500	.6465E-04	.8511E-04
7000	.3073E-04	.5387E-04
7500	.1493E-04	.3437E-04
8000	.7694E-05	.2243E-04
8500	.4198E-05	.1498E-04
9000	.2396E-05	.1024E-04
9500	.1472E-05	.7142E-05
10000	.8744E-06	.5075E-05

TEMP DEG K	PRESSURE = .03 ATM.	ORIGINAL MOLE FRACTION OF COPPER = .0030	INTERNAL PARTITION FUNCTION	IONIZATION POTENTIAL	DERIVE	DEBYE	PRESSURE, ATM
			CUI	LOGGING FOR CUI, EV	RADIUS, CM		
2000	.2C92E 01	.1000E 01	.00000	.31128E -C1	-.31128E -C1	-.11982E -15	
2500	.2011E 01	.1000E 01	.00004	.36615E -02	-.36615E -02	-.96701E -13	
3000	.2035E 01	.1000E 01	.00017	.86776E -03	-.86776E -03	-.82958E -11	
3500	.2077E 01	.1001E 01	.00045	.31798E -03	-.31798E -03	-.19671E -09	
4000	.2141E 01	.1005E 01	.00095	.15189E -03	-.15189E -03	-.20625E -08	
4500	.2225E 01	.1012E 01	.00165	.87448E -04	-.87448E -04	-.12159E -07	
5000	.2329E 01	.1024E 01	.00246	.58637E -04	-.58637E -04	-.44811E -07	
5500	.2449E 01	.1045E 01	.00316	.45525E -04	-.45525E -04	-.10533E -06	
6000	.2583E 01	.1074E 01	.00353	.40831E -C4	-.40831E -C4	-.15926E -06	
6500	.2731E 01	.1113E 01	.00353	.40758E -04	-.40758E -04	-.17347E -06	
7000	.288CE 01	.1163E 01	.00337	.42697E -04	-.42697E -04	-.16250E -06	
7500	.3061E 01	.1224E 01	.00318	.45335E -04	-.45335E -04	-.14544E -06	
8000	.3244E 01	.1297E C1	.00299	.48206E -04	-.48206E -04	-.12904E -06	
8500	.3439E 01	.1379E 01	.00281	.51157E -04	-.51157E -04	-.11472E -06	
9000	.3648E 01	.1473E 01	.00266	.54139E -04	-.54139E -04	-.10248E -06	
9500	.3873E 01	.1575E 01	.00252	.57134E -04	-.57134E -04	-.92040E -07	
10000	.4115E 01	.1687E C1	.00239	.60135E -04	-.60135E -04	-.83093E -07	

PRESSURE = .03 ATM.		ORIGINAL MOLE FRACTION OF COPPER = .0030		
TEMP CEG K	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC. ATM
2900	.1101E 18	.1098E 18	.3303E 15	.4915E 06
2500	.8058E 17	.8782E 17	.2642F 15	.4589E 10
3000	.7340E 17	.7316E 17	.2291E 15	.9486E 11
3500	.6291E 17	.6272E 17	.1879E 15	.8242E 12
4000	.5505E 17	.5488E 17	.1610E 15	.4128E 13
4500	.4893E 17	.4877E 17	.1327E 15	.1401E 14
5000	.4404E 17	.4387E 17	.9739E 14	.3463E 14
5500	.4004E 17	.3985E 17	.5673E 14	.6319E 14
6000	.3670E 17	.3650E 17	.2415E 14	.8569E 14
6500	.3388E 17	.3368E 17	.8184E 13	.9317E 14
7000	.3146E 17	.3127E 17	.2668E 13	.9143E 14
7500	.2936E 17	.2919E 17	.9315E 12	.8689E 14
8000	.2753E 17	.2736E 17	.3569E 12	.8197E 14
8500	.2591E 17	.2575E 17	.1496E 12	.7734E 14
9000	.2447E 17	.2432E 17	.6791E 11	.7311E 14
9500	.2318E 17	.2304E 17	.3305E 11	.6930E 14
10000	.2202E 17	.2189E 17	.1710F 11	.6585E 14

TEMP DEG K	LAMBDA 5105.5	LAMBDA 5153.2	ORIGINAL MOLE FRACTION OF COPPER= .0030
2000	.6278E-08	.5954E-12	
2500	.4192E-06	.6253E-09	
3000	.6614E-05	.6193E-07	
3500	.4558E-04	.1585E-05	
4000	.1843E-03	.1715E-04	
4500	.5003E-03	.1001E-03	
5000	.9387E-03	.3464E-03	
5500	.1163E-02	.7086E-03	
6000	.9185E-03	.8493E-03	
6500	.5195E-03	.6839E-03	
7000	.2604E-03	.4640E-03	
7500	.1309E-03	.3032E-03	
8000	.6844E-04	.1995E-03	
8500	.3747E-04	.1338E-03	
9000	.2142E-04	.9153E-04	
9500	.1272E-04	.6388E-04	
10000	.7824E-05	.4542E-04	

TEMP DEG K	PRESSURE= .013 ATM.	INTERNAL PARTITION FUNCTION CUI	IONIZATION POTENTIAL LOWERING FOR CUI, EV	DEBYE	DEBYE PRESSURE, ATM
2000	.2932E 01	.1900E 01	.00001	.73037E-C1	-.29558E-15
2500	.2011E 01	.1900E 01	.00005	.26654E-02	-.23857E-12
3000	.2035E 01	.1900E 01	.00022	.64215E-C3	-.20472E-10
3500	.2077E 01	.1001E 01	.00061	.23516E-C3	-.48619E-09
4000	.2141E 01	.1005E 01	.00128	.11207E-03	-.51355E-08
4500	.2225E 01	.1912E 01	.00225	.63968E-04	-.31064E-07
5000	.2329E 01	.1324E 01	.00343	.41921E-04	-.12264E-06
5500	.2449E 01	.1045E 01	.00464	.31007E-C4	-.33336E-06
6000	.2583E 01	.1074E 01	.00558	.25817E-C4	-.63007E-06
6500	.2731E 01	.1113E 01	.00600	.24011E-C4	-.84844E-06
7000	.2890E 01	.1163E 01	.00596	.24151E-04	-.89795E-06
7500	.3061E 01	.1224E 01	.00571	.25206E-04	-.84627E-06
8000	.3244E 01	.1297E 01	.00541	.26623E-C4	-.76603E-06
8500	.3439E 01	.1379E 01	.00511	.28178E-04	-.68645E-06
9000	.3648E 01	.1473E 01	.00483	.29788E-C4	-.61529E-06
9500	.3873E 01	.1575E 01	.00458	.31419E-04	-.55344E-06
10000	.4115E 01	.1687E 01	.00436	.33061E-04	-.50001E-06

TEMP DEG K	PRESSURE = .03 ATM.			ORIGINAL MOLE FRACTION OF COPPER = .0100		
	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS + ELECTRONS PER CC.	CU IONS - ELECTRONS PER CC.	EQUILIBRIUM CONSTANT, ATM
2000	.1101E 19		.1099E 18	.1101E 16	.8973E 08	.1993E-17
2500	.8808E 17		.8720E 17	.8808E 15	.8379E 10	.2715E-13
3000	.7340E 17		.7267E 17	.7338E 15	.1732E 12	.1671E-10
3500	.6291E 17		.6228E 17	.6276E 15	.1507E 13	.1725E-08
4000	.5505E 17		.5449E 17	.5428E 15	.7584E 13	.5774E-07
4500	.4893E 17		.4842E 17	.4629E 15	.2619E 14	.9081E-06
5000	.4404E 17		.4353E 17	.3720E 15	.6775E 14	.8405E-05
5500	.4004E 17		.3950E 17	.2628E 15	.1362E 15	.5290E-04
6000	.3670E 17		.3612E 17	.1505E 15	.2144E 15	.2495E-03
6500	.3388E 17		.3327E 17	.6764E 14	.2685E 15	.9435E-03
7000	.3146E 17		.3086E 17	.2595E 14	.2858E 15	.3001E-02
7500	.2936E 17		.2879E 17	.9710E 13	.2811E 15	.8314E-02
8000	.2753F 17		.2698E 17	.3823E 13	.2687E 15	.2059E-01
8500	.2591E 17		.2540E 17	.1620E 13	.2549E 15	.4644E-01
9000	.2447E 17		.2398E 17	.7389E 12	.2415E 15	.9679E-01
9500	.2318E 17		.2272E 17	.3605E 12	.2291E 15	.1885E 00
10000	.2202E 17		.2158E 17	.1868E 12	.2178E 15	.3462E 00

TEMP DEG K	PRESSURE = .03 ATM.	ORIGINAL MOLE FRACTION OF COPPER = .0100	SPECTRAL LINE INTENSITIES, WATTS/1CC STERI
	LAMBDA 5105.5	LAMBDA 5153.2	
2000	.2093E-07	.1985E-11	
2500	.1397E-05	.2085E-08	
3000	.2205E-04	.2065E-06	
3500	.1522E-03	.5295E-05	
4000	.6214E-03	.5783E-04	
4500	.1745E-02	.3490E-03	
5000	.3585E-02	.1323E-02	
5500	.5369E-02	.3282E-02	
6000	.5724E-02	.5293E-02	
6500	.4294E-02	.5652E-02	
7000	.2513E-02	.4513E-02	
7500	.1364E-02	.3160E-02	
8000	.7331E-03	.2137E-02	
8500	.4057E-03	.1448E-02	
9000	.2330E-03	.9959E-03	
9500	.1388E-03	.6967E-03	
10000	.8544E-04	.4959E-03	

TEMP CEG K	PRESSURE= .03 ATM.	INTERNAL PARTITION FUNCTION CUI	ORIGINAL MOLE FRACTION OF COPPER= .0300 CUI	IONIZATION POTENTIAL LOWERING FOR CUI, EV	DEBYE RADIUS, CM	DEBYE PRESSURE, ATM
2000	.22022E 01	.1000E 01	.00001	.-17504E-01	.-67379E-15	
2500	.2011E 01	.1000E 01	.00007	.-20252E-02	.-54386E-12	
3000	.2035E 01	.1000E 01	.00030	.-48788E-03	.-466779E-10	
3500	.2277E 01	.1001E 01	.00081	.-17863E-03	.-11096E-08	
4000	.2141E 01	.1005E 01	.00169	.-85004E-04	.-11768E-07	
4500	.2225L 01	.1012E 01	.00298	.-48306E-04	.-72135E-07	
5000	.2329E 01	.1024E 01	.00460	.-31285E-04	.-29505E-06	
5500	.2447E 01	.1045E 01	.00638	.-22551E-04	.-86660E-06	
6000	.2583E 01	.1074E 01	.00803	.-17913E-04	.-18800E-05	
6500	.2731E 01	.1113E 01	.00918	.-15690E-04	.-30407E-05	
7000	.2890E 01	.1163E 01	.00963	.-14950E-04	.-37857E-05	
7500	.3061E 01	.1224E 01	.00953	.-15106E-04	.-39316E-05	
8000	.3244E 01	.1297E 01	.00916	.-15714E-04	.-37251E-05	
8500	.3439E 01	.1379E 01	.00871	.-16522E-04	.-34051E-05	
9000	.3648E 01	.1473E 01	.00827	.-17415E-04	.-30789E-05	
9500	.3873E 01	.1575E 01	.00785	.-18345E-04	.-27806E-05	
10000	.4115E 01	.1687E 01	.00746	.-19291E-C4	.-25171E-05	

PRESSURE = .03 ATM.		ORIGINAL MOLT. FRACTION OF COPPER = .0300		EQUILIBRIUM	
TEMP CEG K	PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	CONSTANT, ATM
2000	.1101E 16	.1066E 16	.3303E 16	.1554E 09	.1993E-17
2500	.6806E 17	.8544E 17	.2642E 16	.1451E 11	.2715E-13
3000	.7340E 17	.712CE 17	.2202E 16	.3001E 12	.1672E-10
3500	.6291E 17	.6102E 17	.1885E 16	.2612E 13	.1726E-08
4000	.5505E 17	.5339E 17	.1638E 16	.1318E 14	.5781E-07
4500	.4893E 17	.4742E 17	.1421E 16	.4592E 14	.9094E-06
5000	.4404E 17	.4260E 17	.1196E 16	.1216E 15	.8428E-05
5500	.4004E 17	.3859E 17	.9359E 15	.2575E 15	.5310E-04
6000	.3670E 17	.3517E 17	.6435E 15	.4443E 15	.2507E-03
6500	.3388E 17	.3225E 17	.3689E 15	.6287E 15	.9488E-03
7000	.3146E 17	.2979E 17	.1757E 15	.7458E 15	.3019L-02
7500	.2936E 17	.2772E 17	.7482E 14	.7826E 15	.8363F-02
8000	.2753E 17	.2595E 17	.3133E 14	.7714E 15	.207CE-01
8500	.2591E 17	.2441E 17	.1364E 14	.7414E 15	.4667E-01
9000	.2447E 17	.2302E 17	.6294E 13	.7066E 15	.9722E-01
9500	.2318E 17	.2183E 17	.3089E 13	.6722E 15	.1093E 00
10000	.2272E 17	.2174E 17	.1605E 13	.6399E 15	.3474E 00

TEMP DEG K	PRESSURE= .03 ATM. SPECTRAL LINE INTENSITIES, WATTS/1CC STER)	ORIGINAL MOLE FRACTION OF COPPER=.0300 LAMBDA 5153.2
2000	.6278E-07	.5954E-11
2500	.4192E-05	.6254E-08
3000	.6616E-04	.6195E-06
3500	.4572E-03	.1590E-04
4000	.1875E-02	.1745E-03
4500	.5355E-02	.1071E-02
5000	.1153E-01	.4254E-02
5500	.1919E-01	.1169E-01
6000	.2447E-01	.2263E-01
6500	.2341E-01	.3082E-01
7000	.1714E-01	.3055E-01
7500	.1051E-01	.2435E-01
8000	.6007E-02	.1751E-01
8500	.3416E-02	.1219E-01
9000	.1985E-02	.8483E-02
9500	.1189E-02	.5971E-02
10000	.7344E-03	.4263E-02

PRESSURE = .10 ATM. ORIGINAL MOLE FRACTION OF COPPER = .00001

TEMP DEG K	CU1	CU11	IONIZATION POTENTIAL LOWERING FOR CU1, EV	DEBYE	DEBYE
				PRESSURE, ATM	PRESSURE, ATM
2000	.2002E 01	.1000E 01	.00000	.53915E-01	-.23059E-16
2500	.2011E 01	.1030E 01	.00002	.62381E-02	-.18609E-13
3000	.2035E 01	.1000E 01	.00010	.15034E-02	-.15952E-11
3500	.2077E 01	.1001E 01	.00026	.55204E-03	-.37592E-10
4000	.2141E 01	.1005E 01	.00054	.26651E-03	-.38164E-09
4500	.2225E 01	.1012E 01	.00090	.15927E-03	-.20126E-08
5000	.2329E 01	.1024E 01	.00123	.11751E-03	-.55685E-08
5500	.2449E 01	.1045E 01	.00135	.10686E-03	-.81442E-08
6000	.2583E 01	.1074E 01	.00131	.11015E-03	-.81115E-08
6500	.2731E 01	.1113E 01	.00122	.11770E-03	-.72040E-08
7000	.2890E 01	.1163E 01	.00114	.12629E-03	-.62789E-08
7500	.3061E 01	.1224E 01	.00107	.13517E-03	-.54871E-08
8000	.3244E 01	.1297E 01	.00100	.14413E-03	-.48279E-08
8500	.3439E 01	.1379E 01	.00094	.15312E-03	-.42784E-08
9000	.3648E 01	.1473E 01	.00089	.16212E-03	-.38168E-08
9500	.3873E 01	.1575E 01	.00084	.17112E-03	-.34259E-08
10000	.4115E 01	.1687E 01	.00080	.18012E-03	-.30920E-08

PRESSURE =		10 ATM.		ORIGINAL MOLE FRACTION OF COPPER = .0001			
TEMP	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	ATM	EQUILIBRIUM CONSTANT,	
2000	.3670E 16	.3670E 16	.3670E 16	.3670E 16	.1630E 08	.1993E-17	
2500	.2936E 16	.2936E 16	.2936E 16	.2936E 16	.1530E 10	.2715E-13	
3000	.2447E 16	.2446E 16	.2446E 16	.2444E 16	.3160E 11	.1671E-10	
3500	.2097E 16	.2097E 16	.2097E 16	.2070E 16	.2735E 12	.1723E-06	
4000	.1815E 16	.1835E 16	.1835E 16	.1701E 14	.1341E 13	.5762E-07	
4500	.1631E 16	.1631E 16	.1631E 16	.1269E 14	.4224E 13	.9050E-06	
5000	.1468E 16	.1468E 16	.1468E 16	.6057E 13	.8623E 13	.8362E-05	
5500	.1335E 16	.1334E 16	.1334E 16	.1876E 13	.1147E 14	.5254E-04	
6000	.1223E 16	.1223E 16	.1223E 16	.4579E 12	.1177E 14	.2475E-03	
6500	.1129E 16	.1129E 16	.1129E 16	.1182E 12	.1117E 14	.9355E-03	
7000	.1049E 16	.1048E 16	.1048E 16	.3498E 11	.1045E 14	.2977L-02	
7500	.9787E 17	.9785E 17	.9785E 17	.1183E 11	.9774E 13	.8254E-02	
8000	.9175E 17	.9173E 17	.9173E 17	.4480E 10	.9170E 13	.2046E-01	
8500	.8635E 17	.8634E 17	.8634E 17	.1869E 10	.8633E 13	.4617E-01	
9000	.8156E 17	.8154E 17	.8154E 17	.8468E 09	.8154E 13	.9630E-01	
9500	.7726E 17	.7725E 17	.7725E 17	.4119E 09	.7725E 13	.1877E 00	
10000	.7340E 17	.7339E 17	.7339E 17	.2130E 09	.7339E 13	.3447E 00	

TEMP DEG K	LAMRDA 5105.5	LAMRDA 5153.2	PRESSURE = .10 ATM.	ORIGINAL MOLE FRACTION OF COPPER = .0001
2000			.6976E-09	.6615E-13
2500			.4658E-07	.6948E-10
3000			.7342E-06	.6876E-08
3500			.5C21E-05	.1746E-06
4000			.1947E-04	.1B12E-05
4500			.4556E-04	.9114E-05
5000			.5837E-04	.2154E-04
5500			.3847E-04	.2343E-04
6000			.1741E-04	.1610E-04
6500			.7502E-05	.9876E-05
7000			.3414E-05	.6083E-05
7500			.1661E-05	.3849E-05
8000			.8590E-06	.2504E-05
8500			.4682E-06	.1671E-05
9000			.2671E-06	.1141E-05
9500			.1505E-06	.7960E-06
10000			.9741E-07	.5655E-06

TEMP DEG K	PRESSURE= .10 ATM.	INTERNAL PARTITION FUNCTION CUI	ORIGINAL MOLE FRACTION OF COPPER=.0003 IONIZATION POTENTIAL LOWERING FOR CUI, EV	DEBYE RADIUS, CM	DEBYE PRESSURE, ATM
2000	.2002E 01	.1000E 01	.00000	.40967E-01	-.52563E-16
2500	.2011E 01	.10001E 01	.00003	.47399E-02	-.42420E-13
3000	.2035E 01	.1000E 01	.00013	.11422E-02	-.36380E-11
3500	.2077E 01	.1001E 01	.00034	.41885E-C3	-.86066E-10
4000	.2141E 01	.1005E 01	.00072	.20086E-03	-.89193E-09
4500	.2225E 01	.1012E 01	.00123	.11724E-03	-.50461E-08
5000	.2329E 01	.1024E 01	.00177	.81516E-24	-.16679E-07
5500	.2449E 01	.1045E 01	.00212	.67812E-04	-.31870E-07
6000	.2583E 01	.1074E 01	.00219	.65716E-04	-.38201E-07
6500	.2731E 01	.1113E 01	.00210	.68643E-C4	-.36314E-07
7000	.2890E 01	.1163E 01	.00197	.73164E-04	-.32297E-07
7500	.3061E 01	.1224E 01	.00184	.78143E-04	-.28401E-07
8000	.3244E 01	.1297E 01	.00173	.83263E-04	-.25042E-07
8500	.3439E 01	.1379E 01	.00163	.88431E-04	-.22210E-07
9000	.3648E 01	.1473E 01	.00154	.93617E-04	-.19821E-07
9500	.3873E 01	.1575E 01	.00146	.98810E-04	-.17793E-07
10000	.4115E 01	.1697E 01	.00138	.10401E-03	-.16060E-07

PRESSURE =		10 ATM.		ORIGINAL MOLE FRACTION OF COPPER = .0003		
TEMP	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS. ELECTRONS PER CC.	EQUILIBRIUM CONSTANT, ATM
2000	.3670E 18		.3669E 18		.1101E 15	.2838E 08
2500	.2936E 18		.2935E 18		.8808E 14	.2650E 10
3000	.2447E 18		.2446E 18		.7335E 14	.5476E 11
3500	.2097E 18		.2097E 18		.6244E 14	.4750E 12
4000	.1835E 18		.1834E 18		.5269E 14	.3361E 13
4500	.1631E 18		.1631E 18		.4114E 14	.2796E 13
5000	.1468E 18		.1467E 18		.2612E 14	.1792E 14
5500	.1335E 18		.1334E 18		.1155E 14	.2848E 14
6000	.1223E 18		.1223E 18		.3609E 13	.3308E 14
6500	.1129E 18		.1129E 18		.1020E 13	.3285E 14
7000	.1049E 18		.1048E 18		.3102E 12	.3114E 14
7500	.9787E 17		.9781E 17		.1057E 12	.2925E 14
8000	.9175E 17		.9170E 17		.4018E 11	.2748E 14
8500	.8635E 17		.8630E 17		.1678E 11	.2588E 14
9000	.8156E 17		.8151E 17		.7607E 10	.2445E 14
9500	.7726E 17		.7722E 17		.3700E 10	.2317E 14
10000	.7349E 17		.7336E 17		.1914E 10	.2201E 14

TEMP DEG K	SPECTRAL LINE INTENSITIES, WATTS/ICC STERI	ORIGINAL MOLE FRACTION OF COPPER = .0003
	LAMBDA 5153.5	LAMBDA 5153.2
2000	-2293E-08	.1985E-12
2500	.1397E-C6	.2084E-09
3000	.2204E-05	.2064E-07
3500	.1515E-04	.5268E-06
4000	.6032E-04	.5613E-05
4500	.1550E-03	.3102E-04
5000	.2517E-03	.9290E-04
5500	.2368E-03	.1442E-03
6000	.1372E-03	.1269E-03
6500	.6474E-04	.8522E-04
7000	.3027E-C4	.5394E-04
7500	.1485E-04	.3441E-04
8000	.7705E-05	.2246E-04
8500	.4204E-05	.1501E-04
9000	.2399E-05	.1025E-04
9500	.1424E-05	.7151E-05
10000	.8756E-C6	.5082E-05

TEMP DEG K	PRESSURE= .10 ATM.	ORIGINAL MOLE FRACTION OF COPPER=.0C10	INTERNAL PARTITION FUNCTION CU11	IONIZATION POTENTIAL LOWERING FOR CU1, EV	DEBYE PRESSURE,ATM	DEBYE
2000	.2002E 01	.1003E 01		.00000	.3C319E-01	-.12967E-15
2500	.2011E 01	.1009E 01		.00004	.35079E-02	-.10465E-12
3000	.2035E 01	.1009E 01		.00017	.84520E-03	-.89782E-11
3500	.2077E 01	.1011E 01		.00046	.30969E-03	-.21293E-09
4000	.2141E 01	.1005E 01		.00097	.14789L-03	-.22345E-08
4500	.2225E 01	.1012E 01		.00169	.85058E-C4	-.13213E-07
5000	.2329E 01	.1024E 01		.00253	.56881E-C4	-.49C91E-07
5500	.2449E 01	.1045E 01		.00328	.43918E-C4	-.11732E-06
6000	.2583E 01	.1074E 01		.00368	.39090E-04	-.18151E-06
6500	.2731E 01	.1113E 01		.00371	.38792E-04	-.20120E-06
7000	.2890E 01	.1163E 01		.00355	.40529E-C4	-.18999E-06
7500	.3061E 01	.1224E 01		.00335	.42992E-C4	-.17054E-06
8000	.3244E 01	.1297E 01		.00315	.45698E-C4	-.15148E-06
8500	.3439E 01	.1379E 01		.00297	.48489E-04	-.13472E-06
9000	.3648E 01	.1473E 01		.00281	.51313E-C4	-.12037E-06
9500	.3873E 01	.1575E 01		.00266	.54150E-04	-.10811E-06
10000	.4115E 01	.1687E 01		.00253	.56993E-04	-.97607E-07

TEMP DEG K	PRESSURE= .10 ATM.		ORIGINAL MOLE FRACTION OF COPPER= .0010		
	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS. ELECTRONS PER CC.	EQUILIBRIUM CONSTANT. ATM
2000	.3670E 18	.3666E 18	.3670E 15	.5181E 06	.1993E-17
2500	.2936E 18	.2933E 18	.2936E 15	.4838E 10	.2715E-13
3000	.2447E 18	.2444E 18	.2446E 15	.10000E 11	.1671E-10
3500	.2097E 18	.2095E 18	.2088E 15	.8690E 12	.1724E-08
4000	.1835E 18	.1833E 18	.1791E 15	.4355E 13	.5769E-07
4500	.1631E 18	.1629E 18	.1483E 15	.1481E 14	.9068E-06
5000	.1469E 18	.1466E 18	.1100E 15	.3680E 14	.8387E-05
5500	.1335E 18	.1333E 18	.6549E 14	.6790E 14	.5275E-04
6000	.1223E 18	.1221E 18	.2874E 14	.9350E 14	.2486E-03
6500	.1129E 18	.1127E 18	.9970E 13	.1029E 15	.9396E-03
7000	.1049E 18	.1047E 18	.3286E 13	.1015E 15	.2989E-02
7500	.9787E 17	.9767E 17	.1152E 13	.9662E 14	.8284E-02
8000	.9175E 17	.9157E 17	.4419E 12	.9122E 14	.2052E-01
8500	.8635E 17	.8618E 17	.1853E 12	.8608E 14	.4630E-01
9000	.8156E 17	.8139E 17	.8414E 11	.8139E 14	.9654E-01
9500	.7726E 17	.7711E 17	.4096E 11	.7715E 14	.1681E 00
10000	.7342E 17	.7325E 17	.2120E 11	.7331E 14	.3454E 00

TEMP	PRESSURE = .10 ATM.	ORIGINAL MOLE FRACTION OF COPPER = .0010
CEG K	SPECTRAL LINE INTENSITIES, WATTS/ICC STERI	LAMBDA 5105.5
2000	.6976E-08	.6615E-12
2500	.4658E-06	.6948E-09
3000	.7349E-05	.6802E-07
3500	.5066E-04	.1762E-05
4000	.2051E-03	.1908E-04
4500	.5589E-03	.1110E-03
5000	.1060E-02	.3911E-03
5500	.1343E-02	.8180E-03
6000	.1093E-02	.1011E-02
6500	.6328E-03	.2331E-03
7000	.3206E-03	.5713E-03
7500	.1618E-03	.3748E-03
8000	.8473E-04	.2470E-03
8500	.4642E-04	.1657E-03
9000	.2654E-04	.1134E-03
9500	.1577E-04	.7916E-04
10000	.9696E-05	.5628E-04

PRESSURE = .10 ATM. ORIGINAL MOLE FRACTION OF COPPER = .0030

TEMP DEG K	INTERNAL PARTITION FUNCTION CuI	IONIZATION POTENTIAL CuI, EV LOWERING FOR CuI, EV	DEBYE	RADIUS, CM	PRESSURE, ATM
2000	-2002E 01	-1660E 01	.00001	-23037E-01	-•29558E-15
2500	-2011E 01	-1050E 01	.00005	-26654E-02	-•23057E-12
3000	-2035E 01	-1000E 01	.00022	-64215E-03	-•20472E-10
3500	-2077E 01	-1001E 01	.00061	-23518E-03	-•48619E-09
4000	-2141E 01	-1005E 01	.00128	-11206E-C3	-•51359E-08
4500	-2225E 01	-1012E 01	.00225	-63962E-04	-•31073E-07
5000	-2329E 01	-1024E 01	.00344	-41908E-04	-•12275E-06
5500	-2449F 01	-1045E 01	.00465	-30985E-04	-•33408E-06
6000	-2583E 01	-1074E 01	.00559	-25779E-C4	-•63281E-06
6500	-2731E 01	-1113E 01	.00601	-23956E-04	-•85433E-06
7000	-2890E 01	-1163E 01	.00598	-24080E-04	-•90589E-06
7500	-3061E 01	-1224E 01	.00573	-25124E-04	-•85454E-06
8000	-3244E 01	-1297E 01	.00543	-26534E-04	-•77382E-06
8500	-3439E 01	-1379E 01	.00513	-28082E-04	-•59355E-06
9000	-3648E 01	-1473E 01	.00485	-29685E-04	-•62169E-06
9500	-3873E 01	-1575E 01	.00460	-31311E-04	-•55922E-06
10000	-4115E 01	-1687E 01	.00437	-32947E-04	-•50524E-06

		PRESSURE = .10 ATM. ORIGINAL MOLE FRACTION OF COPPER = .0030					
TEMP deg K	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	ATM	CONSTANT,	EQUILIBRIUM
2000	.3670E 18	.3659E 18	.1101E 16	.8973E 08	.1993E-17		
2500	.2936E 18	.2927E 18	.8808E 15	.8379E 10	.2715E-13		
3000	.2447E 18	.2439E 18	.7339E 15	.1732E 12	.1671E-10		
3500	.2097E 18	.2091E 18	.6276E 15	.1507E 13	.1725E-08		
4000	.1835E 18	.1829E 18	.5429E 15	.7584E 13	.5774E-07		
4500	.1631E 18	.1626E 18	.4631E 15	.2619E 14	.9082E-06		
5000	.1468E 18	.1463E 18	.3724E 15	.6779E 14	.8405E-05		
5500	.1333E 18	.1329E 18	.2635E 15	.1364E 15	.5290E-04		
6000	.1223E 18	.1218E 18	.1514E 15	.2150E 15	.2495E-03		
6500	.1129E 18	.1123E 18	.6827E 14	.2697E 15	.9435E-03		
7000	.1049E 18	.1043E 18	.2626E 14	.2875E 15	.3001E-02		
7500	.9787E 17	.9729E 17	.9837E 13	.2829E 15	.8314E-02		
8000	.9175E 17	.9121E 17	.3875E 13	.2706E 15	.2059E-01		
8500	.8635E 17	.8584E 17	.1642E 13	.2566E 15	.4644E-01		
9000	.8156E 17	.8107E 17	.7492E 12	.2432E 15	.9679E-01		
9500	.7726E 17	.7690E 17	.3655E 12	.2307E 15	.1885E 00		
10000	.7340F 17	.7296E 17	.1894E 12	.2194E 15	.3462E 00		

TEMP DEG K	PRESSURE = .10 ATM.	ORIGINAL MOLE FRACTION OF COPPER = .0010	SPECTRAL LINE INTENSITIES, WATTS/ICC SIE'R)
	LAMBDA 5105.5	LAMBDA 5153.2	
2000	.2073E-07	.1985E-11	
2500	.1337E-05	.2085E-08	
3000	.2205E-04	.2965E-06	
3500	.1522E-03	.5295E-C5	
4000	.6215E-03	.5783E-04	
4500	.1745E-02	.3492E-03	
5000	.3589E-02	.1325E-02	
5500	.5405E-02	.3792E-02	
6000	.5757E-02	.5323E-02	
6500	.4334E-02	.5705E-02	
7000	.2563E-02	.4567E-02	
7500	.1382E-02	.3201E-C2	
8000	.7431E-03	.2166E-02	
8500	.4113E-03	.1468E-02	
9000	.2363E-03	.1010E-02	
9500	.1407E-C3	.7064E-03	
10000	.8663E-C4	.5028E-03	

TEMP DEG K	PRESSURE= .10 ATM.	INTERNAL PARTITION FUNCTION CUI	ORIGINAL MOLE FRACTION OF COPPER=.0103 CUII	IONIZATION POTENTIAL LOWERING FOR CUI, EV	DERYE RADIUS, CM	DEBYE PRESSURE, ATM
2000	.2002E 01	.1000E 01	.00001	.1704E-01	.7292E-15	
2500	.2011E 01	.1000E 01	.00007	.1972E-02	.5885E-12	
3000	.2035E 01	.1000E 01	.00030	.4751E-03	.5051E-10	
3500	.2077E 01	.1001E 01	.00083	.1739E-03	.1201E-08	
4000	.2141E 01	.1005E 01	.00174	.8277E-04	.1274E-07	
4500	.2225E 01	.1012E 01	.C0306	.4702E-04	.7821E-07	
5000	.2329E 01	.1024E 01	.00473	.3041E-04	.3210E-06	
5500	.2449E 01	.1045E 01	.00658	.2186E-04	.9503E-06	
6000	.2583E 01	.1074E 01	.00832	.1730E-04	.2091E-05	
6500	.2731E 01	.1113E 01	.00958	.1503E-04	.3457E-05	
7000	.2890E 01	.1163E 01	.01013	.1421E-04	.4407E-05	
7500	.3061E 01	.1224E 01	.01009	.1427E-04	.4657E-05	
8000	.3244E 01	.1297E 01	.00973	.1480E-04	.4455E-05	
8500	.3439E 01	.1379E 01	.00926	.1554E-04	.4090E-05	
9000	.3648E 01	.1473E 01	.00879	.1637E-04	.3706E-05	
9500	.3873E 01	.1575E 01	.00835	.1724E-04	.3350E-05	
10000	.4115E 01	.1687E 01	.00794	.1812E-04	.3034E-05	

PRESSURE = .10 ATM.		ORIGINAL POLE FRACTION OF COPPER = .0100			
TEMP	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	EQUILIBRIUM CONSTANT, ATM
2000	.3675E 18	.3633E 18	.3670E 16	.1638E 09	.1993E-17
2500	.2936E 19	.2907E 18	.2936E 16	.1530E 11	.2715E-13
3000	.2447E 18	.2422E 18	.2446E 16	.3163E 12	.1672E-10
3500	.2097E 18	.2076E 18	.2094E 16	.2753E 13	.1726E-08
4000	.1835E 18	.1817E 18	.1821E 16	.1390E 14	.5782E-07
4500	.1631F 18	.1614E 18	.1582E 16	.4846E 14	.9101E-06
5000	.1468F 18	.1452E 18	.1338E 16	.1287E 15	.8430E-05
5500	.1333E 18	.1319E 18	.1058E 16	.2739E 15	.5312L-04
6000	.1223E 18	.1206E 18	.7415E 15	.4770E 15	.2509E-03
6500	.1129E 18	.1111E 18	.4375E 15	.6849E 15	.9495E-03
7000	.1049E 18	.1033E 18	.2150E 15	.8254E 15	.3021E-02
7500	.9787E 17	.9603E 17	.9372E 14	.8762E 15	.8371E-02
8000	.9175E 17	.8998E 17	.3974E 14	.8691E 15	.2072E-01
8500	.8636E 17	.8466E 17	.1740E 14	.8376E 15	.4670E-01
9000	.8156E 17	.7995E 17	.8057E 13	.7995E 15	.9728E-01
9500	.7727E 17	.7574E 17	.3958E 13	.7611E 15	.1894E 00
10000	.7340E 17	.7195E 17	.2058E 13	.7247E 15	.3476E 00

TEMP DEG K	SPECTRAL LINE INTENSITIES, WATTS/(CC STER)	PRESSURE = .10 ATM.	ORIGINAL MOLE FRACTION OF COPPER = .0100
	LAMBDA 5153.2		
2000	.6976E-07	.6615E-11	
2500	.4658E-05	.6948E-08	
3000	.7351E-04	.6884E-06	
3500	.5080E-03	.1767E-04	
4000	.2085E-02	.1940E-03	
4500	.5964E-02	.1193E-02	
5000	.1290E-01	.4759E-02	
5500	.2170E-01	.1321E-01	
6000	.2820E-01	.2608E-01	
6500	.2777E-01	.3656E-01	
7000	.2098E-01	.3739E-01	
7500	.1311E-01	.3050E-01	
8000	.7619E-02	.2221E-01	
8500	.4359E-02	.1556E-01	
9000	.2541E-02	.1086E-01	
9500	.1524E-02	.7650E-02	
10000	.9416E-03	.5466E-02	

TEMP DEG K	PRESSURE = .10 ATM.	ORIGINAL MOLE FRACTION OF COPPER = .0300	INTERVAL PARTITION FUNCTION CUII	IONIZATION POTENTIAL LOWERING FOR CUI, EV	DEBYE RADIUS, CM	DEBYE PRESSURE, ATM
2000	.2002E 01	-1000E 01		.00001	.1295E-01	-16622E-14
2500	.2011E 01	-1002E 01		.00010	.1498E-02	-13418E-11
3000	.2035E 01	-1000E 01		.00040	.36103E-03	-11519E-09
3500	.2077E 01	-1001E 01		.00109	.13215E-03	-27406E-08
4000	.2141E 01	-1005E 01		.00229	.62824E-04	-29149E-07
4500	.2225E 01	-1012E 01		.00405	.35594E-04	-18031E-06
5000	.2329E 01	-1024E 01		.00630	.22872E-04	-75505E-06
5500	.2449E 01	-1045E 01		.00888	.16206E-04	-23348E-05
6000	.2583E 01	-1074E 01		.01154	.12479E-04	-55787E-05
6500	.2731E 01	-1113E 01		.01387	.10381E-04	-10499E-04
7000	.2890E 01	-1163E 01		.01548	.93004E-05	-15723E-04
7500	.3061E 01	-1224E 01		.01618	.89002E-05	-19222E-04
8000	.3244E 01	-1297E 01		.01611	.89368E-05	-20253E-04
8500	.3439E 01	-1379E 01		.01562	.92209E-05	-19591E-04
9000	.3648E 01	-1473E 01		.01495	.96296E-05	-18212E-04
9500	.3873E 01	-1575E 01		.01426	.10098E-04	-16671E-04
10000	.4115E 01	-1687E 01		.01359	.10595E-04	-15193E-04

		PRESSURE = .10 ATM.		ORIGINAL MOLE FRACTION OF COPPER = .030C			
TEMP	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	CU IONS, ELECTRONS ATM	EQUILIBRIUM CONSTANT,	
2000	.3670E 18	.3560E 18	.1101E 17	.2838E 09	.1993E-17		
2500	.2936E 18	.2848E 18	.8808E 16	.2650E 11	.2716E-13		
3000	.2447E 18	.2373E 18	.7339E 16	.5480E 12	.1673E-10		
3500	.2097E 18	.2034E 18	.6287E 16	.4772E 13	.1728E-08		
4000	.1835E 18	.1760E 18	.5480E 16	.2413E 14	.5791E-07		
4500	.1631E 18	.1581E 18	.4806E 16	.8457E 14	.9124E-06		
5000	.1468E 18	.1422E 18	.4170E 16	.2276E 15	.8461E-05		
5500	.1335E 18	.1290E 18	.3490E 16	.4986E 15	.5338E-04		
6000	.1223E 18	.1178E 18	.2725E 16	.9174E 15	.2524E-03		
6500	.1129E 18	.1082E 18	.1909E 16	.1436E 16	.9568E-03		
7000	.1049E 18	.9986E 17	.1161E 16	.1927E 16	.3048E-02		
7500	.9789E 17	.9276E 17	.6145E 15	.2254E 16	.8450E-02		
8000	.9177E 17	.8670E 17	.2964E 15	.2385E 16	.2091E-01		
8500	.8637E 17	.8147E 17	.1393E 15	.2380E 16	.4711E-01		
9000	.8157E 17	.7688E 17	.6677E 14	.2311E 16	.9806E-01		
9500	.7728E 17	.7291E 17	.3339E 14	.2218E 16	.1908E 00		
10000	.7341E 17	.6915E 17	.1751E 14	.2121E 16	.3499E 00		

TEMP DEG K	PRESSURE= .10 ATM.	ORIGINAL MOLE FRACTION OF COPPER= .300	SPECTRAL LINE INTENSITIES, MATTS/(CC STFR)
	LAMBDA 5105.5	LAMBDA 5153.2	
2000	.2093E-06	.1985E-10	
2500	.1397E-04	.2085E-07	
3000	.2205E-03	.2065E-05	
3500	.1525E-02	.5304E-04	
4000	.6274E-C2	.5838E-03	
4500	.1812E-01	.3624E-02	
5000	.4019E-01	.1483E-01	
5500	.7157E-01	.4359E-01	
6000	.1036E 00	.9563E-01	
6500	.1212E 00	.1595E 00	
7000	.1133E 00	.2020E 00	
7500	.8632E-01	.2000E 00	
8000	.5684E-01	.1657E 00	
8500	.3488E-01	.1245E 00	
9000	.2106E-01	.9000E-01	
9500	.1285E-01	.6452E-01	
10000	.8010E-02	.4650E-01	

TEMP DEG K	PRESSURE= .30 ATM.	ORIGINAL MOLE FRACTION OF COPPER=.0001	INTERVAL PARTITION FUNCTION CUII	IONIZATION POTENTIAL LOWERING FOR CUI, EV	DEBYE RADIUS, CM	DEBYE PRESSURE, ATM
2000	.2002E 01	.1000E 01		.00000	.40967E-01	-.52563E-16
2500	.2011E 01	.1000E 01		.00003	.47399E-02	-.42420E-13
3000	.2035E 01	.1000E 01		.00013	.11422E-02	-.36380E-11
3500	.2077E 01	.1001E 01		.00034	.41885E-03	-.86066E-10
4000	.2141E 01	.1005E 01		.00072	.20086E-03	-.89194E-09
4500	.2225E 01	.1012E 01		.00123	.11724E-03	-.50462E-08
5000	.2329E 01	.1024E 01		.00177	.81514E-04	-.16681E-07
5500	.2449E 01	.1045E 01		.00212	.67809E-04	-.31875E-07
6000	.2583E 01	.1074E 01		.00219	.65711E-04	-.38211E-07
6500	.2731E 01	.1113E 01		.00210	.68636E-04	-.36324E-07
7000	.2890E 01	.1163E 01		.00197	.73156E-04	-.32306E-07
7500	.3061E 01	.1224E 01		.00184	.78135E-04	-.28410E-07
8000	.3244E 01	.1297E 01		.00173	.83255E-04	-.25050E-07
8500	.3439E 01	.1379E 01		.00163	.88422E-04	-.22217E-07
9000	.3648E 01	.1473E 01		.00154	.93608E-04	-.19827E-07
9500	.3873E 01	.1575E 01		.00146	.98801E-04	-.17799E-07
10000	.4115E 01	.1687E 01		.00136	.10400E-03	-.16065E-07

PRESSURE = .30 ATM.		ORIGINAL MOLE FRACTION OF COPPER = .0001			
TEMP	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	EQUILIBRIUM CONSTANT, ATM
2000	.1101E 19	.1101E 19	.1101E 15	.2636E 08	.1993E-17
2500	.8808E 18	.8807E 18	.8808E 14	.2650E 10	.2715E-13
3000	.7340E 18	.7339E 18	.7335E 14	.5476E 11	.1671E-10
3500	.6291E 18	.6291E 18	.6264E 14	.4750E 12	.1723E-08
4000	.5505E 18	.5504E 18	.5269E 14	.2361E 13	.5764E-07
4500	.4893E 18	.4893E 18	.4114E 14	.7796E 13	.9058E-06
5000	.4404E 18	.4403E 18	.2612E 14	.1792E 14	.8372E-05
5500	.4004E 18	.4003E 18	.1155E 14	.2848E 14	.5262E-04
6000	.3670E 18	.3669E 18	.3610E 13	.3309E 14	.2479E-03
6500	.3388E 18	.3387E 18	.1020E 13	.3205E 14	.9369E-03
7000	.3146E 18	.3145E 18	.3103E 12	.3114E 14	.2981E-02
7500	.2936E 18	.2935E 18	.1058E 12	.2925E 14	.8264E-02
8000	.2753E 18	.2752E 18	.4020E 11	.2748E 14	.2048E-01
8500	.2591E 18	.2590E 18	.1679E 11	.2589E 14	.4622E-01
9000	.2447E 18	.2446E 18	.7610E 10	.2446E 14	.9638E-01
9500	.2318E 18	.2317E 18	.3701E 10	.2317E 14	.1678E 00
10000	.2202E 18	.2202E 18	.1915E 10	.2202E 14	.3450E 00

TEMP DEG K	LAMBDA 5105.5	LAMBDA 5153.2
2000	.2093E-08	-1985E-12
2500	.1397E-06	-2084E-09
3000	.2204E-05	-2064E-07
3500	.1515E-04	-5268E-06
4000	.6032E-04	-5613E-05
4500	.1551E-03	-3102E-04
5000	.2518E-03	-9291E-04
5500	.2369E-03	-1443E-03
6000	.1373E-03	-1269E-03
6500	.6476E-04	-8525E-04
7000	.3028E-04	-5397E-04
7500	.1486E-04	-3443E-04
8000	.7708E-05	-2247E-04
8500	.4206E-05	-1501E-04
9000	.2400E-05	-1026E-04
9500	.1425E-05	-7154E-05
10000	.8758E-06	-5084E-05

TEMP DEG K	PRESSURE= .30 ATM.	INTERNAL PARTITION FUNCTION CUI	ORIGINAL MOLE FRACTION OF COPPER=.0003 IONIZATION POTENTIAL LOWERING FOR CUI, EV	DEBYE RADIUS, CM	DEBYE PRESSURE, ATM
2000	.2002E 01	.1000E 01	.000000	.31128E-01	-.11982E-15
2500	.2011E 01	.1000E 01	.00004	.36015E-02	-.96701E-13
3000	.2035E 01	.1000E 01	.00017	.86776E-03	-.82958E-11
3500	.2077E 01	.1001E 01	.00045	.31797E-03	-.19E71E-09
4000	.2141E C1	.1005E 01	.00095	.15189E-03	-.20626E-08
4500	.2225L 01	.1012E 01	.00165	.87442E-04	-.12162E-07
5000	.2329E 01	.1024E 01	.00246	.58625E-04	-.44839E-07
5500	.2449E 01	.1045E 01	.00316	.45503E-04	-.10548E-06
6000	.2583E 01	.1074E 01	.00353	.40796E-04	-.15967E-06
6500	.2731E 01	.1113E 01	.00354	.40711E-04	-.17407E-06
7000	.2890E 01	.1163E 01	.00338	.42642E-04	-.16312E-06
7500	.3061E 01	.1224E 01	.00316	.45276E-04	-.14602E-06
8000	.3244E 01	.1297E 01	.00299	.48141E-04	-.12956E-06
8500	.3439E 01	.1379E 01	.00282	.51089E-04	-.11518E-06
9000	.3648E 01	.1473E C1	.00266	.54067E-04	-.10290E-06
9500	.3873E 01	.1575E 01	.00252	.57057E-04	-.92413E-07
10000	.4115E 01	.1687L 01	.00240	.60054E-04	-.83430E-07

		PRESSURE = .30 ATM.				ORIGINAL MOLE FRACTION OF COPPER = .0003			
TEMP	TOTAL	INERT	CU	CU IONS,	EQUILIBRIUM				
DEG K	PARTICLES	PARTICLES	ATOMS	ELECTRONS	CONSTANT,				
	PER CC.	PER CC.	PER CC.	PER CC.	ATM				
2000	-1101E 19	-1101E 19	-3303E 15	-4915E 08	-1993E-17				
2500	-8878E 18	-8805E 18	-2642E 15	-4589E 10	-2715E-13				
3000	-7340E 18	-7338E 18	-2201E 15	-9486E 11	-1671E-10				
3500	-6291E 18	-6290E 18	-1879E 15	-8243E 12	-1724E-08				
4000	-5605E 18	-5503E 18	-1610E 15	-6128E 13	-5768E-07				
4500	-4893E 18	-4892E 18	-1328E 15	-1401E 14	-9067E-06				
5000	-4404E 18	-4402E 18	-9747E 14	-3464E 14	-8386E-05				
5500	-4004E 18	-4002E 18	-5684E 14	-6325E 14	-5274E-04				
6000	-3670E 18	-3668E 18	-2423E 14	-8584E 14	-2485E-03				
6500	-3388E 18	-3386E 18	-8221E 13	-9338E 14	-9393E-03				
7000	-3146E 18	-3144E 18	-2682E 13	-9166E 14	-2988E-02				
7500	-2936E 18	-2934E 18	-9365E 12	-8712E 14	-8282E-02				
8000	-2753E 18	-2751E 18	-3589E 12	-8219E 14	-2052E-01				
8500	-2591E 18	-2589E 18	-1504E 12	-7754E 14	-4629E-01				
9000	-2447E 18	-2445E 18	-6828E 11	-7331E 14	-9652E-01				
9500	-2310E 18	-2317E 18	-3323E 11	-6948E 14	-1860E 00				
10000	-2202E 18	-2201E 18	-1720E 11	-6602E 14	-3454E 00				

TEMP	PRESSURE= .30 ATM.	ORIGINAL MOLE FRACTION OF COPPER= .0010	INTERNAL PARTITION FUNCTION	IONIZATION POTENTIAL	DEBYE	DEBYE	PRESSURE, ATM
DEG K	CUI	CUII	LOWERING FOR CUI, EV	RADIUS, CM			
2000	.2032E 01	.1000E 01	.00001	.23037E-01			.29558E-15
2500	.2011E 01	.1000E 01	.00005	.26654E-02			.23057E-12
3000	.2035E 01	.1000E 01	.00022	.64215E-03			.20472E-10
3500	.2C77E 01	.1001E 01	.00061	.23518E-03			.48619E-09
4000	.2141E 01	.1005E 01	.00128	.11206E-03			.51360E-08
4500	.22225E 01	.1012E 01	.00225	.63960E-04			.31076E-07
5000	.2329E 01	.1024E 01	.00344	.41905E-04			.12278E-06
5500	.2449E 01	.1045E 01	.00465	.30979E-04			.33429E-06
6000	.2583E 01	.1074E 01	.00559	.25769E-04			.63359E-06
6500	.2731E 01	.1113E 01	.00601	.23946E-04			.85603E-06
7000	.2890E 01	.1163E 01	.00598	.24060E-04			.90818E-06
7500	.3061E 01	.1224E 01	.00574	.25101E-04			.85693E-06
8000	.3244E 01	.1297E 01	.00543	.26508E-04			.77607E-06
8500	.3439E 01	.1379E 01	.00513	.28054E-04			.69560E-06
9000	.3648E 01	.1473E 01	.00486	.29656E-04			.62354E-06
9500	.3873E 01	.1575E 01	.00460	.31280E-04			.56080E-06
10000	.4115E 01	.1687E 01	.00437	.32914E-04			.50675E-06

PRESSURE = .30 ATM.		ORIGINAL MOLE FRACTION OF COPPER = .0010		EQUILIBRIUM	
TEMP DEG K	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS. ELECTRONS PER CC.	CONSTANT, ATM
2000	.1101E 19	.1100E 19	.1101E 16	.8973E 06	.1993E-17
2500	.8879E 18	.8799E 18	.8808E 15	.8379E 10	.2715E-13
3000	.7346E 18	.7333E 18	.7338E 15	.7322E 12	.1671E-10
3500	.6291E 18	.6285E 18	.6276E 15	.6071E 13	.1725E-08
4000	.5505E 18	.5499E 18	.5429E 15	.7584E 13	.5774E-07
4500	.4893E 18	.4888E 18	.4631E 15	.2619E 14	.9082E-06
5000	.4404E 18	.4399E 18	.3725E 15	.6780E 14	.8405E-05
5500	.4004E 18	.3998E 18	.2638E 15	.1365E 15	.5290E-04
6000	.3670E 18	.3664E 18	.1516E 15	.2151E 15	.2495E-03
6500	.3388E 18	.3382E 18	.6845E 14	.2701E 15	.9435E-03
7000	.3146E 18	.3140E 18	.2635E 14	.2879E 15	.3001E-02
7500	.2936E 18	.2930E 18	.9874E 13	.2834E 15	.8314E-02
8000	.2753E 18	.2747E 18	.3890E 13	.2711E 15	.2059E-01
8500	.2591E 18	.2585E 18	.1649E 13	.2572E 15	.4644E-01
9000	.2447E 18	.2442E 18	.7521E 12	.2437E 15	.9679E-01
9500	.2318E 18	.2313E 18	.3670E 12	.2312E 15	.1885E 00
10000	.2202E 18	.2198E 18	.1901E 12	.2198E 15	.3462E 00

TEMP DEG K	SPECTRAL LINE INTENSITIES, WATTS/ICC STERI	ORIGINAL MOLE FRACTION OF COPPER = .0010
LAMBDA 5105.5	LAMBDA 5153.2	
2000	.2093E-07	.1985E-11
2500	.1397E-05	.2085E-08
3000	.2205E-04	.2065E-06
3500	.1522E-03	.5295E-05
4000	.6215E-03	.5783E-04
4500	.1746E-02	.3492E-03
5000	.3591E-02	.1325E-02
5500	.5409E-02	.3294E-02
6000	.5767E-02	.5332E-02
6500	.4345E-C2	.5720E-02
7000	.2571E-02	.4582E-02
7500	.1387E-02	.3213E-02
8000	.7460E-03	.2174E-02
8500	.4129E-03	.1474E-02
9000	.2372E-03	.1014E-02
9500	.1413E-03	.7092E-03
10000	.8697E-04	.5049E-03

PRESSURE = .30 ATM. ORIGINAL MOLE FRACTION OF COPPER = .0030

TEMP DEG K	INTERNAL PARTITION FUNCTION CUI	IONIZATION POTENTIAL LOWERING FOR CUI, EV	DERYE DEBYE	PRESSURE, ATM
2000	.2002E 01	.1000E 01	.00001	.17504E-01
2500	.2011E 01	.1000E 01	.00007	.20252E-02
3000	.2035E 01	.1000E 01	.00030	.48788E-03
3500	.2077E 01	.1001E 01	.00081	.17862E-03
4000	.2141E 01	.1005E 01	.00169	.84999E-04
4500	.2225E 01	.1012E 01	.00298	.48296E-04
5000	.2329E 01	.1024E 01	.00461	.31265E-04
5500	.2449E 01	.1045E 01	.00640	.22514E-04
6000	.2583E 01	.1074E 01	.00806	.17871E-04
6500	.2731E 01	.1113E 01	.00923	.15594E-04
7000	.2890E 01	.1163E 01	.00972	.14816E-04
7500	.3061E 01	.1224E 01	.00964	.14939E-04
8000	.3244E 01	.1297E 01	.00928	.15523E-04
8500	.3439E 01	.1379E 01	.00883	.16313E-04
9000	.3648E 01	.1473E 01	.00838	.17193E-04
9500	.3873E 01	.1575E 01	.00795	.18105E-04
10000	.4115E 01	.1687E 01	.00756	.19038E-04

TEMP DEG K	PRESSURE = .30 ATM.		ORIGINAL MOLE FRACTION OF COPPER = .0030		
	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	EQUILIBRIUM CONSTANT, ATM
2000	.1101E 19	-1098E 19	.3303E 16	.1554E 09	.1993E-17
2500	.8808E 18	-8782E 18	.2642E 16	.1451E 11	.2715E-13
3000	.7340E 18	-7318E 18	.2202E 16	.3001E 12	.1672E-10
3500	.6291E 18	-6273E 18	.1685E 16	.2612E 13	.1726E-08
4000	.5505E 18	-5488E 18	.1638E 16	.1318E 14	.5781E-07
4500	.4893E 18	-4878E 18	.1422E 16	.4594E 14	.9099E-06
5000	.4404E 18	-4390E 18	.1199E 16	.1218E 15	.8428E-05
5500	.4004E 18	-3989E 18	.9420E 15	.2584E 15	.5310E-04
6000	.3670E 18	-3655E 18	.6524E 15	.4473E 15	.2507E-03
6500	.3308E 18	-3371E 18	.3780E 15	.6364E 15	.9489E-03
7000	.3146E 18	-3129E 18	.1821E 15	.7593E 15	.3019E-02
7500	.2936E 18	-2919E 18	.7622E 14	.8002E 15	.8365E-02
8000	.2753E 18	-2736E 18	.3290E 14	.7905E 15	.2070E-01
8500	.2551E 18	-2575E 18	.1435E 14	.7606E 15	.4667E-01
9000	.2447E 18	-2432E 18	.6629E 13	.7252E 15	.9723E-01
9500	.2318E 18	-2304E 18	.3255E 13	.6901E 15	.1693E 00
10000	.2202E 18	-2189E 18	.1692E 13	.6569E 15	.3475E 00

TEMP CEG K	SPECTRAL LINE INTENSITIES, WATTS/(CC STER)	ORIGINAL MOLE FRACTION OF COPPER= .0C30
LAMBDA 5153.5	LAMBDA 5153.2	
2000	.6278E-07	.5954E-11
2500	.4192E-05	.6254E-08
3000	.6616E-04	.6195E-06
3500	.4572E-03	.1590E-04
4000	.1875E-02	.1745E-03
4500	.5360E-02	.1072E-02
5000	.1156E-01	.4265E-02
5500	.1932E-01	.1177E-01
6000	.2481E-01	.2294E-01
6500	.2399E-01	.3159E-01
7000	.1777E-01	.3167E-01
7500	.1099E-01	.2546E-01
8000	.6308E-02	.1639E-01
8500	.3595E-02	.1283E-01
9000	.2091E-02	.8935E-02
9500	.1253E-02	.6292E-02
10000	.7741E-03	.4493E-02

TEMP DEG K	PRESSURE = .30 ATM.	INTERNAL PARTITION FUNCTION CUII	ORIGINAL MOLE FRACTION OF COPPER = .010C	IONIZATION POTENTIAL LOWERING FOR CUI, EV	DERIVE DEBYE	PRESSURE, ATM
2000	.2602E 01	.1000E 01	.00001	.1295E-C1	-.16622E-14	
2500	.2011E 01	.1000E 01	.00010	.14988E-02	-.13418E-11	
3000	.2035E 01	.1000E 01	.00040	.36103E-03	-.111519E-09	
3500	.2077E 01	.1001E 01	.00109	.13215E-03	-.27407E-08	
4000	.2141E 01	.1005E 01	.00229	.62823E-04	-.29151E-07	
4500	.2225E 01	.1012E 01	.00405	.35591E-04	-.18035E-06	
5000	.2329E 01	.1024E 01	.00630	.22866E-04	-.75565E-06	
5500	.2449E 01	.1045E 01	.00889	.16196E-04	-.233395E-05	
6000	.2583E 01	.1074E 01	.01155	.12461E-C4	-.56026E-05	
6500	.2731E 01	.1113E 01	.01391	.10353E-04	-.10584E-04	
7000	.2890E 01	.1163E 01	.01555	.92592E-05	-.15934E-04	
7500	.3061E 01	.1224E 01	.01628	.88441E-05	-.19591E-04	
8000	.3244E 01	.1297E 01	.01624	.88673E-05	-.20733E-04	
8500	.3439E 01	.1379E 01	.01575	.91407E-05	-.20110E-04	
9000	.3648E 01	.1473E 01	.01509	.95413E-05	-.18723E-04	
9500	.3873E 01	.1575E 01	.01439	.10003E-04	-.17151E-04	
10000	.4115E 01	.1687E 01	.01372	.10494E-04	-.15637E-04	

TEMP CEG K	PRESSURE = .30 ATM.			ORIGINAL MOLE FRACTION OF COPPER = .0100		
	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	CU IONS, ELECTRONS ATM.	EQUILIBRIUM CONSTANT.
2000	.1101E 19	.1090E 19	.1101E 17	.2630E 09	.1993E-17	
2500	.8898E 18	.8720E 18	.8808E 16	.2650E 11	.2716E-13	
3000	.7340E 18	.7267E 18	.7339E 16	.5480E 12	.1673E-10	
3500	.6291E 18	.6228E 18	.6287E 16	.4772E 13	.1728E-08	
4000	.5505E 18	.5450E 18	.5481E 16	.2413E 14	.5791E-07	
4500	.4893E 18	.4844E 18	.4808E 16	.8459E 14	.9124E-06	
5000	.4404E 18	.4358E 18	.4174E 16	.2277E 15	.8461E-05	
5500	.4004E 18	.3959E 18	.3499E 16	.4993E 15	.5338E-04	
6000	.3670E 18	.3624E 18	.2741E 16	.9200E 15	.2524E-03	
6500	.3388E 18	.3340E 18	.1929E 16	.1444E 16	.9569E-03	
7000	.3146E 18	.3095E 18	.1182E 16	.1944E 16	.3049E-02	
7500	.2936E 18	.2884E 18	.6302E 15	.2283E 16	.8451E-02	
8000	.2753E 18	.2701E 18	.3058E 15	.2423E 16	.2091E-01	
8500	.2591E 18	.2541E 18	.1462E 15	.2422E 16	.4712E-01	
9000	.2447E 18	.2399E 18	.6927E 14	.2354E 16	.9808E-01	
9500	.2318E 18	.2272E 18	.3467E 14	.2261E 16	.1908E 00	
10000	.2202E 18	.2159E 18	.1620E 14	.2162E 16	.3499E 00	

TEMP DEG K	WAVELENGTH NM	PRESSURE = .30 ATM. SPECTRAL LINE INTENSITIES, WATTS/ICC STER)	ORIGINAL MOLE FRACTION OF COPPER = .010C LAMBDA 5153.2
2000		.2093E-06	.1985E-10
2500		-.1397E-04	-.2085E-07
3000		-.2205E-03	-.2065E-05
3500		-.1525E-02	-.5304E-04
4000		-.6274E-02	-.5838E-03
4500		-.1612E-01	-.3625E-02
5000		-.4023E-01	-.1485E-01
5500		-.7176E-01	-.4371E-01
6000		-.1042E 00	-.9638E-01
6500		-.1225E 00	-.1612E 00
7000		-.1154E 00	-.2056E 00
7500		-.0853E-01	-.2051E 00
8000		-.5864E-01	-.1709E 00
8500		-.3612E-01	-.1289E 00
9000		-.2185E-01	-.9337E-01
9500		-.1335E-01	-.6101E-01
10000		-.8323E-02	-.4831E-01

TEMP DEG K	PRESSURE= .30 ATM.	INTERNAL PARTITION FUNCTION CUI	IONIZATION POTENTIAL LOWERING FOR CUI, EV	DEBYE RADIUS, CM	DEBYE PRESSURE, ATM
2000	-2002E 01	.1000E 01	.00001	.98434E-02	-.37891E-14
2500	-2011E 01	.1000E 01	.00013	.11388E-02	-.30589E-11
3000	-2035E 01	.1000E 01	.00052	.27429E-03	-.26269E-09
3500	-2077E 01	.1001E 01	.00143	.10037E-03	-.62542E-08
4000	-2141E 01	.1005E 01	.00302	.47688E-04	-.66646E-07
4500	-2225E 01	.1012E 01	.00534	.26972E-04	-.41439E-06
5000	-2329E 01	.1024E 01	.00834	.17259E-04	-.17574E-05
5500	-2449E 01	.1045E 01	.01168	.12118E-04	-.55842E-05
6000	-2583E 01	.1074E 01	.01570	.91727E-05	-.14048E-04
6500	-2731E 01	.1113E 01	.01942	.74142E-05	-.28818E-04
7000	-2890E 01	.1163E 01	.02260	.63718E-05	-.48894E-04
7500	-3061E 01	.1224E 01	.02461	.58047E-05	-.69289E-04
8000	-3244E 01	.1297E 01	.02585	.55703E-05	-.83637E-04
8500	-3439E 01	.1379E 01	.02588	.55626E-05	-.89235E-04
9000	-3648E 01	.1473E 01	.02528	.56957E-05	-.88011E-04
9500	-3873E 01	.1575E 01	.02437	.59083E-05	-.83228E-04
10000	-4115E 01	.1687E 01	.02336	.61632E-05	-.77182E-04

TEMP DEG K	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	ORIGINAL MOLE FRACTION OF COPPER = .0300	CU ATOMS, PER CC.	CU IONS, ELECTRONS PER CC.	EQUILIBRIUM CONSTANT, ATM
2000	-1101E 19	-1068E 19	.3303E 17	-4915E 09	-1993E-17	
2500	.8608E 18	.6544E 18	.2642E 17	.4590E 11	.2716E-13	
3000	-7340E 18	-7120E 18	.2202E 17	.9495E 12	.1673E-10	
3500	.6291E 18	.6103E 18	.1887E 17	.8272E 13	.1730E-08	
4000	.5505E 18	.5339E 18	.1647E 17	.4188E 14	.5803E-07	
4500	-4893E 18	-4745E 18	-1453E 17	-1473E 15	-9154E-06	
5000	.4404E 18	.4268E 18	.1280E 17	.3997E 15	.8501E-05	
5500	.4C04E 18	.3875E 18	.1109E 17	.8917E 15	.5372E-04	
6000	-3670E 18	-3544E 18	.9262E 16	.1698E 16	.2545E-03	
6500	.3389E 18	.3259E 18	.7264E 16	.2816E 16	.9663E-03	
7000	-3146E 18	-3012E 18	.5210E 16	.4105E 16	.3084E-02	
7500	-2937E 18	-2797E 18	.3351E 16	.5303E 16	.8563E-02	
8000	.2753E 18	.2611E 18	.1936E 16	.6139E 16	.2121E-01	
8500	.2591E 18	.2450E 18	.1037E 16	.6541E 16	.4777E-01	
9000	-2447E 18	-2310E 18	.5383E 15	.6606E 16	.9937E-01	
9500	.2319E 18	.2186E 18	.2813E 15	.6480E 16	.1931E 00	
10000	.2203E 18	.2076E 18	.1512E 15	.6268E 16	.3539E 00	

TEMP DEG K	LAMBDA 5105.5	LAMBDA 5153.2
2000	-6278E-06	-5954E-10
2500	-4192E-04	-6254E-07
3000	.6616E-03	.6196E-05
3500	-4576E-02	-1592E-03
4000	.1886E-01	-1755E-02
4500	-5476E-01	-1095E-01
5000	.1234E 00	-4553E-01
5500	.2275E 00	-1385E 00
6000	-3522E 00	.3257E 00
6500	-4611E 00	-6070E 00
7000	.5084E 00	.9060E 00
7500	-4708E 00	-1091E 01
8000	.3713E 00	-1082E 01
8500	-2597E 00	-9269E 00
9000	.1698E 00	.7255E 00
9500	.1083E 00	-5437E 00
10000	-6918E-01	-4016E 00

TEMP DEG K	PRESSURE= 1.00 ATM.	INTERNAL PARTITION FUNCTION CUII	ORIGINAL MOLE FRACTION OF COPPER= .0001	IONIZATION POTENTIAL LOWERING FOR CUI, EV	DEBYE	DEBYE PRESSURE, ATM
2000	.2002E 01	.1000E 01	.000000	.30319E-01	-.12967E-15	
2500	.2011E 01	.1000E 01	.00004	.35079E-02	-.10465E-12	
3000	.2035E 01	.1000E 01	.00017	.84520E-03	-.89782E-11	
3500	.2077E 01	.1001E 01	.00046	.30969E-03	-.21293E-09	
4000	.2141E 01	.1005E 01	.00097	.14789E-03	-.22346E-08	
4500	.2225E 01	.1012E 01	.00169	.85056E-04	-.13214E-07	
5000	.2329E 01	.1024E 01	.00253	.56878E-04	-.49100E-07	
5500	.2449E 01	.1045E 01	.00328	.43911E-04	-.11738E-06	
6000	.2583E 01	.1074E 01	.00368	.39079E-04	-.18166E-06	
6500	.2731E 01	.1113E 01	.00371	.38777E-04	-.20143E-06	
7000	.2890E 01	.1163E 01	.00355	.40512E-04	-.19023E-06	
7500	.3061E 01	.1224E 01	.00335	.42973E-04	-.17077E-06	
8000	.3244E 01	.1297E 01	.00315	.45677E-04	-.15168E-06	
8500	.3439E 01	.1379E 01	.00297	.48467E-04	-.13490E-06	
9000	.3640E 01	.1473E 01	.00281	.51290E-04	-.12053E-06	
9500	.3873E 01	.1575E 01	.00266	.54125E-04	-.10826E-06	
10000	.4115E 01	.1687E 01	.00253	.56967E-04	-.97739E-07	

		PRESSURE= 1.00 ATM.		ORIGINAL MOLE FRACTION OF COPPER=.0001		
TEMP DEG K	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	ATM CONSTANT,	EQUILIBRIUM
2000	-3670E 19	-3670E 19	-3670E 15	-5101E 08	-1993E-17	
2500	-2936E 19	-2936E 19	-2936E 15	-4838E 10	-2715E-13	
3000	-2447E 19	-2446E 19	-2446E 15	1.000E 11	-1671E-10	
3500	-2097E 19	-2097E 19	-2088E 15	-8690E 12	-1724E-08	
4000	-1835E 19	-1835E 19	-1791E 15	-4355E 13	-5769E-07	
4500	-1631E 19	-1631E 19	-1483E 15	-1481E 14	-9068E-06	
5000	-1468E 19	-1468E 19	-1100E 15	-3680E 14	-8387E-05	
5500	-1335E 19	-1334E 19	-6553E 14	-6792E 14	-5275E-04	
6000	-1223E 19	-1223E 19	-2877E 14	-9355E 14	-2486E-03	
6500	-1129E 19	-1129E 19	-9985E 13	-1029E 15	-9396E-03	
7000	-1049E 19	-1048E 19	-3291E 13	-1016E 15	-2989E-02	
7500	-9787E 18	-9785E 18	-1154E 13	-9670E 14	-8284E-02	
8000	-9175E 18	-9173E 18	-4427E 12	-9130E 14	-2052E-01	
8500	-8635E 18	-8634E 18	-1057E 12	-8616E 14	-4630E-01	
9000	-8156E 18	-8154E 18	-6429E 11	-8146E 14	-9654E-01	
9500	-7726E 18	-7725E 18	-4103E 11	-7721E 14	-1881L 00	
10000	-7340E 18	-7339E 18	-2123E 11	-7337E 14	-3454E 00	

TEMP	DEG K	LAMBDA 5105.5	LAMBDA 5153.2
2000		.6976E-08	.6615E-12
2500		.4658E-06	.6948E-09
3000		.7349E-05	.6882E-07
3500		.5066E-04	.1762E-05
4000		.2051E-03	.1908E-04
4500		.5590E-03	.1118E-03
5000		.1060E-02	.3912E-03
5500		.1344E-02	.8185E-03
6000		.1094E-02	.1012E-02
6500		.6338E-03	.8344E-03
7000		.3212E-03	.5723E-03
7500		.1621E-03	.3754E-03
8000		.8488E-04	.2474E-03
8500		.4650E-04	.1660E-03
9000		.2658E-04	.1136E-03
9500		.1579E-04	.7930E-C4
10000		.9713E-05	.5630E-04

TEMP DEG K	PRESSURF = 1.00 ATM. INTERNAL PARTITION FUNCTION CUII	ORIGINAL MOLE FRACTION OF COPPER = .0003 IONIZATION POTENTIAL LOWERING FOR CUI.EV	DEBYE RADIUS, CM	DEBYE PRESSURE, ATM
2000	.2C02E 01	.1C00E 01	.00001	.23037E-01
2500	.2D11E 01	.1000E 01	.00005	.26654E-02
3000	.2D35E 01	.1000E 01	.00022	.64215E-03
3500	.2D77E 01	.1001E 01	.00061	.23518E-03
4000	.2D14E 01	.1005E 01	.00128	.11206E-03
4500	.2D22E 01	.1012E 01	.00225	.63960E-04
5000	.2D29E 01	.1024E 01	.00364	.41904E-04
5500	.2D44E 01	.1045E 01	.00465	.30976E-04
6000	.2D58E 01	.1074E 01	.00559	.25765E-04
6500	.2D73E 01	.1113E 01	.00602	.23934E-04
7000	.2D89E 01	.1163E 01	.00599	.24053E-04
7500	.3061E 01	.1224E 01	.00574	.25092E-04
8000	.3244E 01	.1297E 01	.00543	.26499E-04
8500	.3439E 01	.1379E 01	.00513	.28045E-04
9000	.3648E 01	.1473E 01	.00486	.29645E-04
9500	.3873E 01	.1575E 01	.00460	.31269E-04
10000	.4115E 01	.1687E 01	.00438	.32903E-04

		PRESSURE= 1.00 ATM. ORIGINAL MOLE FRACTION OF COPPER=.0003					
TEMP	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS. ELECTRONS PER CC.	ATM	EQUILIBRIUM CONSTANT.	
2000	.3670E 19	.3669E 19	.1101E 16	.8973E 08	.1993E-17		
2500	.2936E 19	.2935E 19	.8808E 15	.8379E 10	.2715E-13		
3000	.2447E 19	.2446E 19	.7338E 15	.1732E 12	.1671E-10		
3500	.2097E 19	.2097E 19	.6276E 15	.1507E 13	.1725E-08		
4000	.1635E 19	.1634E 19	.5429E 15	.7584E 13	.5774E-07		
4500	.1631E 19	.1631E 19	.4631E 15	.2619E 14	.9082E-06		
5000	.1468E 19	.1467E 19	.3726E 15	.6780E 14	.8405E-05		
5500	.1335E 19	.1334E 19	.2638E 15	.1365E 15	.5290E-04		
6000	.1223E 19	.1223E 19	.1517E 15	.2152E 15	.2495E-03		
6500	.1129E 19	.1129E 19	.6851E 14	.2702E 15	.9435E-03		
7000	.1049E 19	.1048E 19	.2638E 14	.2881E 15	.3001E-02		
7500	.9787E 18	.9781E 18	.9886E 13	.2836E 15	.8314E-02		
8000	.9175E 18	.9170E 18	.3896E 13	.2713E 15	.2059E-01		
8500	.8635E 18	.8630E 18	.1651E 13	.2573E 15	.4644E-01		
9000	.8156E 18	.8151E 18	.7532E 12	.2438E 15	.9679E-01		
9500	.7726E 18	.7722E 18	.3675E 12	.2314E 15	.1885E 00		
10000	.7340E 18	.7336E 18	.1904E 12	.2199E 15	.3462E 00		

TEMP DEG K	SPECTRAL LINE INTENSITIES, MASTERS/(CC STER)	ORIGINAL MOLE FRACTION OF COPPER= .0003
	LAMBDA 5105.5	LAMBDA 5153.2
2000	.2093E-07	.1985E-11
2500	.1397E-05	.2065E-08
3000	.2205E-04	.2065E-06
3500	.1522E-03	.5295E-05
4000	.6215E-03	.5783E-04
4500	.1746E-02	.3492E-03
5000	.3591E-02	.1325E-02
5500	.5411E-02	.3295E-02
6000	.5770E-02	.5335E-02
6500	.4349E-02	.5725E-02
7000	.2574E-02	.4587E-02
7500	.1389E-02	.3218E-02
8000	.7470E-03	.2177E-02
8500	.4135E-03	.1476E-02
9000	.2375E-03	.1015E-02
9500	.1414E-03	.7102E-03
10000	.8704E-04	.5056E-03

TEMP CEG K	PRESSURE= 1.00 ATM.	INTERNAL PARTITION FUNCTION CUII	ORIGINAL MOLE FRACTION OF COPPER= .0010 IONIZATION POTENTIAL LUMERING FOR CUI, EV	DEBYE RADIUS, CM	DEBYE PRESSURE, ATM
2000	.2902E 01	.1000E 01	.00001	.17049E-01	-.72920E-15
2500	.2011E 01	.1000E 01	.00007	.19725E-02	-.58858E-12
3000	.2035E 01	.1000E 01	.00030	.47519E-03	-.50518E-10
3500	.2077E 01	.1001E 01	.00083	.17397E-03	-.12010E-08
4000	.2141E 01	.1005E 01	.00174	.82778E-04	-.12743E-07
4500	.2225E 01	.1012E 01	.00306	.47018E-04	-.78228E-07
5000	.2329E 01	.1024E 01	.00473	.30411E-04	-.32125E-06
5500	.2449E 01	.1045E 01	.00659	.21857E-04	-.95183E-06
6000	.2583E 01	.1074E 01	.00833	.17287E-04	-.20987E-05
6500	.2731E 01	.1113E 01	.00960	.15003E-04	-.34780E-05
7000	.2890E 01	.1163E 01	.01016	.14169E-04	-.44464E-05
7500	.3061E 01	.1224E 01	.01012	.14224E-04	-.47093E-05
8000	.3244E 01	.1297E 01	.00977	.14744E-04	-.45099E-05
8500	.3439E 01	.1379E 01	.00930	.15477E-04	-.41432E-05
9000	.3648E 01	.1473E 01	.00883	.16300E-04	-.37550E-05
9500	.3873E 01	.1575E 01	.00839	.17164E-04	-.33948E-05
10000	.4115E 01	.1687E 01	.00798	.18046E-04	-.30748E-05

PRESSURE= 1.00 ATM.		ORIGINAL MOLE FRACTION OF COPPER= .0010			
TEMP DEG X	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	EQUILIBRIUM CONSTANT, ATM
2000	.3670E 19	.3666E 19	-3670E 16	.1638E 09	.1993E-17
2500	.2936E 19	.2933E 19	.2936E 16	.1530E 11	.2715E-13
3000	.2447E 19	.2444E 19	.2446E 16	.3163E 12	.1672E-10
3500	.2097E 19	.2095E 19	.2094E 16	.2753E 13	.1726E-08
4000	.1635E 19	.1633E 19	.1621E 16	.1390E 14	.5782E-07
4500	.1631E 19	.1629E 19	.1583E 16	.4847E 14	.9101E-06
5000	.1468E 19	.1466E 19	.1339E 16	.1287E 15	.8430E-05
5500	.1335E 19	.1333E 19	.1060E 16	.2741E 15	.5312E-04
6000	.1223E 19	.1222E 19	.7448E 15	.4781E 15	.2509E-03
6500	.1129E 19	.1127E 19	.4409E 15	.6876E 15	.9495E-03
7000	.1049E 19	.1047E 19	.2175E 15	.8302E 15	.3022E-02
7500	.9787E 18	.9768E 18	.9510E 14	.8627E 15	.8371E-02
8000	.9175E 18	.9157E 18	.4039E 14	.8762E 15	.2072E-01
8500	.8635E 18	.8618E 18	.1770E 14	.8450E 15	.4670E-01
9000	.8156E 18	.8139E 18	.8199E 13	.8066E 15	.9729E-01
9500	.7726E 18	.7711E 18	.4029E 13	.7678E 15	.1894E 00
10000	.7340E 18	.7325E 18	.2095E 13	.7312E 15	.3476E 00

TEMP DEG K	LAMBDA 5105.5	LAMBDA 5153.2
2000	.6976E-07	.6615E-11
2500	.4658E-05	.6948E-08
3000	.7351E-04	.6884E-06
3500	.5080E-03	.1767E-04
4000	.2085E-02	.1940E-03
4500	.5965E-02	.1193E-02
5000	.1291E-01	.4763E-02
5500	.2174E-01	.1324E-01
6000	.2832E-01	.2619E-01
6500	.2799E-01	.3685E-01
7000	.2123E-01	.3763E-01
7500	.1336E-01	.3095E-01
8000	.7745E-02	.2257E-01
8500	.4434E-02	.1583E-01
9000	.2586E-02	.1105E-01
9500	.1551E-02	.7786E-02
10000	.9584E-03	.5563E-02

TEMP DEG X	PRESSURE= 1.00 ATM.	INTERNAL PARTITION FUNCTION CUII	ORIGINAL MOLE FRACTION OF COPPER=.0030 IONIZATION POTENTIAL LOWERING FOR CUI.EV	DEBYE RADIUS.CM	DEBYE PRESSURE,ATM
2000	.2002E 01	.1000E 01	.00001	.12955E-01	-.16622E-14
2500	.2011E 01	.1000E 01	.00010	.14988E-02	-.13418E-11
3000	.2035E 01	.1000E 01	.00040	.36103E-03	-.11519E-09
3500	.2077E 01	.1001E 01	.00109	.13214E-03	-.27407E-08
4000	.2141E 01	.1005E 01	.00229	.62822E-04	-.29152E-07
4500	.2225E 01	.1012E 01	.00405	.35590E-04	-.18037E-06
5000	.2329E 01	.1024E 01	.00630	.22864E-04	-.75586E-06
5500	.2449E 01	.1045E 01	.00889	.16192E-04	-.23411E-05
6000	.2583E 01	.1074E 01	.01156	.12455E-04	-.56109E-05
6500	.2731E 01	.1113E 01	.01392	.10343E-04	-.10614E-04
7000	.2890E 01	.1163E 01	.01558	.92447E-05	-.16009E-04
7500	.3061E 01	.1224E 01	.01632	.88244E-05	-.19722E-04
8000	.3244E 01	.1297E 01	.01628	.88429E-05	-.20905E-04
8500	.3439E 01	.1379E 01	.01580	.91126E-05	-.20297E-04
9000	.3648E 01	.1473E 01	.01514	.95102E-05	-.18907E-04
9500	.3873E 01	.1575E 01	.01444	.99693E-05	-.17325E-04
10000	.4115E 01	.1687E 01	.01377	.10458E-04	-.15797E-04

		PRESSURE= 1.00 ATM.				ORIGINAL MOLE FRACTION OF COPPER= .0030		
TEMP	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	ATM	CU IONS, ELECTRONS CONSTANT,	EQUILIBRIUM	
2000	-3670E 19	-3659E 19	-1101E 17	-2838E 09	-1993E-17			
2500	.2936E 19	.2927E 19	.8808E 16	.2650E 11	.2716E-13			
3000	-2447E 19	-2439E 19	-7339E 16	-5480E 12	-1673E-10			
3500	-2097E 19	-2091E 19	-6267E 16	-4772E 13	-1728E-08			
4000	-1835E 19	-1829E 19	-5481E 16	-2413E 14	-5791E-07			
4500	-1631E 19	-1626E 19	-4809E 16	-8459E 14	-9124E-06			
5000	-1468E 19	-1463E 19	-4176E 16	-2277E 15	-8461E-05			
5500	-1335E 19	-1330E 19	-3503E 16	-4995E 15	-5338E-04			
6000	-1223E 19	-1219E 19	-2746E 16	-9209E 15	-2524E-03			
6500	-1129E 19	-1124E 19	-1937E 16	-1447E 16	-9569E-03			
7000	-1049E 19	-1044E 19	-1190E 16	-1950E 16	-3049E-02			
7500	-9787E 18	-9735E 18	-6358E 15	-2293E 16	-8452E-02			
8000	-9175E 18	-9123E 18	-3092E 15	-2436E 16	-2092E-01			
8500	.8635E 18	.8585E 18	.1460E 15	.2437E 16	.4712E-01			
9000	-8156E 18	-8108E 18	-7018E 14	-2369E 16	-9808E-01			
9500	-7726E 18	-7681E 18	-3514E 14	-2276E 16	.1908E 00			
10000	.7340E 18	.7296E 18	.1844E 14	.2177E 16	.3500E 00			

TEMP DEG K	SPECTRAL LINE INTENSITIES,WATTS/(CC STER)	ORIGINAL MOLE FRACTION OF COPPER = .0030
LAMBDA 5105.5	LAMBDA 5153.2	
2000	.2093E-06	.1965E-10
2500	.1397E-04	.2085E-07
3000	.2205E-03	.2065E-05
3500	.1525E-02	.5304E-04
4000	.6274E-02	.5838E-03
4500	.1812E-01	.3626E-02
5000	.4024E-01	.1485E-01
5500	.7183E-01	.4375E-01
6000	.1044E 00	.9657E-01
6500	.1229E 00	.1618E 00
7000	.1161E 00	.2069E 00
7500	.8932E-01	.2069E 00
8000	.5929E-01	.1728E C0
8500	.3657E-01	.1305E 00
9000	.2213E-01	.9460E-01
9500	.1353E-01	.6791E-01
10000	.8437E-02	.4898E-01

TEMP DEG K	PRESSURE= 1.00 ATM. CUI	INTERNAL PARTITION FUNCTION CUII	ORIGINAL MOLE FRACTION OF COPPER=.0100 IONIZATION POTENTIAL LUMINATING FOR CUI.EV	DEBYE RADIUS, CM	DEBYE PRESSURE, ATM
2000	.2002E 01	.1000E 01	.00002	.95875E-02	-.41007E-14
2500	.2011E 01	.1000E 01	.00013	.11092E-02	-.33105E-11
3000	.2035E 01	.1000E 01	.00054	.26715E-03	-.28430E-09
3500	.2077E 01	.1001E 01	.00147	.97759E-04	-.67693E-08
4000	.2141E 01	.1005E 01	.00310	.46443E-04	-.72149E-07
4500	.2225E 01	.1012E 01	.00548	.26264E-04	-.44883E-06
5000	.2329E 01	.1024E 01	.00857	.16799E-04	-.19058E-05
5500	.2449E 01	.1045E 01	.01222	.11785E-04	-.60715E-05
6000	.2583E 01	.1074E 01	.01617	.89057E-05	-.15350E-04
6500	.2731E 01	.1113E 01	.02006	.71791E-05	-.31756E-04
7000	.2890E 01	.1163E 01	.02344	.61426E-05	-.54574E-04
7500	.3061E 01	.1224E 01	.02586	.55645E-05	-.78655E-04
8000	.3244E 01	.1297E 01	.02713	.53073E-05	-.96694E-04
8500	.3439E 01	.1379E 01	.02731	.52723E-05	-.10480E-03
9000	.3648E 01	.1473E 01	.02677	.53709E-05	-.10450E-03
9500	.3873E 01	.1575E 01	.02586	.55674E-05	-.99474E-04
10000	.4115E 01	.1687E 01	.02482	.58003E-05	-.92595E-04

PRESSURE = 1.00 ATM.		ORIGINAL MOLE FRACTION OF COPPER = .0100			
TEMP DEG K	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	EQUILIBRIUM CONSTANT, ATM
2000	.3670E 19	.3633E 19	.3670E 17	.5101E 09	.1993E-17
2500	.2936E 19	.2907E 19	.2936E 17	.4839E 11	.2716E-13
3000	.2447E 19	.2422E 19	.2447E 17	.1001E 13	.1673E-10
3500	.2097E 19	.2076E 19	.2096E 17	.8720E 13	.1730E-08
4000	.1835E 19	.1817E 19	.1831E 17	.4416E 14	.5804E-C7
4500	.1631E 19	.1615E 19	.1615E 17	.1553E 15	.9157E-06
5000	.1468E 19	.1453E 19	.1425E 17	.4219E 15	.8506E-05
5500	.1335E 19	.1320E 19	.1239E 17	.9429E 15	.5375E-04
6000	.1223E 19	.1209E 19	.1041E 17	.1801E 16	.2547E-03
6500	.1129E 19	.1115E 19	.8259E 16	.3004E 16	.9674E-03
7000	.1049E 19	.1034E 19	.6025E 16	.4417E 16	.3089E-02
7500	.9787E 18	.9632E 18	.3962E 16	.5760E 16	.8570E-02
8000	.9176E 18	.9017E 18	.2346E 16	.6763E 16	.2125E-01
8500	.8636E 18	.8478E 18	.1282E 16	.7281E 16	.4787E-01
9000	.8156E 18	.8002E 18	.6755E 15	.7407E 16	.9957E-01
9500	.7727E 18	.7578E 18	.3562E 15	.7298E 16	.1935E 00
10000	.7341E 18	.7197E 18	.1925E 15	.7077E 16	.3545E 00

TEMP DEG K	LAMBDA 5105.5	LAMBDA 5153.2
2000	.6976E-06	-6615E-10
2500	.4658E-04	-6948E-07
3000	.7352E-03	-6884E-05
3500	.5085E-02	-1769E-03
4000	.2096E-01	.1950E-02
4500	.6089E-01	-1218E-01
5000	.1374E 00	-5070E-01
5500	.2541E 00	.1548E 00
6000	.3960E 00	.3662E 00
6500	.5242E 00	.6901E 00
7000	.5879E 00	.1048E 01
7500	.5566E 00	.1290E 01
8000	.4498E 00	.1311E 01
8500	.3212E 00	.1147E 01
9000	.2130E 00	.9105E 00
9500	.1371E 00	.6885E 00
10000	.8806E-01	.5111E 00

PRESSURE = 1.00 ATM.		ORIGINAL MOLE FRACTION OF COPPER = .0300		DEBYE	
TEMP DEG K	INTERNAL PARTITION FUNCTION CU11	IONIZATION POTENTIAL CU11	DEBYE LOWERING FOR CU11/EV	RADIUS, CM PRESSURE, ATM	
2000	.2052E 01	.1000E 01	.00002	.72849E-02	-.93478E-14
2500	.2011E 01	.1000E 01	.00017	.84274E-03	-.75474E-11
3000	.2035E 01	.1000E 01	.00071	.20296E-03	-.64839E-09
3500	.2077E 01	.1001E 01	.00194	.74249E-04	-.15451E-07
4000	.2141E 01	.1005E 01	.00408	.35255E-04	-.16494E-06
4500	.2225E 01	.1012E 01	.00723	.19914E-04	-.10296E-05
5000	.2329E 01	.1024E 01	.01133	.12706E-04	-.44048E-05
5500	.2449E 01	.1045E 01	.01624	.88685E-05	-.14248E-04
6000	.2583E 01	.1074E 01	.02169	.66389E-05	-.37052E-04
6500	.2731E 01	.1113E 01	.02734	.52665E-05	-.80406E-04
7000	.2890E 01	.1163E 01	.03274	.43974E-05	-.14874E-03
7500	.3061E 01	.1224E 01	.03739	.38510E-05	-.23729E-03
8000	.3244E 01	.1297E 01	.04081	.35282E-05	-.32914E-03
8500	.3439E 01	.1379E 01	.04275	.333680E-05	-.40203E-03
9000	.3648E 01	.1473E 01	.04329	.33259E-05	-.44205E-03
9500	.3873E 01	.1575E 01	.04279	.33650E-05	-.45050E-03
10000	.4115E 01	.1687E 01	.04166	.34561E-05	-.43771E-03

		PRESSURE= 1.00 ATM. ORIGINAL MOLE FRACTION OF COPPER= .0300			
TEMP DEG K	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	EQUILIBRIUM CONSTANT, ATM
2000	-3670E 19	-3560E 19	-1101E 18	-8974E 09	-1993E-17
2500	-2936E 19	-2848E 19	-8808E 17	-8382E 11	-2717E-13
3000	-2447E 19	-2373E 19	-7340E 17	-1734E 13	-1675E-10
3500	-2097E 19	-2034E 19	-6290E 17	-1512E 14	-1732E-08
4000	-1835E 19	-1780E 19	-5497E 17	-7663E 14	-5821E-07
4500	-1631E 19	-1582E 19	-4866E 17	-2702E 15	-9199E-06
5000	-1468E 19	-1423E 19	-4328E 17	-7375E 15	-8560E-05
5500	-1335E 19	-1293E 19	-3832E 17	-1665E 16	-5421E-04
6000	-1223E 19	-1184E 19	-3336E 17	-3241E 16	-2574E-03
6500	-1129E 19	-1090E 19	-2813E 17	-5580E 16	-9801E-03
7000	-1049E 19	-1009E 19	-2258E 17	-8619E 16	-3137E-02
7500	-9789E 18	-9379E 18	-1696E 17	-1204E 17	-8732E-02
8000	-9178E 18	-8754E 18	-1177E 17	-1530E 17	-2167E-01
8500	-8639E 18	-8207E 18	-7538E 16	-1764E 17	-4889E-01
9000	-8159E 18	-7726E 18	-4523E 16	-1937E 17	-1017E 00
9500	-7730E 18	-7304E 18	-2614E 16	-1998E 17	-1975E 00
10000	-7343E 18	-6930E 18	-1497E 16	-1993E 17	-3615E 00

PRESSURE= 1.00 ATM. ORIGINAL MOLE FRACTION OF COPPER= .0300
 TEMP SPECTRAL LINE INTENSITIES(WATTS/ICC STER)
 DEG K LAMBDA 5105.5 LAMBDA 5153.2

2000	.2093E-05	.1985E-09
2500	.1397E-03	.2085E-06
3000	.2206E-02	.2065E-04
3500	.1526E-01	.5306E-03
4000	.6293E-01	.5856E-02
4500	.1634E 00	.3669E-01
5000	.4171E 00	.1539E 00
5500	.7859E 00	.4786E 00
6000	.1269E 01	.1173E 01
6500	.1786E 01	.2351E 01
7000	.2204E 01	.3927E 01
7500	.2383E 01	.5521E 01
8000	.2257E 01	.6580E 01
8500	.1888E 01	.6740E 01
9000	.1426E 01	.6096E 01
9500	.1006E 01	.5052E 01
10000	.6848E 00	.3975E 01

TEMP DEG K	PRESSURE= 3.00 ATM.	ORIGINAL MOLE FRACTION OF COPPER= .0001	INTERVAL PARTITION FUNCTION CUII	IONIZATION POTENTIAL LOWERING FOR CUI, EV	DEBYE RADIUS, CM	DEBYE PRESSURE, ATM
2000	.2002E 01	.1000E 01	.00001	.23037E-01	-.29558E-15	
2500	.2011E 01	.1000E 01	.00005	.26654E-02	-.23857E-12	
3000	.2035E 01	.1000E 01	.00022	.64215E-03	-.20472E-10	
3500	.2077E 01	.1001E 01	.00061	.23518E-03	-.48619E-09	
4000	.2141E 01	.1005E 01	.00128	.11206E-03	-.51361E-08	
4500	.2225E 01	.1012E 01	.00225	.63960E-04	-.31077E-07	
5000	.2329E 01	.1024E 01	.00344	.41903E-04	-.12279E-06	
5500	.2449E 01	.1045E 01	.00465	.30976E-04	-.33438E-06	
6000	.2583E 01	.1074E 01	.00559	.25764E-04	-.63395E-06	
6500	.2731E 01	.1113E 01	.00602	.23933E-04	-.85679E-06	
7000	.2890E 01	.1163E 01	.00599	.24051E-04	-.90921E-06	
7500	.3061E 01	.1224E 01	.00574	.25090E-04	-.85801E-06	
8000	.3244E 01	.1297E 01	.00543	.26496E-04	-.77709E-06	
8500	.3439E 01	.1379E 01	.00513	.28042E-04	-.69652E-06	
9000	.3645E 01	.1473E 01	.00486	.29642E-04	-.62437E-06	
9500	.3873E 01	.1575E 01	.00461	.31266E-04	-.56164E-06	
10000	.4115E 01	.1687E 01	.00438	.32899E-04	-.50743E-06	

		PRESSURE= 3.00 ATM. ORIGINAL MOLE FRACTION OF COPPER= .0001					
TEMP DEG K	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	CU IONS, ELECTRONS ATM	CONSTANT.	EQUILIBRIUM
2000	-1101E 20	-1101E 20	-1101E 16	-8973E 08	-1993E-17		
2500	-6808E 19	-8007E 19	-6808E 15	-8379E 10	-2715E-13		
3000	-7340E 19	-7339E 19	-7338E 15	-1732E 12	-1671E-10		
3500	-6291E 19	-6291E 19	-6276E 15	-1507E 13	-1725E-08		
4000	-5505E 19	-5504E 19	-5429E 15	-7584E 13	-5774E-07		
4500	-4893E 19	-4893E 19	-4631E 15	-2619E 14	-9082E-06		
5000	-4404E 19	-4404E 19	-3726E 15	-6780E 14	-8405E-05		
5500	-4004E 19	-4003E 19	-2639E 15	-1365E 15	-5290E-04		
6000	-3670E 19	-3669E 19	-1517E 15	-2152E 15	-2495E-03		
6500	-3388E 19	-3387E 19	-6853E 14	-2702E 15	-9435E-03		
7000	-3146E 19	-3145E 19	-2639E 14	-2682E 15	-3001E-02		
7500	-2936E 19	-2935E 19	-9890E 13	-2837E 15	-8314E-02		
8000	-2753E 19	-2752E 19	-3897E 13	-2713E 15	-2059E-01		
8500	-2591E 19	-2590E 19	-1652E 13	-2574E 15	-4644E-01		
9000	-2447E 19	-2446E 19	-7535E 12	-2439E 15	-9679E-01		
9500	-2318E 19	-2317E 19	-3676E 12	-2314E 15	-1085E 00		
10000	-2202E 19	-2202E 19	-1905E 12	-2200E 15	-3462E 00		

TEMP DEG K	LAMBDA 5105.5	LAMBDA 5153.2
2000	.2093E-07	.1985E-11
2500	.1397E-05	.2085E-08
3000	.2205E-04	.2065E-06
3500	.1527E-03	.5295E-05
4000	.6215E-03	.5783E-04
4500	.1746E-02	.3492E-03
5000	.3591E-02	.1325E-02
5500	.5411E-02	.3296E-02
6000	.5771E-02	.5336E-02
6500	.4350E-02	.5727E-02
7000	.2575E-02	.4589E-02
7500	.1389E-02	.3219E-02
8000	.7473E-03	.2178E-02
8500	.4137E-03	.1477E-02
9000	.2376E-03	.1016E-02
9500	.1415E-03	.7105E-03
10000	.8713E-04	.5058E-03

PRESSURE= 3.00 ATM. ORIGINAL MOLE FRACTION OF COPPER= .0003

TEMP DEG K	INTERNAL PARTITION FUNCTION CUII	IONIZATION POTENTIAL LOWERING FOR CUI, EV	DEBYE RADIUS, CM	DEBYE PRESSURE, ATM
2000	-2002E 01	.1000E 01	.00001	.17504E-01
2500	-2011E 01	.-1000E 01	.-00007	.-20252E-02
3000	-2035E 01	.-1000E 01	.-00030	.-46679E-10
3500	-2077E 01	.-1001E 01	.00081	.-11097E-08
4000	-2141E 01	.-1005E 01	.-00169	.-11770E-07
4500	-2225E 01	.-1012E 01	.00298	.-72186E-07
5000	-2329E 01	.-1024E 01	.00461	.-29569E-06
5500	-2449E 01	.-1045E 01	.00640	.-87125E-06
6000	-2583E 01	.-1074E 01	.00806	.-19014E-05
6500	-2731E 01	.-1113E 01	.-00924	.-31027E-05
7000	-2890E 01	.-1163E 01	.00973	.-38999E-05
7500	-3061E 01	.-1224E 01	.-00965	.-40785E-05
8000	-3244E 01	.-1297E 01	.-00929	.-38788E-05
8500	-3439E 01	.-1379E 01	.-00884	.-35518E-05
9000	-3648E 01	.-1473E 01	.00839	.-32141E-05
9500	-3873E 01	.-1575E 01	.-00796	.-29038E-05
10000	-4115E 01	.-1687E 01	.-00757	.-26291E-05

PRESSURE = 3.00 ATM. ORIGINAL MOLE FRACTION OF COPPER = .0003

TEMP DEG K	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	CU IONS, ELECTRONS PER CC.	EQUILIBRIUM CONSTANT, ATM
2000	.1101E 20	.1101E 20	.3303E 16	.1554E 09	.1993E-17	
2500	.8808E 19	.8805E 19	.2642E 16	.1451E 11	.2715E-13	
3000	.7349E 19	.7338E 19	.2202E 16	.3001E 12	.1672E-10	
3500	.6291E 19	.6290E 19	.1885E 16	.2612E 13	.1726E-08	
4000	.5505E 19	.5503E 19	.1638E 16	.1318E 14	.5781E-07	
4500	.4893E 19	.4892E 19	.1422E 16	.4594E 14	.9099E-06	
5000	.4404E 19	.4403E 19	.1199E 16	.1216E 15	.8428E-05	
5500	.4004E 19	.4002E 19	.9426E 15	.2584E 15	.5310E-04	
6000	.3670E 19	.3668E 19	.6532E 15	.4476E 15	.2507E-03	
6500	.3388E 19	.3386E 19	.3789E 15	.6372E 15	.9489E-03	
7000	.3146E 19	.3144E 19	.1626E 15	.7607E 15	.3019E-02	
7500	.2936E 19	.2934E 19	.7857E 14	.8020E 15	.8365E-02	
8000	.2753E 19	.2751E 19	.3306E 14	.7925E 15	.2070E-01	
8500	.2591E 19	.2589E 19	.1443E 14	.7625E 15	.46667E-01	
9000	.2447E 19	.2445E 19	.6664E 13	.7271E 15	.9723E-01	
9500	.2318E 19	.2317E 19	.3273E 13	.6919E 15	.1893E 00	
10000	.2202E 19	.2201E 19	.1701E 13	.6587E 15	.3475E 00	

TEMP DEG K	PRESSURE= 3.0C ATM.	SPECTRAL LINE INTENSITIES, WATTS/(CC STER)	ORIGINAL MOLE FRACTION OF COPPER= .0003
	LAMBDA 5153.5	LAMBDA 5153.2	
2000	.6278E-07	-5954E-11	
2500	-4192E-05	-6254E-08	
3000	-6616E-04	-6195E-06	
3500	-4572E-03	-1590E-04	
4000	-1876E-02	-1745E-03	
4500	-5360E-02	-1072E-02	
5000	-1156E-01	-4266E-02	
5500	-1933E-C1	-1177E-01	
6000	-2484E-C1	-2297E-01	
6500	-2405E-01	-3166E-01	
7000	-1784E-01	-3178E-01	
7500	-1104E-01	-2557E-01	
8000	-6339E-02	-1848E-01	
8500	-3613E-02	-1290E-01	
9000	-2102E-02	-8982E-02	
9500	-1260E-02	-6325E-02	
10000	-7782E-03	-4517E-02	

PRESSURE= 3.00 ATM. ORIGINAL MOLE FRACTION OF COPPER=.0010

TEMP DEG K	INTERNAL PARTITION FUNCTION CU11	IONIZATION POTENTIAL LOMNERING FOR CU11, EV	DEBYE RADIUS, CM	DEBYE PRESSURE, ATM
2000	.2002E 01	.1000E 01	.00001	.12955E-01
2500	.2011E 01	.1000E 01	.00010	.14988E-02
3000	.2035E 01	.1000E 01	.00040	.36103E-03
3500	.2077E 01	.1001E 01	.00109	.13214E-03
4000	.2141E 01	.1005E 01	.00229	.62822E-04
4500	.2225E 01	.1012E 01	.00405	.35590E-04
5000	.2329E 01	.1024E 01	.00630	.22864E-04
5500	.2449E 01	.1045E 01	.00889	.16191E-04
6000	.2583E 01	.1074E 01	.01156	.12453E-04
6500	.2731E 01	.1113E 01	.01392	.10341E-04
7000	.2890E 01	.1163E 01	.01553	.92406E-05
7500	.3061E 01	.1224E 01	.01633	.88168E-05
8000	.3244E 01	.1297E 01	.01630	.88359E-05
8500	.3439E 01	.1379E 01	.01561	.91045E-05
9000	.3648E 01	.1473E 01	.01515	.95013E-05
9500	.3873E 01	.1575E 01	.01446	.99598E-05
10000	.4115E 01	.1687E 01	.01378	.10448E-04

PRESSURE = 3.00 ATM.		ORIGINAL MOLE FRACTION OF COPPER = .0010				
TEMP DEG K	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	EQUILIBRIUM CONSTANT, ATM	
2000	.1101E 20	-1100E 20	-1101E 17	-2838E 09	-1993E-17	
2500	.8608E 19	.8799E 19	.0608E 16	.2650E 11	.2716E-13	
3000	.7340E 19	.7333E 19	.7339E 16	.5480E 12	.1673E-10	
3500	.6291E 19	.6285E 19	.6287E 16	.4772E 13	.1728E-08	
4000	.5505E 19	.5499E 19	.5481E 16	.2413E 14	.5791E-07	
4500	.4893E 19	.4888E 19	.4809E 16	.8459E 14	.9124E-06	
5000	.4404E 19	.4399E 19	.4176E 16	.2277E 15	.8461E-05	
5500	.4004E 19	.3999E 19	.3504E 16	.4996E 15	.5338E-04	
6000	.3670E 19	.3665E 19	.2748E 16	.9212E 15	.2524E-03	
6500	.3388E 19	.3383E 19	.1939E 16	.1447E 16	.9569E-03	
7000	.3146E 19	.3141E 19	.1192E 16	.1952E 16	.3049E-02	
7500	.2936E 19	.2931E 19	.6375E 15	.2296E 16	.8452E-02	
8000	.2753E 19	.2747E 19	.3102E 15	.2440E 16	.2092E-01	
8500	.2591E 19	.2586E 19	.1465E 15	.2442E 16	.4712E-01	
9000	.2447E 19	.2442E 19	.7045E 14	.2374E 16	.9809E-01	
9500	.2318E 19	.2313E 19	.3527E 14	.2280E 16	.1908E 00	
10000	.2222E 19	.2198E 19	.1052E 14	.2191E 16	.3500E 00	

TEMP DEG K	LAMBDA 5105.5	LAMBDA 5153.2
2000	-2093E-06	-1985E-10
2500	-1397E-04	-2085E-07
3000	.2205E-03	.2065E-05
3500	.1525E-02	.5304E-04
4000	.6274E-02	.5838E-03
4500	.1812E-01	.3626E-02
5000	.4025E-01	.1485E-01
5500	.7185E-01	.4376E-01
6000	.1045E 00	.9663E-01
6500	.1231E 00	.1620E 00
7000	.1163E 00	.2072E 00
7500	.8955E-01	.2075E 00
8000	.5948E-01	.1734E 00
8500	.3670E-01	.1310E 00
9000	.2222E-01	.9495E-01
9500	.1358E-01	.6817E-01
10000	.8470E-02	.4917E-01

PRESSURE = 3.00 ATM.		ORIGINAL MOLE FRACTION OF COPPER = .0030		DEBYE	
TEMP CEG K	INTERNAL PARTITION FUNCTION CUI	IONIZATION POTENTIAL CUI	LOWERING FOR CUI.EV	RADIUS, CM DEBYE	PRESSURE, ATM
2000	.2002E 01	.1000E 01	.00001	.98434E-02	-.37891E-14
2500	.2011E 01	.1000E 01	.00013	.113688E-02	-.30589E-11
3000	.2035E 01	.1000E 01	.00052	.27429E-03	-.26269E-09
3500	.2077E 01	.1001E 01	.00143	.10037E-03	-.62543E-08
4000	.2141E 01	.1005E 01	.00302	.47687E-04	-.66650E-07
4500	.2225E 01	.1012E 01	.00534	.26970L-04	-.41447E-06
5000	.2329E 01	.1024E 01	.00834	.17255E-04	-.17585E-05
5500	.2449E 01	.1045E 01	.01189	.12112E-04	-.55928E-05
6000	.2583E 01	.1074E 01	.01571	.91624E-05	-.14095E-04
6500	.2731E 01	.1113E 01	.01946	.73982E-05	-.29005E-04
7000	.2890E 01	.1163E 01	.02268	.63480E-05	-.49446E-04
7500	.3061E 01	.1224E 01	.02495	.57710E-05	-.70511E-04
8000	.3244E 01	.1297E 01	.02606	.55255E-05	-.85688E-04
8500	.3439E 01	.1379E 01	.02614	.55073E-05	-.91950E-04
9000	.3648E 01	.1473E 01	.02557	.56317E-05	-.91048E-04
9500	.3873E 01	.1575E 01	.02467	.58372E-05	-.86306E-04
10000	.4115E 01	.1687E 01	.02366	.60863E-05	-.80146E-04

PRESSURE= 3.00 ATM.		ORIGINAL MOLE FRACTION OF COPPER= .0030				EQUILIBRIUM	
TEMP DEG K	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	CONSTANT, ATM		
2000	-1101E 20	-1098E 20	-3303E 17	-4915E 09	-1993E-17		
2500	-8608E 19	-8702E 19	-2642E 17	-4590E 11	-2716E-13		
3000	-7340E 19	-7318E 19	-2202E 17	-9495E 12	-1673E-10		
3500	-6291E 19	-6273E 19	-1887E 17	-8272E 13	-1730E-08		
4000	-5505E 19	-5488E 19	-1647E 17	-4188E 14	-5803E-07		
4500	-4893E 19	-4879E 19	-1453E 17	-1473E 15	-9154E-06		
5000	-4404E 19	-4390E 19	-1281E 17	-3999E 15	-8501E-05		
5500	-4004E 19	-3991E 19	-1112E 17	-8927E 15	-5372E-04		
6000	-3670E 19	-3657E 19	-9303E 16	-1702E 16	-2545E-03		
6500	-3388E 19	-3375E 19	-7327E 16	-2828E 16	-9664E-03		
7000	-3146E 19	-3132E 19	-5289E 16	-4136E 16	-3085E-02		
7500	-2936E 19	-2922E 19	-3430E 16	-5362E 16	-8565E-02		
8000	-2753E 19	-2738E 19	-2000E 16	-6239E 16	-2121E-01		
8500	-2591E 19	-2576E 19	-1079E 16	-6673E 16	-4779E-01		
9000	-2447E 19	-2433E 19	-5631E 15	-6757E 16	-9941E-01		
9500	-2312E 19	-2304E 19	-2953E 15	-6639E 16	-1932E 00		
10000	-2202E 19	-2189E 19	-1590E 15	-6428E 16	-3540E 00		

TEMP DEG K	PRESSURE= 3.00 ATM. SPECTRAL LINE INTENSITIES,WATTS/(CC STER)	ORIGINAL MOLE FRACTION OF COPPER= .0030 LAMBDA 5153.2
2000	.6278E-06	.5954E-10
2500	-4192E-04	-6254E-07
3000	-6616E-03	-6196E-05
3500	-4576E-02	-1592E-03
4000	-1886E-01	-1755E-02
4500	-5478E-01	-1096E-01
5000	-1235E 00	-4557E-01
5500	-2279E 00	-1388E 00
6000	-3538E 00	-3271E 00
6500	-4651E 00	-6123E 00
7000	-5161E 00	-9197E 00
7500	-4918E 00	-1116E 01
8000	-3835E 00	-1118E 01
8500	-2703E 00	-9646E 00
9000	-1776E 00	-7590E 00
9500	-1136E 00	-5706E 00
10000	-7274E-01	-4222E 00

PRESSURE= 3.00 ATM.		ORIGINAL MOLE FRACTION OF COPPER= .0100		IONIZATION POTENTIAL		DEBYE		PRESSURE, ATM	
TEMP	INTERNAL PARTITION FUNCTION	CUII	LOWERING FOR CUI, EV	RADIUS, CM	DEBYE				
DEG K									
2000	.2002E 01	.1000E 01	.00002	.72849E-02	-93478E-14				
2500	.2011E 01	.1000E 01	.00017	.84274E-03	-75474E-11				
3000	.2035E 01	.1000E 01	.00071	.20296E-03	-64839E-09				
3500	.2077E 01	.1001E 01	.00194	.74249E-04	-15451E-07				
4000	.2141E 01	.1005E 01	.00408	.35255E-04	-16494E-06				
4500	.2225E 01	.1012E 01	.00723	.19913E-04	-10297E-05				
5000	.2329E 01	.1024E 01	.01133	.12704E-04	-44059E-05				
5500	.2449E 01	.1045E 01	.01624	.88666E-05	-14257E-04				
6000	.2583E 01	.1074E 01	.02170	.66359E-05	-37102E-04				
6500	.2731E 01	.1113E 01	.02736	.52618E-05	-80619E-04				
7000	.2890E 01	.1163E 01	.03279	.43906E-05	-14944E-03				
7500	.3061E 01	.1224E 01	.03748	.38412E-05	-23911E-03				
8000	.3244E 01	.1297E 01	.04097	.35147E-05	-33294E-03				
8500	.3439E 01	.1379E 01	.04298	.33503E-05	-40841E-03				
9000	.3648E 01	.1473E 01	.04358	.33040E-05	-45088E-03				
9500	.3873E 01	.1575E 01	.04312	.33394E-05	-46097E-03				
10000	.4115E 01	.1687E 01	.04201	.34272E-05	-44889E-03				

TEMP DEG K	PRESSURE = 3.00 ATM. ORIGINAL MOLE FRACTION OF COPPER = .0100				
	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	EQUILIBRIUM CONSTANT, ATM
2000	-1101E 20	-1090E 20	-1101E 18	-8974E 09	-1993E-17
2500	-68008E 19	-6720E 19	-6808E 17	-8302E 11	-2717E-13
3000	-7340E 19	-7267E 19	-7340E 17	-1734E 13	-1675E-10
3500	-6291E 19	-6229E 19	-6290E 17	-1512E 14	-1732E-08
4000	-5505E 19	-5450E 19	-5497E 17	-7663E 14	-5821E-07
4500	-4893E 19	-4844E 19	-4866E 17	-2702E 15	-9199E-06
5000	-4404E 19	-4359E 19	-4330E 17	-7376E 15	-8560E-05
5500	-4004E 19	-3962E 19	-3835E 17	-1666E 16	-5421E-04
6000	-3670E 19	-3630E 19	-3342E 17	-3244E 16	-2574E-03
6500	-3388E 19	-3348E 19	-2823E 17	-5590E 16	-9801E-03
7000	-3146E 19	-3106E 19	-2273E 17	-8646E 16	-3137E-02
7500	-2936E 19	-2895E 19	-1714E 17	-1210E 17	-6733E-02
8000	-2753E 19	-2710E 19	-1195E 17	-1542E 17	-2168E-01
8500	-2591E 19	-2547E 19	-7698E 16	-1803E 17	-4890E-01
9000	-2447E 19	-2403E 19	-4643E 16	-1963E 17	-1017E 00
9500	-2318E 19	-2275E 19	-2695E 16	-2028E 17	-1976E 00
10000	-2202E 19	-2160E 19	-1548E 16	-2027E 17	-3616E 00

TEMP DEG K	SPECTRAL LINE INTENSITIES,WATTS/(CC STER)	PRESSURE= 3.00 ATM. ORIGINAL MOLE FRACTION OF COPPER= .0100 LAMBDA 5105.5 LAMBDA 5153.2
2000	.2093E-05	.1985E-09
2500	.1397E-03	.2085E-06
3000	.2206E-02	.2065E-04
3500	.1526E-01	.5307E-03
4000	.6293E-01	.5856E-02
4500	.1834E 00	.3669E-01
5000	.4173E 00	.1540E 00
5500	.7865E 00	.4790E 00
6000	.1271E 01	.1175E 01
6500	.1792E 01	.2359E 01
7000	.2218E 01	.3952E 01
7500	.2408E 01	.5578E 01
8000	.2292E 01	.6681E 01
8500	.1928E 01	.6882E 01
9000	.1464E 01	.6259E 01
9500	.1037E 01	.5208E 01
10000	.7082E 00	.4111E 01

TEMP CEG K	INTERNAL PARTITION FUNCTION CUI	IONIZATION POTENTIAL LOWERING FOR CUI, EV	DEBYE RADIUS, CM	DEBYE PRESSURE, ATM
2000	.2002E 01	.1000E 01	.00003	.55352E-02 -.21309E-13
2500	.2011F 01	.1000E 01	.00022	.64031E-03 -.17208E-10
3000	.2035E 01	.1000E 01	.00093	.15410E-03 -.14790E-08
3500	.2077E 01	.1001E 01	.00255	.56387E-04 -.35276E-07
4000	.2141E 01	.1005E 01	.00538	.26759E-04 -.37722E-06
4500	.2225E 01	.1012E 01	.00954	.15100E-04 -.23618E-05
5000	.2329E 01	.1024E 01	.01497	.96162E-05 -.10160E-04
5500	.2449E 01	.1045E 01	.02153	.66690E-05 -.33206E-04
6000	.2583E 01	.1074E 01	.02893	.49769E-05 -.87944E-04
6500	.2731E 01	.1113E 01	.03684	.39083E-05 -.19675E-03
7000	.2890E 01	.1163E 01	.04482	.32122E-05 -.38161E-03
7500	.3061E 01	.1224E 01	.05236	.27497E-05 -.65183E-03
8000	.3244E 01	.1297E 01	.05890	.24444E-05 -.98967E-03
8500	.3439E 01	.1379E 01	.06394	.22519E-05 -.13450E-02
9000	.3648E 01	.1473E 01	.06717	.21437E-05 -.16508E-02
9500	.3873E 01	.1575E 01	.06859	.20994E-05 -.18553E-02
10000	.4115E 01	.1687E 01	.06850	.21021E-05 -.19453E-02

PRESSURE= 3.00 ATM. ORIGINAL MOLE FRACTION OF COPPER=.0300						
TEMP	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	ATM CONSTANT.	EQUILIBRIUM
2000	-1101E 20	-1068E 20	-3303E 18	-1554E 10	-1933E-17	
2500	-6808E 19	-6544E 19	-2642E 18	-1452E 12	-2717E-13	
3000	-7340E 19	-7120E 19	-2202E 18	-3005E 13	-1676E-10	
3500	-6291E 19	-6103E 19	-1887E 18	-2621E 14	-1736E-08	
4000	-5505E 19	-5340E 19	-1650E 18	-1330E 15	-5843E-07	
4500	-4893E 19	-4746E 19	-1463E 18	-4699E 15	-9254E-06	
5000	-4404E 19	-4271E 19	-1308E 18	-1287E 16	-8633E-05	
5500	-4004E 19	-3861E 19	-1171E 18	-2927E 16	-5482E-04	
6000	-3670E 19	-3554E 19	-1042E 18	-5768E 16	-2611E-03	
6500	-3388E 19	-3276E 19	-9120E 17	-1013E 17	-9969E-03	
7000	-3146E 19	-3036E 19	-7775E 17	-1615E 17	-3200E-02	
7500	-2937E 19	-2826E 19	-6377E 17	-2362E 17	-8936E-02	
8000	-2753E 19	-2640E 19	-4977E 17	-3188E 17	-2225E-01	
8500	-2592E 19	-2475E 19	-3664E 17	-3991E 17	-5032E-01	
9000	-2448E 19	-2329E 19	-2541E 17	-4663E 17	-1049E 00	
9500	-2319E 19	-2200E 19	-1672E 17	-5132E 17	-2038E 00	
10000	-2203E 19	-2085E 19	-1060E 17	-5389E 17	-3729E 00	

TEMP DEG K	SPECTRAL LINE INTENSITIES, WATTS/(CC STER)	ORIGINAL MOLE FRACTION OF COPPER=.0300
LAMBDA 5153.2		
2000	.6278E-05	.5954E-09
2500	.4192E-03	.6254E-06
3000	.6617E-02	.6196E-04
3500	.4578E-01	.1592E-02
4000	.1689E 00	.1758E-01
4500	.5515E 00	.1103E 00
5000	.1261E 01	.4652E 00
5500	.2401E 01	.1463E 01
6000	.3961E 01	.3663E 01
6500	.5789E 01	.7621E 01
7000	.7587E 01	.1352E 02
7500	.8959E 01	.2076E 02
8000	.9543E 01	.2781E 02
8500	.9178E 01	.3276E 02
9000	.8013E 01	.3425E 02
9500	.6434E 01	.3231E 02
10000	.4849E 01	.2815E 02

TEMP DEG K	INTERNAL PARTITION FUNCTION CUI	IONIZATION POTENTIAL LOWERING FOR CUI, EV	DEBYE RADIUS, CM	DEBYE PRESSURE, ATM
2000	.2002E 01	.1000E 01	.000001	.17049E-01
2500	.2011E 01	.1000E 01	.000007	.19725E-02
3000	.2035E 01	.1000E 01	.000030	.47519E-03
3500	.2077E 01	.1001E 01	.000083	.17397E-C3
4000	.2141E 01	.1005E 01	.001174	.82777E-04
4500	.2225E 01	.1012E 01	.003006	.47018E-04
5000	.2329E 01	.1024E 01	.00473	.30410E-04
5500	.2449E 01	.1045E 01	.006559	.21855E-04
6000	.2583E 01	.1074E 01	.00833	.17285E-04
6500	.2731E 01	.1113E 01	.00960	.15000E-04
7000	.2890E 01	.1163E 01	.01016	.14165E-04
7500	.3061E 01	.1224E 01	.01013	.14219E-04
8000	.3244E 01	.1297E 01	.00977	.14738E-04
8500	.3439E 01	.1379E 01	.00931	.15470E-04
9000	.3648E 01	.1473E 01	.00884	.16293E-04
9500	.3873E 01	.1575E 01	.00839	.17156E-04
10000	.4115E 01	.1687E 01	.00798	.18036E-04

PRESSURE= 10.00 ATM.		ORIGINAL MOLE FRACTION OF COPPER= .0001			
TEMP CEG K	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS. ELECTRONS PER CC.	EQUILIBRIUM CONSTANT, ATM
2000	.3670E 20	.3670E 20	.3670E 16	.1638E 09	.1993E-17
2500	.2936E 20	.2936E 20	.2936E 16	.1530E 11	.2715E-13
3000	.2447E 20	.2446E 20	.2446E 16	.3163E 12	.1672E-10
3500	.2097E 20	.2097E 20	.2094E 16	.2753E 13	.1726E-08
4000	.1835E 20	.1835E 20	.1821E 16	.1390E 14	.5782E-07
4500	.1631E 20	.1631E 20	.1583E 16	.4847E 14	.9101E-06
5000	.1468E 20	.1468E 20	.1339E 16	.1287E 15	.8430E-05
5500	.1335E 20	.1334E 20	.1060E 16	.2742E 15	.5312E-04
6000	.1223E 20	.1223E 20	.7451E 15	.4782E 15	.2509E-03
6500	.1129E 20	.1129E 20	.4413E 15	.6879E 15	.9495E-03
7000	.1049E 20	.1048E 20	.2178E 15	.8307E 15	.3022E-02
7500	.9787E 19	.9785E 19	.9524E 14	.8633E 15	.8371E-02
8000	.9175E 19	.9173E 19	.4046E 14	.8770E 15	.2072E-01
8500	.8635E 19	.8634E 19	.1773E 14	.8457E 15	.4670E-01
9000	.8156E 19	.8154E 19	.8213E 13	.8073E 15	.9729E-01
9500	.7726E 19	.7725E 19	.4036E 13	.7685E 15	.1894E 00
10000	.7340E 19	.7339E 19	.2099E 13	.7318E 15	.3476E 00

TEMP DEG K	LAMBDA 5105.5	LAMBDA 5153.2
2000	.6976E-07	.6615E-11
2500	-.4658E-05	.6948E-08
3000	-.7351E-04	-.6884E-06
3500	-.5080E-03	-.1767E-04
4000	-.2085E-02	-.1940E-03
4500	-.5965E-02	-.1193E-02
5000	-.1291E-01	-.4764E-02
5500	-.2174E-01	-.1324E-01
6000	-.2834E-01	-.2620E-01
6500	-.2801E-01	-.3688E-01
7000	-.2125E-01	-.3787E-01
7500	-.1338E-01	-.3100E-01
8000	-.7757E-02	-.2261E-01
8500	-.4442E-02	-.1586E-01
9000	-.2590E-02	-.1107E-01
9500	-.1553E-02	-.7800E-02
10000	-.9601E-03	-.5573E-02

PRESSURE= 10.00 ATM. ORIGINAL MOLE FRACTION OF COPPER=.0003

TEMP DEG K	INTERNAL PARTITION FUNCTION CUII	IONIZATION POTENTIAL LOWERING FOR CUI, EV	DEBYE	DEBYE PRESSURE, ATM
2000	.2002E 01	.1000E 01	.00001	.12955E-01
2500	.2011E 01	.1000E 01	.00010	.14288E-02
3000	.2035E 01	.1000E 01	.00040	.36103E-03
3500	.2077E 01	.1001E 01	.00109	.13214E-03
4000	.2141E 01	.1005E 01	.00229	.62822E-04
4500	.2225E 01	.1012E 01	.00405	.35590E-04
5000	.2329E 01	.1024E 01	.00630	.22863E-04
5500	.2449E 01	.1045E 01	.00869	.16190E-04
6000	.2583E 01	.1074E 01	.01156	.12453E-04
6500	.2731E 01	.1113E 01	.01393	.10340E-04
7000	.2900E 01	.1163E 01	.01558	.92392E-05
7500	.3061E 01	.1224E 01	.01633	.80168E-05
8000	.3244E 01	.1297E 01	.01630	.88334E-05
8500	.3439E 01	.1379E 01	.01582	.91017E-05
9000	.3648E 01	.1473E 01	.01516	.94982E-05
9500	.3873E 01	.1575E 01	.01446	.99564E-05
10000	.4115E 01	.1687E 01	.01379	.10444E-04

PRESSURE= 10.00 ATM.			ORIGINAL MOLE FRACTION OF COPPER= .0003			EQUILIBRIUM		
TEMP DEG K	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	ATM	CONSTANT,		
2000	.3670E 20	.36669E 20	.1101E 17	.2636E 09	.1993E-17			
2500	.2936E 20	.2935E 20	.8808E 16	.2650E 11	.2716E-13			
3000	.2447E 20	.24466E 20	.7339E 16	.5480E 12	.1673E-10			
3500	.2097E 20	.2097E 20	.6287E 16	.4772E 13	.1728E-08			
4000	.1835E 20	.1834E 20	.5481E 16	.2413E 14	.5791E-07			
4500	.1631E 20	.1631E 20	.4809E 16	.8459E 14	.9124E-06			
5000	.1468E 20	.1468E 20	.4176E 16	.2278E 15	.8461E-05			
5500	.1335E 20	.1334E 20	.3504E 16	.4996E 15	.5338E-04			
6000	.1223E 20	.1223E 20	.2748E 16	.9213E 15	.2524E-03			
6500	.1129E 20	.1129E 20	.1940E 16	.1448E 16	.9569E-03			
7000	.1049E 20	.1048E 20	.1193E 16	.1953E 16	.3049E-02			
7500	.9787E 19	.9781E 19	.6380E 15	.2297E 16	.8452E-02			
8000	.9175E 19	.9170E 19	.3105E 15	.2441E 16	.2092E-01			
8500	.8635E 19	.8630E 19	.1467E 15	.2443E 16	.4712E-01			
9000	.8156E 19	.8151E 19	.7054E 14	.2375E 16	.9609E-01			
9500	.7726E 19	.7722E 19	.3532E 14	.2282E 16	.1908E 00			
10000	.7340E 19	.7336E 19	.1854E 14	.2183E 16	.3500E 00			

TEMP	SPECTRAL LINE INTENSITIES, WATTS/(CC STER)	ORIGINAL MOLE FRACTION OF COPPER= .0003
CEC K	LAMBDA 5105.5	LAMBDA 5153.2
2000	.2093E-06	.1985E-10
2500	*1397E-04	*2065E-07
3000	.2205E-03	.2065E-05
3500	.1525E-02	.5304E-04
4000	.6274E-02	.5838E-03
4500	.1813E-01	.3626E-02
5000	.4025E-01	.1485E-01
5500	.7185E-01	.4376E-01
6000	.1045E 00	.9665E-01
6500	.1231E 00	.1621E 00
7000	.1164E 00	.2074E 00
7500	.8963E-01	.2077E 00
8000	.5955E-01	.1736E 00
8500	.3674E-01	.1311E 00
9000	.2225E-01	.9507E-01
9500	.1360E-01	.6827E-01
10000	.8482E-02	.4923E-01

PRESSURE = 10.00 ATM. ORIGINAL MOLE FRACTION OF COPPER = .0010

TEMP DEG K	CUI	INTERNAL PARTITION FUNCTION	IONIZATION POTENTIAL LOWERING FOR CUI, EV	DEBYE	DEBYE
			RADIUS, CM	PRESSURE, ATM	PRESSURE, ATM
2000	.2002E 01	.1000E 01	.00002	.95875E-02	-.41007E-14
2500	.2011E 01	.1000E 01	.00013	.11092E-02	-.33105E-11
3000	.2035E 01	.1000E 01	.00054	.26715E-03	-.28430E-09
3500	.2077E 01	.1001E 01	.00147	.97759E-04	-.67693E-08
4000	.2141E 01	.1005E 01	.00310	.46443E-04	-.72150E-07
4500	.2225E 01	.1012E 01	.00548	.26263E-04	-.44886E-06
5000	.2329E 01	.1024E 01	.00857	.16798E-04	-.19062E-05
5500	.2449E 01	.1045E 01	.01222	.11783E-04	-.69745E-05
6000	.2583E 01	.1074E 01	.01617	.89025E-05	-.15366E-04
6500	.2731E 01	.1113E 01	.02007	.71732E-05	-.31821E-04
7000	.2890E 01	.1163E 01	.02347	.61353E-05	-.54769E-04
7500	.3061E 01	.1224E 01	.02592	.55541E-05	-.79096E-04
8000	.3244E 01	.1297E 01	.02720	.52935E-05	-.97455E-04
8500	.3439E 01	.1379E 01	.02740	.52551E-05	-.10583E-03
9000	.3648E 01	.1473E 01	.02687	.53588E-05	-.10568E-03
9500	.3873E 01	.1575E 01	.02597	.55450E-05	-.10069E-03
10000	.4115E 01	.1687E 01	.02493	.57760E-05	-.93771E-04

PRESSURE= 10.00 ATM. ORIGINAL MOLE FRACTION OF COPPER=.0010

TEMP DEG K	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	PARTICLES PER CC.	ATOMS PER CC.	CU IONS. ELECTRONS PER CC.	EQUILIBRIUM CONSTANT, ATM
2000	.3670E 20		.3666E 20		.3670E 17	.5181E 09
2500	.2936E 20		.2933E 20		.2936E 17	.4839E 11
3000	.2447E 20		.2444E 20		.2447E 17	.1001E 13
3500	.2097E 20		.2095E 20		.2096E 17	.8720E 13
4000	.1835E 20		.1833E 20		.1831E 17	.4416E 14
4500	.1631E 20		.1629E 20		.1616E 17	.1553E 15
5000	.1468E 20		.1466E 20		.1426E 17	.4219E 15
5500	.1335E 20		.1333E 20		.1240E 17	.9432E 15
6000	.1223E 20		.1222E 20		.1043E 17	.1603E 16
6500	.1129E 20		.1128E 20		.8281E 16	.3008E 16
7000	.1049E 20		.1047E 20		.6053E 16	.4428E 16
7500	.9787E 19		.9771E 19		.3992E 16	.5789E 16
8000	.9175E 19		.9159E 19		.2370E 16	.6798E 16
8500	.8635E 19		.8619E 19		.1299E 16	.7329E 16
9000	.8156E 19		.8140E 19		.6857E 15	.7462E 16
9500	.7726E 19		.7711E 19		.3620E 15	.7357E 16
10000	.7340E 19		.7326E 19		.1958E 15	.7137E 16

TEMP DEG K	SPECTRAL LINE INTENSITIES,WATTS/(CC STER)	PRESSURE= 10.00 ATM. ORIGINAL MOLE FRACTION OF COPPER=.0010
LAMBDA 5105.5	LAMBDA 5153.2	
2000	.6976E-06	.6615E-10
2500	.4658E-04	.6948E-07
3000	.7352E-03	.6884E-05
3500	.5085E-02	.1769E-03
4000	.2096E-01	.1950E-02
4500	.6089E-01	.1210E-01
5000	.1374E 00	.5071E-01
5500	.2543E 00	.1549E 00
6000	.3966E 00	.3667E 00
6500	.5257E 00	.6920E 00
7000	.5907E 00	.1053E 01
7500	.5608E 00	.1299E 01
8000	.4545E 00	.1325E 01
8500	.3254E 00	.1162E 01
9000	.2163E 00	.9242E 00
9500	.1394E 00	.6997E 00
10000	.8955E-01	.5198E 00

TEMP DEG K	PRESSURE= 10.00 ATM. INTERNAL PARTITION FUNCTION CUII	IONIZATION POTENTIAL LOWERING FOR CUI, EV	DEBYE	DEBYE PRESSURE, ATM
2000	.2002E 01	-1000E 01	.00002	-.72849E-02
2500	.2011E 01	-1000E 01	.00017	-.84274E-03
3000	.2035E 01	-1000E 01	.00071	-.64839E-09
3500	.2077E 01	-1001E 01	.00194	-.15451E-07
4000	.2141E 01	-1005E 01	.00406	-.16495E-06
4500	.2225E 01	-1012E 01	.00723	-.10297E-05
5000	.2329E 01	-1024E 01	.01133	-.12704E-04
5500	.2449E 01	-1045E 01	.01624	-.88660E-05
6000	.2583E 01	-1074E 01	.02170	-.66348E-05
6500	.2731E 01	-1113E 01	.02737	-.52602E-05
7000	.2890E 01	-1163E 01	.03281	-.43882E-05
7500	.3061E 01	-1224E 01	.03752	-.38378E-05
8000	.3244E 01	-1297E 01	.04102	-.35100E-05
8500	.3439E 01	-1379E 01	.04306	-.33442E-05
9000	.3648E 01	-1473E 01	.04368	-.32963E-05
9500	.3873E 01	-1575E 01	.04323	-.33303E-05
10000	.4115E 01	-1687E 01	.04214	-.34170E-05

PRESSURE= 10.00 ATM. ORIGINAL MOLE FRACTION OF COPPER=.0030

TEMP DEC K	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	EQUILIBRIUM CONSTANT, ATM
2000	.3670E 20	.3659E 20	.1101E 18	.8974E 09	.1993E-17
2500	.2936E 20	.2927E 20	.8608E 17	.6302E 11	.2717E-13
3000	.2447E 20	.2439E 20	.7340E 17	.1734E 13	.1675E-10
3500	.2097E 20	.2091E 20	.6290E 17	.1512E 14	.1732E-08
4000	.1835E 20	.1829E 20	.5497E 17	.7663E 14	.5821E-07
4500	.1631E 20	.1626E 20	.4866E 17	.2702E 15	.9199E-06
5000	.1468E 20	.1464E 20	.4330E 17	.7377E 15	.8560E-05
5500	.1335E 20	.1330E 20	.3837E 17	.1666E 16	.5421E-04
6000	.1223E 20	.1219E 20	.3345E 17	.3245E 16	.2574E-03
6500	.1129E 20	.1125E 20	.2827E 17	.5593E 16	.9802E-03
7000	.1049E 20	.1045E 20	.2278E 17	.8656E 16	.3137E-02
7500	.9787E 19	.9745E 19	.1720E 17	.1212E 17	.8733E-02
8000	.9175E 19	.9132E 19	.1202E 17	.1546E 17	.2168E-01
8500	.8636E 19	.8592E 19	.7755E 16	.1810E 17	.4891E-01
9000	.8156E 19	.8112E 19	.4687E 16	.1972E 17	.1018E 00
9500	.7727E 19	.7683E 19	.2724E 16	.2039E 17	.1976E 00
10000	.7340E 19	.7298E 19	.1567E 16	.2039E 17	.3617E 00

PRESSURE = 10.00 ATM. ORIGINAL MOLE FRACTION OF COPPER = .0030

TEMP DEG K	SPECTRAL LINE INTENSITIES, WATTS/(CC STER)
2000	.2093E-05 .1985E-09
2500	.1397E-03 .2085E-06
3000	.2206E-02 .2065E-04
3500	.1526E-01 .5307E-03
4000	.6293E-01 .5856E-02
4500	.1834E 00 .3669E-01
5000	.4173E 00 .1540E 00
5500	.7868E 00 .4792E 00
6000	.1272E 01 .1176E 01
6500	.1794E 01 .2362E 01
7000	.2223E 01 .3961E 01
7500	.2416E 01 .5598E 01
8000	.2304E 01 .6717E 01
8500	.1942E 01 .6933E 01
9000	.1478E 01 .6317E 01
9500	.1049E 01 .5265E 01
10000	.7166E 00 .4160E 01

PRESSURE= 10.00 ATM. ORIGINAL MOLE FRACTION OF COPPER=.0100

TEMP DEG K	INTERNAL PARTITION FUNCTION CUII	IONIZATION POTENTIAL LONERING FOR CUI, EV	DEBYE RADIUS, CM	DEBYE PRESSURE, ATM
2000	.2002E 01	.1000E 01	.00003	.53913E-02
2500	.2011E 01	.1000E 01	.00023	.62366E-03
3000	.2035E 01	.1000E 01	.00096	.15017E-03
3500	.2077E 01	.1001E 01	.00262	.54918E-04
4000	.2141E 01	.1005E 01	.00553	.26060E-04
4500	.2225E 01	.1012E 01	.00979	.14704E-04
5000	.2329E 01	.1024E 01	.01538	.93624E-05
5500	.2449E 01	.1045E 01	.02212	.65103E-05
6000	.2583E 01	.1074E 01	.02974	.48411E-05
6500	.2731E 01	.1113E 01	.03791	.37978E-05
7000	.2890E 01	.1163E 01	.04620	.31167E-05
7500	.3061E 01	.1224E 01	.05409	.26620E-05
8000	.3244E 01	.1297E 01	.06103	.23592E-05
8500	.3439E 01	.1379E 01	.06651	.21650E-05
9000	.3648E 01	.1473E 01	.07017	.20519E-05
9500	.3873E 01	.1575E 01	.07197	.20006E-05
10000	.4115E 01	.1687E 01	.07216	.19953E-05

		PRESSURE = 10.00 ATM. ORIGINAL MOLE FRACTION OF COPPER = .0100					
TEMP DEG K	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	ATM	EQUILIBRIUM CONSTANT.	
2000	.3670E 20	.3633E 20	.3670E 18	.1638E 10	.1993E-17		
2500	.2936E 20	.2907E 20	.2936E 18	.1530E 12	.2717E-13		
3000	.2447E 20	.2422E 20	.2447E 18	.3168E 13	.1676E-10		
3500	.2097E 20	.2076E 20	.2097E 18	.2763E 14	.1736E-08		
4000	.1835E 20	.1817E 20	.1834E 18	.1402E 15	.5845E-07		
4500	.1631E 20	.1615E 20	.1626E 18	.4956E 15	.9260E-06		
5000	.1468E 20	.1453E 20	.1454E 18	.1358E 16	.8641E-05		
5500	.1335E 20	.1321E 20	.1303E 18	.3090E 16	.5489E-04		
6000	.1223E 20	.1211E 20	.1162E 18	.6096E 16	.2615E-03		
6500	.1129E 20	.1117E 20	.1021E 18	.1073E 17	.9988E-03		
7000	.1049E 20	.1036E 20	.8753E 17	.1716E 17	.3208E-02		
7500	.9787E 19	.9665E 19	.7242E 17	.2520E 17	.8960E-02		
8000	.9176E 19	.9050E 19	.5719E 17	.3422E 17	.2232E-01		
8500	.8637E 19	.8508E 19	.4275E 17	.4318E 17	.5050E-01		
9000	.8157E 19	.8025E 19	.3016E 17	.5090E 17	.1053E 00		
9500	.7728E 19	.7595E 19	.2019E 17	.5652E 17	.2047E 00		
10000	.7342E 19	.7209E 19	.1301E 17	.5981E 17	.3745E 00		

TEMP DEG K	PRESSURE= 10.00 ATM. SPECTRAL LINE INTENSITIES, WATTS/1CC STERI	ORIGINAL MOLE FRACTION OF COPPER=.0100 LAMBDA 5105.5 LAMBDA 5153.2
2000	.6976E-05	.6615E-09
2500	.4658E-03	.6948E-06
3000	.7352E-02	.6884E-04
3500	.5086E-01	.1769E-02
4000	.2099E-00	.1953E-01
4500	.6129E-00	.1226E-00
5000	.1402E-01	.5173E-00
5500	.2673E-01	.1628E-01
6000	.4418E-01	.4085E-01
6500	.6480E-01	.8531E-01
7000	.8542E-01	.1522E-02
7500	.1017E-02	.2357E-02
8000	.1097E-02	.3197E-02
8500	.1071E-02	.3822E-02
9000	.9513E-01	.4065E-02
9500	.7773E-01	.3903E-02
10000	.5951E-01	.3455E-02

TEMP DEG K	INTERNAL PARTITION FUNCTION CUI	IONIZATION POTENTIAL LOWERING FOR CUI, EV	DEBYE	DEBYE	PRESSURE, ATM
2000	.2002E 01	.1000E 01	.00004	.40965E-02	-.52570E-13
2500	.2011E 01	.1000E 01	.0003C	.47384E-03	-.42462E-10
3000	.2035E 01	.1000E 01	.00126	.11407E-03	-.36521E-08
3500	.2077E 01	.1001E 01	.00345	.41699E-04	-.87223E-07
4000	.2141E 01	.1005E 01	.00726	.19775E-04	-.93469E-06
4500	.2225E 01	.1012E 01	.01292	.11147E-04	-.58712E-05
5000	.2329E 01	.1024E 01	.02032	.70867E-05	-.25385E-04
5500	.2449E 01	.1045E 01	.02929	.49162E-05	-.83640E-04
6000	.2583E 01	.1074E 01	.03954	.36419E-05	-.22445E-03
6500	.2731E 01	.1113E 01	.05070	.28402E-05	-.51266E-03
7000	.2890E 01	.1163E 01	.06223	.23099E-05	-.10262E-02
7500	.3061E 01	.1224E 01	.07394	.19474E-05	-.16349E-02
8000	.3244E 01	.1297E 01	.08492	.16955E-05	-.29660E-02
8500	.3439E 01	.1379E 01	.09469	.15206E-05	-.43686E-02
9000	.3648E 01	.1473E 01	.10266	.14023E-05	-.58980E-02
9500	.3873E 01	.1575E 01	.10849	.13271E-05	-.73439E-02
10000	.4115E 01	.1687E 01	.11199	.12857E-05	-.85025E-02

PRESSURE = 10.00 ATM.				ORIGINAL MOLE FRACTION OF COPPER = .0300		
TEMP DEG K	TOTAL PARTICLES PER CC.	INERT PARTICLES PER CC.	CU ATOMS PER CC.	CU IONS, ELECTRONS PER CC.	EQUILIBRIUM CONSTANT, ATM	
2000	.3670E 20	.3560E 20	.1101E 19	.2838E 10	.1993E-17	
2500	.2936E 20	.2848E 20	.8808E 18	.2651E 12	.2718E-13	
3000	.2447E 20	.2373E 20	.7340E 18	.5490E 13	.1678E-10	
3500	.2097E 20	.2034E 20	.6291E 18	.4793E 14	.1741E-08	
4000	.1835E 20	.1700E 20	.5503E 18	.2436E 15	.5875E-07	
4500	.1631E 20	.1582E 20	.4884E 18	.8624E 15	.9335E-06	
5000	.1468E 20	.1424E 20	.4380E 18	.2371E 16	.8741E-05	
5500	.1335E 20	.1294E 20	.3948E 18	.5419E 16	.5573E-04	
6000	.1223E 20	.1186E 20	.3559E 18	.1077E 17	.2665E-03	
6500	.1129E 20	.1094E 20	.3190E 18	.1919E 17	.1022E-02	
7000	.1049E 20	.1014E 20	.2824E 18	.3124E 17	.3295E-02	
7500	.9768E 19	.9449E 19	.2452E 18	.4709E 17	.9240E-02	
8000	.9178E 19	.8838E 19	.2071E 18	.6627E 17	.2311E-01	
8500	.8639E 19	.8295E 19	.1690E 18	.8754E 17	.5248E-01	
9000	.8160E 19	.7810E 19	.1326E 18	.1090E 18	.1098E 00	
9500	.7732E 19	.7375E 19	.9968E 17	.1284E 18	.2140E 00	
10000	.7346E 19	.6986E 19	.7202E 17	.1441E 18	.3922E 00	

TEMP DEG K	PRESSURE= 10.00 ATM. SPECTRAL LINE INTENSITIES.(WATTS/(CC STER))	ORIGINAL MOLE FRACTION OF COPPER= .0300 LAMBDA 5153.2
2000	.2093E-04	.1985E-08
2500	.1397E-02	.2085E-05
3000	.2205E-01	.2065E-03
3500	.1526E 00	.5307E-02
4000	.6299E 00	.5861E-01
4500	.1841E 01	.3683E 00
5000	.4221E 01	.1558E 01
5500	.8096E 01	.4931E 01
6000	.1354E 02	.1252E 02
6500	.2025E 02	.2666E 02
7000	.2756E 02	.4911E 02
7500	.3444E 02	.7979E 02
8000	.3971E 02	.1157E 03
8500	.4233E 02	.1511E 03
9000	.4180E 02	.1787E 03
9500	.3837E 02	.1926E 03
10000	.3294E 02	.1912E 03

SCOMPILE MAD,EXECUTE,PRINT OBJECT,DUMP,COPIES(13)

MAD (112 MAR 1964 VERSION) PROGRAM LISTING

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010839 05/09/64 11 00 29.0 AM

CALCULATION OF SELF-CONSISTANT PARTITION FUNCTION, EQUILIBRIUM CONSTANT, SPECIES CONCENTRATION AND SELECTED SPECTRAL LINE INTENSITIES FROM COPPER VAPOR IN A HOT INERT GAS.
READ DATA V-JI, EI, JII, EII, EIOM1, EIOM2, LAM(I)...LAM(2).
GAI(1)...GAI(2), ENI(1)...ENI(2)

10=0
10=C
15=C
THROUGH EPSI, FOR VALUES OF P=-0.03,-1,-3,-1,-3,-1,-10.
10=10+1
THROUGH EPSI, FOR VALUES OF M=-0.001,-0.003,-0.001,-0.003,-0.01,-0.03
1
WHENEVER IR.GE.6, IR=0
IR=IR+1
THROUGH EPSI, FOR
WHENEVER IS.GE.17, IS=0
IS=IS+1
IE=0
DEBE(10)=0.
THROUGH BETA, FOR I=1,1,I,G, 67
WHENEVER EIII.E-CORE,TRANSFER TO DELTA
AIII=(2.0*JIII(1)+1.0)*EXP(-EIII(1)*1.4388/T)
WHENEVER IE.E-0
WHENEVER AI((1).L.(1.-E-6)), TRANSFER TO DELTA
OTHERWISE
WHENEVER EIII.G-ED1
.DR.-IA((1).L.(1.E-6)), TRANSFER TO DELTA
1
END OF CONDITIONAL
2I=0.
THROUGH GAMMA, FOR IN=1,1,IN.G-1
ZI=ZI+AII(1)
THROUGH ETA, FOR II=1,0,II,G,25
WHENEVER EIII(1).E-CORE,TRANSFER TO ELTA
AIII(1)=12.0*(JIII(1)+1.0)*EXP(-EIII(1)*1.4388/T)
WHENEVER IE.E-0
WHENEVER AI((1).L.(1.E-6)), TRANSFER TO ELTA
OTHERWISE
WHENEVER EIII(1).G-ED2
.DR.-IA((1).L. 1.E-6), TRANSFER TO ELTA
1
END OF CONDITIONAL
2I=0.
THROUGH ALPH, FOR IX=1,1,IX,G,11
ZI=ZI+AII(IX)
WHENEVER IE.E-0
VION=V
OTHERWISE
DELV=(1.4987 E-7)/DEBE(IE)
VION=V -0.01V
1
END OF CONDITIONAL
LCRP=-4.342945ELOG(1.2*ZI/1.0)
1
*ELNG,(IT)-6.48286
RP=EXP(12.302585*LCRP)
C=4.*OPEN/ (KPE((IM+1). P-2))

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WHENEVER C.LE.040          e046
  XE=(12.*P/M+1.)+1.5.-125*C+ .0625*C*C
  OTHERWISE
  XE=IKP((M+1)/12.*P))+(SORT((1.*C)-1.))
END IF CONDITIONAL
M11=734002E229/P1          e047
  THROUGH MENT FOR J=1,1, 1.G.6
  M11+11=411-(INIT1+11.145E-4)*(1.XE*W11/T1).P.1.5)-W11)          e048
  /((11.7125E-4)+11.XE/T1).P.1.5)*SQR((W11)-1.)          e049
  WHENEVER .ABS. ((1.- W11))/W11<1.E-5, TRANSFER TO
  SKIP
  1   RT=W11+1)          e050
  NE=IE=M1
  IE=IE+1          e051
  DEBYE1 IF J = 4.87965*SORT(J/NE)
  ED1=EI(P1-(1.164E-3))/DEBYE1(T1)
  ED2=EI(M2-(2.322E-3))/DEBYE1(T1)
  WHENEVER .ABS.(11.-DEBYE1 IE -1)/DEBYE1 IE 1).GE. 1.E-5
  1   *AND. 1 IE *L. S1. TRANSFER TO PAR1
  PDEB=-11.80693E-24*T/DEBYE1(T1).P.3
  XNEU= XE*(M+1.)
  XNEU=XNEU
  MIN=M1*XIN          e052
  THROUGH SWB, FOR A=1,1, A.G.2
  M1(A)= (2.99793E10) / (1.00027*AM(A)+04)
  INT(A)= ((6.62517E-27)*NU(A)/12.5664)-(11.E-7)*XNEU*GA(A)*
  1   EXP(-11.3388*EN(A)/T1/Z1)          e053
  SP 11Q,IR,IS 1=P
  SM 11Q,IR,IS 1=M
  ST 11Q,IR,IS 1=T
  SNT 11Q,IR,IS 1=N1T
  SMN 11Q,IR,IS 1=M1N
  SME 11Q,IR,IS 1=ME
  SKP 11Q,IR,IS 1=KP
  SIT1 11Q,IR,IS 1=INT(1)
  SIT2 11Q,IR,IS 1=INT(2)
  WHENEVER IS.E.1, PRINT FORMAT T11, P,M
  PRINT FORMAT ANSI,T. 21 .Z1,DELV,
  1   DEBYE1,PDEB          e054
  1   THROUGH PR11 .FOR IQ =1,1, 10.G.6
  THROUGH PR11 .FOR IR =1,1, IR.G.6
  THROUGH PR11 .FOR IS =1,1, IS.G.17
  WHENEVER IS .E. 1, PRINT FORMAT T112 .
  SP11Q,IR,IS1
  1   PRINT FORMAT ANSI, S11IQ,IR,IS1,SM11IQ,IR,IS1,SMIN1IQ,IR,IS
  1   ,SME11Q,IR,IS1,SME11IQ,IR,IS1,SKP11Q,IR,IS1          e055
  THROUGH PR13 .FOR IQ =1,1, 10.G.6
  THROUGH PR13 .FOR IR =1,1, IR.G.6
  THROUGH PR13 .FOR IS =1,1, IS.G.17
  WHENEVER IS .E. 1, PRINT FORMAT T113 .
  SP11Q,IR,IS1
  1   SM11Q,IR,IS1,LAN1(2)
  PRINT FORMAT ANSI, S11IQ,IR,IS1,S11IQ,IR,IR,1          e056
  S1   VECTOR VALUES VSP = 3.1,6,19
  VECTOR VALUES VSM = 3.1,6,19
  VECTOR VALUES VST = 3.1,6,19
  VECTOR VALUES VSN1 = 3.1,6,19
  VECTOR VALUES VSNIN= 3.1,6,19          e057
  e058
  e059
  e060
  e061
  e062
  e063
  e064
  e065
  e066
  e067
  e068
  e069
  e070
  e071
  e072
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  e077
  e078
  e079
  e080
  e081
  e082
  e083
  e084
  e085
  e086
  e087
  e088
  e089
  e090
  e091
  e092
  e093
  e094
  e095
  e096

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VECTOR VALUES VSME = 3.1,6.19          *097
VECTOR VALUES VSMEU= 3.1,6.19          *098
VECTOR VALUES VSKP = 3.1,6.19          *099
VECTOR VALUES VSIT1= 3.1,6.19          *100
VECTOR VALUES VSIT2= 3.1,6.19          *101
VECTOR VALUES TIT1=S1M1,S20,9PRESSURE=F6.2,5H ATM.=SS,33HORI
1 GINAL MOLE FRACTION OF COPPER=F6.4// *102
2 S13,4HTEMP,S6,2THINTER PARTITION FUNCTION=S1M1,S6,20HIONIZATION *102
3 POTENTIAL,S7,SHDREYE,S12,SHDDEBYE// *102
4 S13,SHDGEK,S9,3MCUI,S12,4HCUI1,S10,19HDERING FOR CUT, EV *102
5 S8,9RADIUS,CH,Se,12HPRESSURE,ATM *S *102
6 VECTOR VALUES T112,S1M1,S20,9PRESSURE=F6.2,5H ATM.=SS,33HORI *103
1 GINAL MOLE FRACTION OF COPPER=F6.4// *103
2 S13,4HTEMP,S7,5HTOTAL,S10,5HIMERT,S10,2HCU,S13,8HCU IONS,S7. *103
3 LINEQUALIS:1UM// *103
4 S13,SHDGEK,S6,9HARTICLES,S6,9HATOMS,S6,9HATOMS,S10,9HEL *103
5 ELECTRONS,S6,9HCONSTANT, // *103
6 S29,7HPER CC.,S8,7HPER CC.,S8,7HPER CC.,S8,7HPER CC.,S8,3HATM *103
7 ///*9
VECTOR VALUES TIT3=S1M1,S20,9PRESSURE=F6.2,5H ATM.=SS,33HORI *104
1 GINAL MOLE FRACTION OF COPPER=F6.4// *104
2 S13,4HTEMP,S7,4HSPECTRAL LINE INTENSITIES,SWATTS/CC STER1// *104
3 S13,SHDGEK,S6,6HLAMDDA,F1,1,S6,6HLAMDDA,*7.1// *104
VECTOR VALUES ANSI= S1M0,S10,F7.0,2E15.,6,F18.5,E26.5,E20.5e6 *105
VECTOR VALUES ANS2= $1M0,S10,F7.0,5E15.4 *8 *106
VECTOR VALUES ANS3= $1M0,S10,F7.0,5E15.4 *8 *107
DIMENSION J11100),E11100), E18-4,E19-4*5 J111100),E111100),LAM(13
1 *GAI(3),EM(13),DEBEY(6),INT(13),MM(13),SP1684,VSP1,SM1 684,VSM *108
2 *ST1684,VST1,SM1(684,VSM1),SM1(684,VSM1),SM1(684,VSM1),SM1(684,VSM1) *108
3 *SM1(684,VSM1),SKP(684,VSKP1),ST11684,VST11),ST121684,VST1 *108
4 21,A(1100),A11(100),N110) *108
PRINT COMMENT S1 J AND E VALUES USED S *109
PRINT RESULTS J1111-J111671-E1111...E111251 *110
1 J1111...J111251,E1111...E111251 *110
1 PRINT COMMENT S1 DATA FOR COPPERS *111
PRINT RESULTS V,E101,E102,LAM(2),GAI(1),EN(1),LAM(2),GAI(2) *112
1 EN(2) *112
INTEGER 10,IR,IS,IE,I,IN,II,IX,A *113
END OF PROGRAM *114

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J AND E VALUES USED

DATA FOR COPPER
V = 7.725880.
GA(1) = 5.100000E 06.
ER(2) = 4.993520E 04.

E10M1 = 6.231720E 04.
EM(1) = 3.078370E 04.

E10M2 = 1.636656E 05.
LM(1) = 5105.539976
CA(2) = 4.700000E 08

REFERENCES

1. J. W. M. DuMond and E. R. Cohen, "Fundamental Constants of Atomic Physics," Part 7, Chapter 10 in Handbook of Physics, E. U. Condon and H. Odishaw, (eds.), McGraw-Hill, New York, N. Y., 1958.
2. P. J. Dickerman (ed.), Optical Spectrometric Measurements of High Temperatures, University of Chicago Press, Chicago, Ill., 1961.
3. W. Lochte-Holtgreven, "Production and Measurement of High Temperature," Reports on Progress in Physics, Vol. 21, p. 312, 1958.
4. W. Finkelnberg and H. Maecker, "Electric Arcs and Thermal Plasmas," Handbuch du Physik, Vol. XXII, p. 254, Springer, 1957.
5. S. S. Penner, Chemistry Problems in Jet Propulsion, Pergamon Press, New York, 1957.
6. M. Dole, Introduction to Statistical Thermodynamics, Prentice-Hall, Princeton, N. J., 1954.
7. C. E. Moore, Atomic Energy Levels, NBS Circular 487, National Bureau of Standards, Washington, D. C., Vol. II, 1952.
8. C. C. Corliss and W. R. Bozman, Experimental Transition Probabilities for Spectral Lines of Seventy Elements, NBS Monograph 53, National Bureau of Standards, Washington, D. C., July 1962.
9. B. M. Glennon and W. L. Wiese, Bibliography on Atomic Transition Probabilities, NBS Monograph 50, August 1962; also Addendum to NBS Monograph 50, June 1963, National Bureau of Standards, Washington, D. C.
10. L. H. Aller, Part 7, Chapter 3, in Handbook of Physics, E. U. Condon and H. Odishaw (eds.), McGraw-Hill, New York, N. Y., 1958.
11. R. B King, "Measurement of Absolute Oscillator Strengths for Lines of Neutral Atoms," 1963, J. Quant. Spectros. and Radiation Transfer, Vol. 3, p. 299.
12. E. Fermi, Thermodynamics, Dover, New York, N. Y., 1956.
13. L. H. Aller, Astrophysics, the Atmospheres of the Sun and Stars, Ronald Press, New York, N. Y., 1953.
14. L. Spitzer, Physics of Fully Ionized Gases, Interscience Press, New York, N. Y., 1962.
15. L. Pauling and S. Goudsmit, The Structure of Line Spectra, McGraw-Hill, New York, N. Y., 1930.
16. H. N. Olsen, "Measurement of Argon Transition Probabilities Using the Thermal Arc Plasma as a Radiation Source," 1963, J. Quant. Spectros. Radiation Transfer, Vol. 3, p. 59, 1963.
17. K. S. Dreilishak, Partition Functions and Thermodynamic Properties of High Temperature Gases, Gas Dynamics Laboratory, Northwestern University, Evanston, Ill., AEDC TDR-64-22, January 1964.
18. H. R. Griem, "High Density Corrections in Plasma Spectroscopy," Phys. Rev., November 1, 1962, Vol. 128, No. 3, p. 997.
19. H. N. Olsen, "The Measurement of Argon Transition Probabilities and Computation of Thermodynamic Properties of the Argon Plasma," Am. Ins. Aero. and Astro., preprint 63-369, 5th Biannual Gas Dynamics Symposium, August 1963.

20. R. L. Wilkins, Theoretical Evaluation of Chemical Propellents, Prentice Hall, Princeton, N. J., 1963.
21. K. S. Drellishak, C. F. Knopp, and A. B. Cambel, Partition Functions and Thermodynamics Properties of Argon Plasma, Gas Dynamics Laboratory, Northwestern University, Evanston, Ill., AEDC TDR 63-146, August 1963.
22. H. E. White, Introduction to Atomic Spectra, McGraw-Hill, New York, N. Y., 1963.
23. B. Arden, An Introduction to Digital Computing, Addison-Wesley, Cambridge, Mass., 1963.

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the partition function cutoff, all electron states whose classical semi-major axis exceeds one-half the Debye distance are suppressed. The results should be of use in spectroscopic plasma diagnostics and other situations where the trace amounts of copper usually present can be used for temperature measurement.

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